

**EXPLORING ANTIBIOTIC CONJUGATION TO CATIONIC
AMPHIPHILIC POLYPROLINE HELICES**

by

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Dedicated to my family.

*A secondary dedication to STEM PhD students struggling with mental illness, of whom 40% reported symptoms consistent with generalized anxiety disorder and 37% with major depressive disorder in 2020.
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LIST OF ABBREVIATIONS

AcOH	Acetic Acid
ACN	Acetonitrile
AFU	Arbitrary Fluorescence Units
AMP	Antimicrobial Peptide
Ar	Argon
Boc	t-butyl carbonate
CAPH	Cationic Amphiphilic Polyproline Helices
CBP	Chlorobiphenyl
Cbz	Carboxybenzyl
CPP	Cell-penetrating Peptide
DCM	Dichloromethane
DIEA	N, N'-diisopropylethylamine
DiSC3(5)	3,3'-Dipropylthiadicarbocyanine Iodide
DMEM	Dulbecco's Modified Eagle Medium
DMF	N,N-dimethylformamide
DMSO	Dimethylsulfoxide
DIW	Deionized Water
ESI	Electrospray Ionization
EtOAc	Ethyl Acetate
FACS	Fluorescence Activated Cell Sorting
FBS	Fetal Bovine Serum
FDA	Food and Drug Administration
Fl	Fluorescein
Fmoc	Fluorenylmethyloxycarbonyl
Fmoc-OSu	N-(fluorenylmethyloxycarbonyloxy) succinamide
HATU	Hexafluorophosphate Azabenzotriazole Tetramethyl Uronium
HFIP	Hexafluoroisopropanol
hRBC	Human Red Blood Cells

HPLC	High Performance Liquid Chromatography
IPTG	Isopropyl β -D-1-thiogalactopyranoside
KHMDS	Potassium Hexamethyldisilazide, Potassium bis(trimethylsilyl)amide
MALDI-TOF	Matrix Assisted Laser Desorption Ionization-Time of Flight
MgSO ₄	Magnesium Sulfate
MBC	Minimum Bactericidal Concentration
MeOH	Methanol
MIC	Minimum Inhibitory Concentration
Mtt	4-methyltrityl
MRSA	Methicillin-resistant Staphylococcus aureus
N ₂	Nitrogen
NaH	Sodium Hydride
NHS-FI	N-hydroxy-succinimidyl-ester-fluorescein
NPN	N-Phenylnaphthalen-1-amine
ONPG	ortho-Nitrophenyl- β -galactoside
prAMP	Proline-rich Antimicrobial Peptides
PBS	Phosphate Buffered Saline
PPII	Type II Polyproline Helix
Rh	Rhodamine
RP-HPLC	Reverse Phase High Performance Chromatography
RT	room temperature
Tat	transactivator of transcription
Tb	Tuberculosis
TB	Trypan Blue
TFA	Trifluoroacetic Acid
THF	Tetrahydrofuran
TIPS	Triisopropylsilane
VRE	Vancomycin-resistant Enterococci
WHO	World Health Organization
Z-Hyp-OH	N-carboxybenzyl-hydroxy-L-proline

ABSTRACT

Pathogenic bacteria present a critical threat to modern medicine. Therapeutic strategies to target and eliminate resilient bacteria are not advancing at the same rate as the emergence of bacterial resistance. An associated urgent concern regarding antibiotic resistance is the existence and proliferation of intracellular bacteria, which find refuge from bactericidal mechanisms by hiding within mammalian cells. Therefore, many once-successful antibiotics become ineffective through the development of resistance, or through failure to reach intracellular locations in therapeutic concentration. To overcome these challenges, the covalent combination of a conventional antibiotic with an antibiotic, cell-penetrating peptide was explored to develop dual-action antibiotic conjugates.

Herein, we utilized a strategy in conjugating the antibiotics by a cleavable linkage to cationic amphiphilic polyproline helices (CAPHs) to improve vancomycin and linezolid antibiotics. This approach enables the conjugate to penetrate cells and deliver two potent monomeric antimicrobial drugs. The vancomycin-CAPH conjugate, **VanP14S**, showed enhanced mammalian cell uptake compared to vancomycin, a poor mammalian cell-penetrating agent; and **VanP14S** was capable of cleaving and releasing two antibiotics under mimicked physiological conditions. Enhanced antibacterial activity was observed against a spectrum of Gram-positive and Gram-negative pathogens, including drug-resistant strains. Further investigation revealed that this conjugate's bactericidal activity was not entirely the result of significant membrane perturbation such as a lytic mode of action. Mammalian cell toxicity and red blood cell lysis were insignificant at relevant bactericidal concentrations below 20 μ M. The current results suggest an enhanced binding to the peptidoglycan of bacteria, the target of vancomycin, although more work is needed to justify this claim. Preliminary results on **VanP14GAPS**, a conjugate with a more rigid CAPH, convey similar activity to **VanP14S**; however, moderate increases in red blood cell lysis and cytotoxicity were observed.

Regarding the **LnzP14** conjugate, preliminary data reveal that the conjugate has Gram-negative activity against *Escherichia coli*, whereas linezolid is ineffective in killing Gram-negative bacteria. This conjugate showed significant enhancement in cellular uptake compared to the CAPH, and the release of linezolid and CAPH in physiological conditions was confirmed. Overall, arming a conventional antibiotic with an antimicrobial, cell-penetrating peptide appears to be a powerful strategy in providing novel antibiotic conjugates with the propensity to overcome the limitations in treating challenging pathogens.

CHAPTER 1. STRATEGIES IN DEVELOPING POTENT ANTIBIOTICS THROUGH PEPTIDE CONJUGATION

1.1 The Imminent Threat of Pathogens

Antibiotics have transformed medicine across the globe since the first antibiotic was discovered nearly a century ago. However, almost as quickly as each antibiotic became available for widespread usage, resistance has eclipsed the clinical usefulness of many of these antimicrobial drugs (**Table 1.1**).¹⁻² The isolation of pathogens resistant to most, if not all, available antibiotics is occurring routinely, such as drug-resistant *N. gonorrhoeae*, *A. baumannii* and *C. auris* infections, each listed as an urgent threat by the US Centers for Disease Control.³ Over time the development of new antibiotics has slowed. Pharmaceutical companies traditionally dominated antibiotic research and development, but many withdrew their efforts due to the lack of long-term success and low financial return of bringing new drugs to market.⁴ There is currently an urgent need to address resistance and only two of the eight antibiotics approved since 2017 represent a new chemical scaffold.⁵ As of December 2019, approximately a quarter of the 41 drugs in development represented a novel drug class or mechanism of action, however none have targeted activity against Gram-negative ESKAPE pathogens or critical threat pathogens identified by the World Health Organization.⁶

In addition to small molecule antibiotics, antimicrobial peptides (AMPs) have been extensively examined and are a promising class of antibiotics. Their antimicrobial activities target a range of Gram-negative and Gram-positive bacteria, fungi, parasites, and viruses through a variety of mechanisms including membrane disruption, cellular penetration, and immunomodulation.⁷⁻⁸ Currently there are ten peptide-based antibacterials, including glycopeptides, approved for use with just over 40 more in the clinical pipeline.⁹ Undesired characteristics and barriers to AMP clinical treatment have driven a push toward synthetic peptides and peptide mimics that offer greater flexibility in structure. These synthetic peptides may overcome toxicity and stability disadvantages and improve antimicrobial efficacy and selectivity.¹⁰⁻¹¹

Table 1.1. Onset of bacterial resistance after antibiotic release³

Antibiotic	Year Released	Resistance Identified	Year Identified
Penicillin	1941	Penicillin-resistant <i>Staphylococcus aureus</i>	1942
		Penicillin-resistant <i>Streptococcus pneumoniae</i>	1967
		Penicillinase-producing <i>Neisseria gonorrhoeae</i>	1976
Vancomycin	1958	Plasmid-mediated vancomycin-resistant <i>Enterococcus faecium</i>	1988
		Vancomycin-resistant <i>Staphylococcus aureus</i>	2002
Amphotericin B	1959	Amphotericin B-resistant <i>Candida auris</i>	2016
Methicillin	1960	Methicillin-resistant <i>Staphylococcus aureus</i>	1960
Cephalosporins	1980	Extended-spectrum beta-lactamase- producing <i>Escherichia coli</i>	1983
Azithromycin	1980	Azithromycin-resistant <i>Neisseria gonorrhoeae</i>	2011
Imipenem	1985	<i>Klebsiella pneumoniae</i> carbapenemase (KPC)-producing <i>Klebsiella pneumoniae</i>	1996
Ciprofloxacin	1987	Ciprofloxacin-resistant <i>Neisseria gonorrhoeae</i>	2007
Fluconazole	1990	Fluconazole-resistant <i>Candida</i>	1988
Caspofungin	2001	Caspofungin-resistant <i>Candida</i>	2004
Daptomycin	2003	Daptomycin-resistant methicillin-resistant <i>Staphylococcus aureus</i>	2004
Ceftazidime-avibactam	2015	Ceftazidime-avibactam-resistant KPC-producing <i>Klebsiella pneumoniae</i>	2015

Not only is the development of resistance an impediment in treating bacterial infections, but mammalian membrane penetration is also an obstacle for antibiotics to reach their full therapeutic potential. For instance, options in treating intracellular *Staphylococcus* infections are limited due to the poor mammalian membrane permeability of many hydrophilic antibiotics, including aminoglycosides and glycopeptides.¹²⁻¹⁶ Therefore, even if an antibiotic is effective against extracellular bacteria, it may be ineffective in clearing intracellular pathogens that may proliferate and trigger a reinfection. A route to contest both antimicrobial resistance (AMR) and therapeutically restricted infections that focuses on modifying existing antibiotics is promising.¹⁷ Peptides may provide such a tool for use in tandem with antibiotics to combat challenging bacterial infections. Whereas some peptide classes may improve the potency of antibiotics and their delivery to target intracellular locations.

1.2 Cell-Penetrating Peptides

Accessing intracellular target sites is a major challenge for many therapeutic agents. The use of cell-penetrating peptides (CPPs; protein transduction domains, PDTs) over the past few decades has proved to be a powerful tool in overcoming this obstacle. CPPs enable the delivery of

cargo such as proteins, nucleic acids, and small molecule drugs at therapeutic concentrations to cellular and intracellular target sites.¹⁸⁻²⁰ Unlike other delivery methods, CPPs provide advantages such as versatility, efficiency, and low toxicity. These short peptides are typically cationic with multiple arginine or lysine residues, or amphiphilic. Classic examples of CPPs include the Tat peptide derived from the HIV-1 Tat protein, penetratin derived from the third helix of the *Antennapedia* homeodomain protein, and oligomers of arginine or lysine (**Table 1.2**). Over a thousand CPPs have been isolated or synthetically designed, and a CPP delivery database is available that may meet a therapeutic need.²¹

Table 1.2. Examples of common cell-penetrating peptides

Peptide	Sequence
Tat	GRKKRRQRRRPQ
Penetratin	RQIKIWFQNRRMKWKK
Transportan	GWTLNSAGYLLGKINLKALAALAKKIL
Poly-arginine	(Arg) _R R=6-12

1.2.1 Antibacterial CPPs

AMPs and cell-penetrating peptides share similar physiochemical properties such as short peptide length, cationic character, and at times amphiphilicity.²² This structural redundancy leads to some peptides behaving as both AMPs and CPPs, leading to a more effective therapy. The ability of these dual-function peptides to penetrate and disrupt bacterial membranes, yet nondestructively translocate across mammalian phospholipid bilayers has been reasoned to be a result of electric potential differentiation among bacterial and mammalian membranes.²³⁻²⁴ Therefore, membrane-interacting peptides may exhibit a dual functionality as some AMPs can cross mammalian cell membranes, and a few CPPs display antimicrobial activity.²⁵

Recently, the CPP Tat has also demonstrated modest broad-spectrum antimicrobial activity. In dimerizing Tat, the authors found an increase in antibacterial activity through bacterial membrane disruption, but a decrease in mammalian cell membrane translocation.²⁶ In this instance, there seems to be a tradeoff in AMP and CPP capability for the Tat peptide. Similarly, the well-known penetratin CPP has also been examined for antimicrobial activity. Penetratin displays broad-spectrum antibacterial activity through bacterial membrane disruption. Moreover,

dimerization of penetratin increases cytotoxicity, and an amino acid shuffling modification decreases CPP mammalian cell penetration.^{25, 27}

The cationic AMP LL-37 exhibits antimicrobial and anti-biofilm activity against Gram-positive and Gram-negative pathogens through pore formation on bacterial membranes.²⁸⁻³⁰ LL-37 discriminates against bacterial membranes while leaving mammalian membranes intact in its CPP capacity. Through non-covalent linkage, LL-37, has guided the delivery of DNA into mammalian cells, thereby demonstrating CPP activity.²⁹⁻³⁰

Within the past decade, cyclic cell-penetrating peptides (cCPPs) have been comprehensively studied as an improved method of reaching intracellular targets.³¹⁻³² Oh and coworkers showed that a library of amphiphilic cCPPs, rich in tryptophan and arginine residues, displayed moderate broad-spectrum antibacterial activity. These peptides also showed synergistic effects with tetracycline in time-kill experiments against multidrug-resistant pathogens.³³

Bacterial membrane permeation and disruption with AMPs is a conventional mode of action in bactericidal activity;³⁴ however, in some cases CPPs display antimicrobial activity without bacterial membrane disruption. In the case of synthetic cationic amphiphilic polyproline helices (CAPHs), cell penetration has been observed with subcellular localization, and broad-spectrum antibacterial activity was seen against bacteria in liquid culture and against intracellular infections with minimal cytotoxicity or membrane lysis.³⁵⁻⁴⁵ Modifications to these CAPHs has modulated both CPP and AMP activities. Increasing cationic character with extended scaffolds increased mammalian cellular accumulation and antibacterial activity without increased bacterial membrane disruption.^{41, 43, 45} These examples demonstrate that merging antibacterial components and cell-penetrating components can expand therapies in treating difficult intracellular infections.

1.2.2 Coadministration of antibiotics and CPPs

The effectiveness of antibiotics and ancillary drug combinations of has been verified at the clinical level and is gaining support in designing new strategies to overcome microbial resistance. For instance, it is common to see the use of β -lactam antibiotics with β -lactamase inhibitors to suppress the emergence of resistance;⁴⁶ as well as the well-known combination therapy of isoniazid (INH), rifampin (RIF), ethambutol (EMB), and/or pyrazinamide (PZA) in treating tuberculosis.⁴⁷⁻⁴⁸ There is evidence beginning to surface that antibiotic efficacy may be improved through co-administration with cell-penetrating peptides (CPPs).

Randhawa and coworkers have shown that two arginine-rich cationic CPPs, labeled P3 and P8, with little antibiotic activity alone, synergistically improved the efficacy and delivery of conventional antibiotics against MRSA.⁴⁹ Oxacillin, norfloxacin, and vancomycin MICs with sensitive and resistant *S. aureus* were significantly lowered by the combination with P3 and P8 CPPs, thus decreasing the antibiotic cytotoxicity observed with higher dosing. Recently, a non-conjugated curcumin and octa-arginine R8 combination showed enhanced antibacterial activity against Gram-negative and positive bacterial isolates and faster killing kinetics. This complex showed enhanced mammalian cellular accumulation compared to curcumin alone; and the complex displayed bacterial membrane damage to *E. coli* and *S. aureus* as a proposed mode of bactericidal action (**Figure 1.1**).⁵⁰

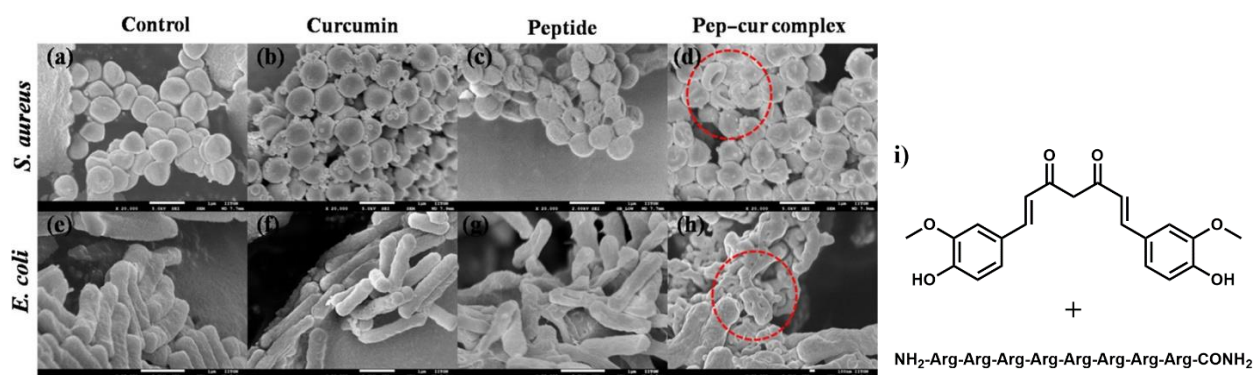


Figure 1.1. SEM micrographs of bacteria, the top panel shows images of *S. aureus* while the bottom one shows images of *E. coli*. (a,e) Control and (b,f) treated with 17 nmol raw curcumin; (c,g) treated with 7.9 nmol peptide; (d,h) treated with MIC90 concentration of the peptide–curcumin complex (1:5). (i) Structure of curcumin and R8. Reprinted (adapted).⁵⁰ Copyright (2020), American Chemical Society.

There is great interest in using antibiotic-CPP conjugates. Enhanced activity against resistant bacterial strains and reduced side-effects or toxicity is the objective in the administration of antibiotics with CPPs.⁵¹ This review will highlight and further investigate the recent progress on antibiotic-CPP covalent conjugates to effectively target and treat drug-resistant bacterial infections, while minimizing mammalian cell toxicity, and efforts to treat intracellular pathogenic bacteria.

1.3 Cell Penetrating Peptide Conjugates with Antibiotics

1.3.1 Glycopeptide-CPP conjugates

Vancomycin is a glycopeptide antibacterial used to treat several Gram-positive infections, typically those unresponsive to other antibiotics, thus deeming it the “antibiotic of last resort”. Many strategies to overcome vancomycin-resistance and improve mammalian cellular penetration have been presented. Recent examples include combining positively charged groups and lipophilic components. The Haldar group has explored the conjugation of vancomycin to cationic lipids⁵²⁻⁵⁶ and bacterial cell wall pyrophosphate binding moieties.⁵⁷⁻⁵⁸ The incorporation of sulfonium-based cationic lipophilic components to vancomycin led to enhanced interaction with the negatively charged bacterial cell membrane and increased bactericidal activity through membrane disruption.⁵⁹ Boger and coworkers have built upon the chlorobiphenyl (CBP) modification used in ortavancin to incorporate a quaternary ammonium ion for membrane activity resulting in better bactericidal activity.⁶⁰⁻⁶¹ Other lipophilic and cell surface-interacting vancomycin derivatives have shown increased activity in treating vancomycin resistance.⁶²⁻⁶⁷

The simple addition of an arginine to the C-terminus of vancomycin by amide linkage showed that the narrow spectrum Gram-positive antibiotic was capable of inhibiting resistant Gram-negative bacterial infections through enhanced membrane permeability and cell wall synthesis disruption.⁶⁸ An elongated version with an octaarginine transporter on the C-terminus, V-r8, showed great activity against antibiotic-insensitive bacteria due to enhanced cell-wall interaction (**Figure 1.2**). This conjugate also was able to eradicate pre-formed MRSA biofilms at 80 μ M concentration and persister cells at 10 μ M *in vitro*. In an *in vivo* skin wound biofilm model, a 0.05% solution of V-r8 significantly reduced bacterial load with no toxicity.⁶⁹

Blaskovich and coworkers designed selective membrane active vancomycin conjugates. Their design involved a threefold coupling of vancomycin to a lysine-rich electrostatic effector peptide sequence (EEPS) coupled to a lipid membrane insertive element. This EEPS mediates membrane binding in bacteria and mammalian cells. These cationic lipoglycopeptides showed enhanced activity against MRSA and other Gram-positive bacteria with some vancaptins showing a 100-fold improvement in MIC. Although they did not report intracellular activity, they focused on the mode of action these conjugates displayed in antibacterial activity. Membrane interaction of their vancaptins, rather than enhanced ligand binding or dimerization accounted for the excellent activity.⁷⁰

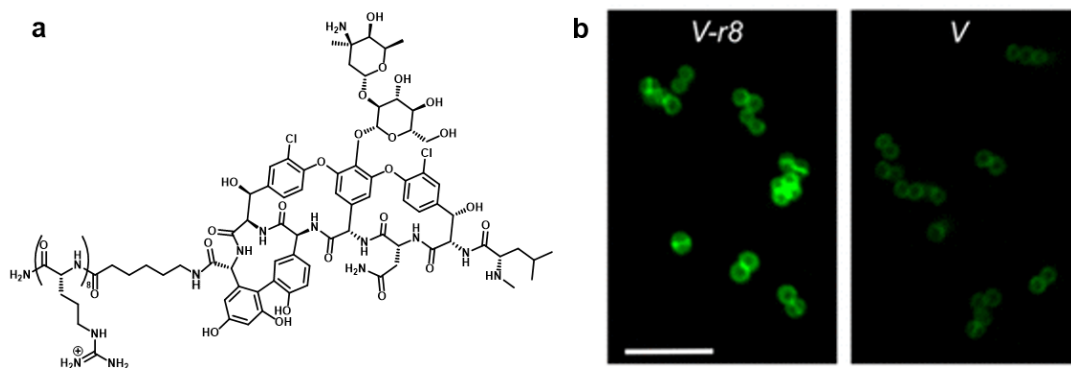


Figure 1.2. (a) structure of vancomycin-r8 (V-r8). (b) MRSA USA400 MW2 bacteria treated with FI-V-r8 exhibits greater cell-associated and protoplast-associated fluorescence than FI-V. Confocal microscopy of bacteria treated with 5 μ M FI-V-r8 and FI-V for 5 min. Reprinted (adapted) with permission.⁶⁹ Copyright (2018), American Chemical Society.

The penetrating activity of PP-G, a flexible guanidinium-rich polypeptide, was previously shown to be greater than that of Tat and oligo-arginine CPPs.⁷¹ The PP-G polypeptide was conjugated to vancomycin (**Figure 1.3 a**) using click chemistry and found to display exceptional intracellular and *in vivo* antibacterial activity against resistant *S. aureus*. This activity was through a dual mechanism of action; vancomycin targeting bacterial cell wall synthesis and the polypeptide targeting bacterial cell membrane disruption. This conjugate showed efficient mammalian cell internalization (**Figure 1.3 b**) to effectively eradicate extracellular and intracellular MRSA with an MIC of 3 μ M and an IMBC of 9 μ M, respectively. Additionally, this VPP-G conjugate eradicated extracellular and intracellular VRE with an MIC of 12 μ M and an IMBC of 48 μ M, respectively. In a mouse intravenous MRSA infection model, this VPP-G conjugate outperformed vancomycin and effectively eradicated MRSA at 72 mg/kg.

Other locations on vancomycin were explored as options in conjugation. The C-terminus and vancosamine locations of vancomycin were modified with PEG tethers of varying lengths to covalently conjugate vancomycin and the CPP transportan (TP10). Locations of conjugation on TP10 were also used. All conjugates showed increased antibacterial activity against resistant *S. aureus* and most showed activity against *E. faecium*. Notably, the conjugates displayed intracellular antibacterial activity against MRSA in HEK293 cells, and accumulation across the blood brain barrier.⁷²

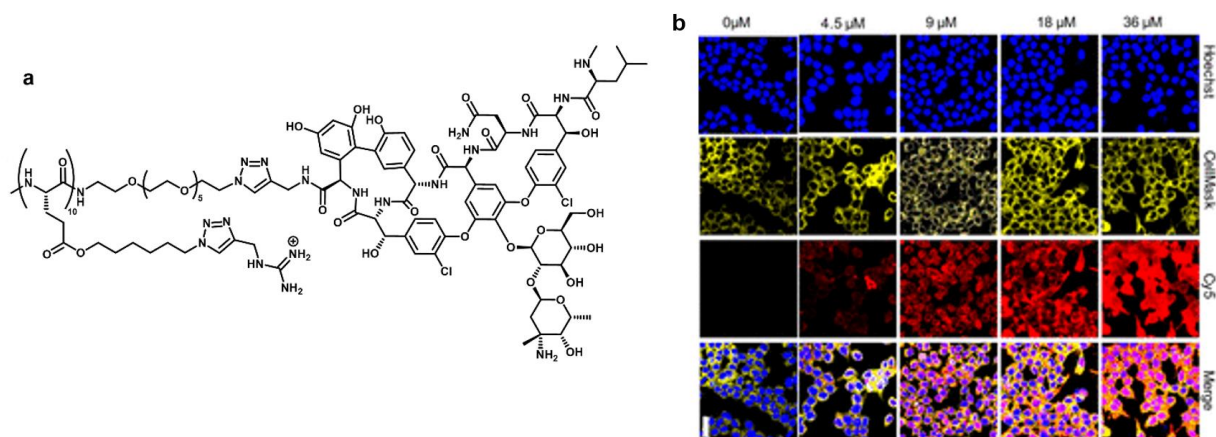


Figure 1.3. (a) Structure of VPP-G. (b) Cellular internalization and membrane penetration of VPP-G. Confocal images of RAW264.7 macrophages treated with Cy5-VPP-G of various concentrations. Blue: Hoechst; yellow: CellMask; red: Cy5-VPP-G. Scale bar: 20 μm . Reprinted (adapted).⁷¹ Copyright (2020), American Chemical Society.

Modifying different positions of vancomycin linkage to poly-arginine CPPs through a maleimide was investigated by the Uhl group.⁷³⁻⁷⁴ In one example, Muhlberg and coworkers explored conjugating triarginine and a fatty acid to different locations on vancomycin.⁷⁴ The most potent vancaptin conjugate was that which the peptide and lipid were conjugated from the vancosamine handle of vancomycin. This conjugate showed exemplary activity against VanA, VanB, and VanC resistant *enterococci*. The Uhl group also used the same maleimide linker strategy to conjugate a 6-mer polyarginine CPP on various positions of vancomycin.⁷³ In their lead conjugate, FU002, the hexaarginine was conjugated from the secondary amine handle of vancomycin and displayed activity against all three types of vancomycin-resistant *enterococci* with MICs less than 1×10^{-6} mmol/mL. Both examples of vancomycin conjugates showed negligible cytotoxicity at relevant concentrations, but the authors did not explore intracellular efficacy in mammalian cells.

1.3.2 Aminoglycoside-CPP conjugates

Aminoglycosides are a class of antibiotics traditionally used to treat Gram-negative infections, with inhibiting essential bacterial protein syntheses. Common antibiotics in this class include streptomycin, kanamycin, tobramycin, gentamicin, and neomycin. For instance, gentamicin is an antibiotic with potent antibacterial activity against extracellular bacteria, but fails

to accumulate intracellularly to treat internalized bacteria within mammalian cells. Gomasca and colleagues presented the conjugation of gentamicin to CPPs including Tat, α 1H and, α 2H (isolated from *Y. enterocolitica*) through a maleimide linker strategy.⁷⁵ These conjugates did not alter the *in vitro* bactericidal activity of gentamicin, however, the CPPs successfully delivered gentamicin to intracellular locations of HBMEC and HeLa cells and reduced the intracellular bacterial loads of Gram-negative pathogenic *E. coli*, *Shigella*, and *Salmonella*. Interestingly, the addition of non-conjugated CPPs and gentamicin also reduced the bacterial load compared to free gentamicin in *E. coli*-infected HBMEC by at least 5-fold, suggesting a synergy between the compounds.

The broad-spectrum aminoglycoside, kanamycin, is used to treat a range of bacterial infections and tuberculosis, but primarily used to treat Gram-negative bacteria. This antibiotic was conjugated to a synthetic cationic amphiphilic polyproline helix (CAPH), P14LRR using a reversible disulfide linkage.⁷⁶⁻⁷⁷ The disulfide moiety in the linker of the conjugate was shown to successfully reduce and release both free kanamycin and P14LRR-SH under physiological conditions (**Figure 1.4 A**). The reduction of disulfide and intramolecular rearrangement was modeled after a luciferin-r8 conjugate.⁷⁸⁻⁷⁹ Further, this conjugate showed impressive antimicrobial activity against a spectrum of Gram-negative pathogens including intracellular *Salmonella*, *Brucella*, and *Shigella* infections at 10 μ M concentration (**Figure 1.4 B**). Not only did this conjugate display excellent antimicrobial activity against a broad-spectrum range of susceptible and drug-resistant strains, but it was also able to efficiently accumulate within mammalian macrophage cells to effectively eradicate intracellular *Mycobacteria* at 5 and 10 μ M concentrations (**Figure 1.4 C**) and significantly reduce *Salmonella* in a *C. elegans in vivo* model. This conjugate also was found to disrupt 30%, 50%, 70%, and 80% of mature Gram-positive *S. epidermidis* biofilm mass at 8, 16, 32 and 64 μ M, respectively; and clear Gram-negative *P. aeruginosa* and *A. baumannii* at 32 μ M.⁷⁶⁻⁷⁷

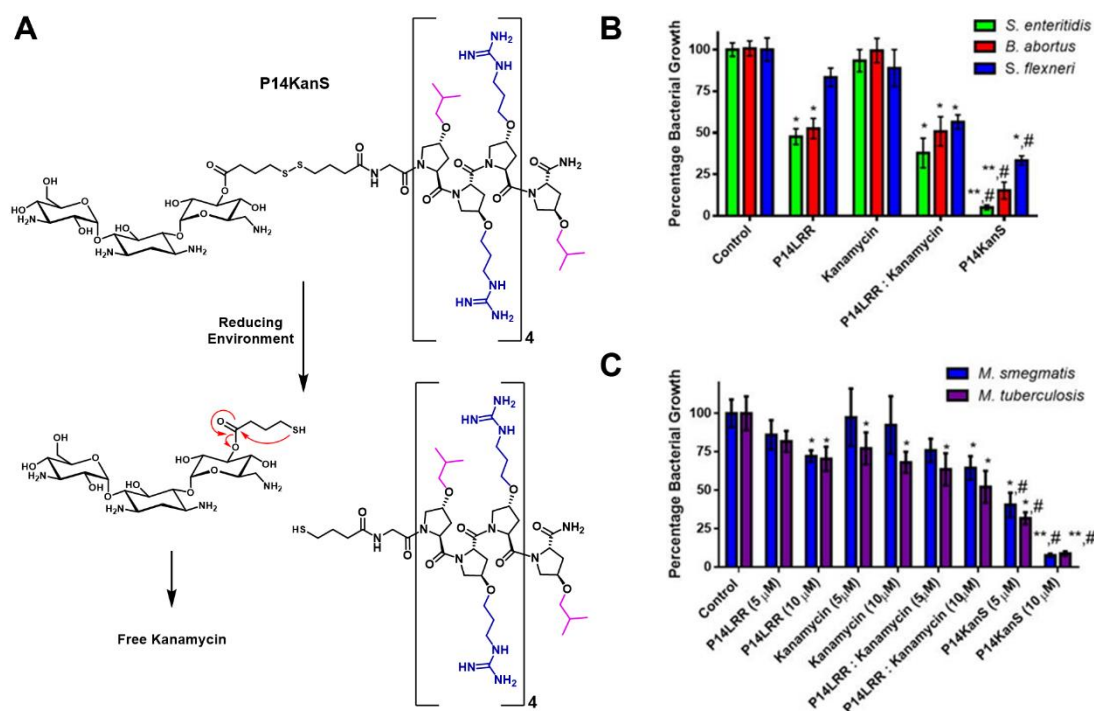


Figure 1.4 (A) Kanamycin-CAPH releases both antibiotics under reducing conditions. (B, C) Intracellular clearance of selected pathogens with P14LRR, kanamycin, and P14KanS. Reprinted (adapted).⁷⁶ Copyright (2016), American Chemical Society.

Membrane-active antibiotic-peptide conjugates (MAAPCs) composed of a short sequence from the CPP penetratin and the broad-spectrum aminoglycoside tobramycin, traditionally used for Gram-negative bacteria, were designed and investigated by Deshayes and coworkers. These conjugates had selective bactericidal membrane permeation and displayed broad-spectrum antibacterial activity including persister cells of pathogenic *E. coli* and *S. aureus*. Although the authors reported selective bacterial membrane disruption over mammalian membranes, intracellular activity against bacteria within mammalian cells was not noted.⁸⁰

Paromomycin (PMM) is another poorly absorbed broad-spectrum aminoglycoside antibiotic used to treat infections caused by bacteria and parasites. This antibiotic was conjugated to Tat peptide through a flexible PEG spacer and amide linkage. This conjugate successfully accumulated intracellularly in *Leishmania* parasites and penetrated within *Leishmania* ulcers better than the poorly penetrating paromomycin (**Figure 1.5**). Although intracellular accumulation into the cytoplasm of parasites was observed, killing assays and mammalian cellular accumulation were not investigated.⁸¹

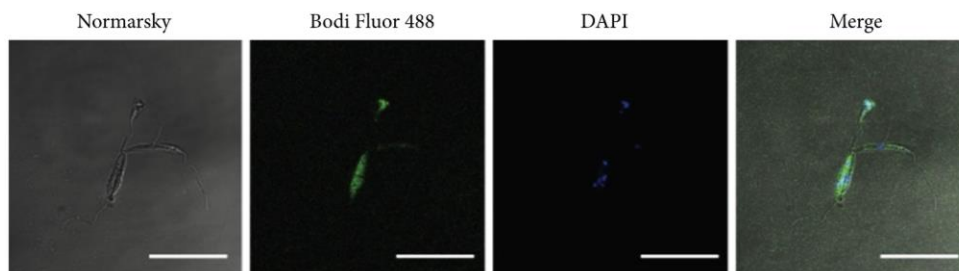


Figure 1.5. Fluorescence of *Leishmania donovani* promastigotes incubated with Bodi Fluor 488-labeled PMM-Tat conjugate (9 μ M, 4 h, 26°C) and stained with DAPI (5 μ g/mL) immediately before observation, unfixed. Settings: conjugate (green, = 488 nm/ = 520 nm); DAPI (blue, = 350 nm/ = 460 nm). Bar = 20 μ m. Experiment is representative of two other ones performed independently. Reprinted (adapted) with permission.⁸¹ Copyright (2017), Hindawi.

1.3.3 Other small molecule antibiotic-CPP conjugates

Levofloxacin and a truncated levofloxacin-Q, broad-spectrum fluoroquinolone antibiotics, were conjugated to the cyclic CPP R4W4 and its linear equivalent through the carboxylic acid group and the terminal lysine on the CPP.⁸² It was shown that the levofloxacin-Q-[R4W4] conjugate and non-conjugated mixture of levofloxacin-Q and CPP had increased antibacterial activity against MRSA and *Klebsiella* with MICs of 8 and 32 μ M compared to inactive levofloxacin-Q alone. Interestingly, the levofloxacin-cCPP conjugate did not show an improvement in antibacterial activity compared to levofloxacin alone. The conjugates incorporating the linear equivalent of the CPP were less potent than those with cCPP. Ghaffar and coworkers also investigated amide-linked and ester-linked levofloxacin-CPP conjugates.⁸³ Unfortunately, neither of their levofloxacin-Tat conjugates or the non-conjugated mixture of levofloxacin and Tat cleared bacterial infections more efficiently than antibiotic alone. Another levofloxacin (LVX)-CPP was designed with an acid-reducible thioester linkage.⁸⁴ Upon endocytosis-mediated internalization of conjugate in mammalian cells, the acidic internal endosomal conditions were proposed to cleave the thioester linkage between antibiotic and the CPP, CGAFPHR. The authors showed successful cleavage and release of CPP and free LVX in a 4.5 pH environment. However, in regard to bactericidal activity, only a modest improvement in antibacterial activity was seen for the conjugate against *S. aureus* and *E. coli* cultures.

The treatment of intracellular *Listeria monocytogenes* using methotrexate (Mtx) and a synthetic cyclohexyl- and guanidinium-rich CPP was explored by the Kelley group.⁸⁵ This Mtx-peptide exhibited enhanced bactericidal activity compared to Mtx with MICs of 3.7 and 13.1 μ M

against extracellular and intracellular *L. monocytogenes*, respectively. The authors tailored the CPP and demonstrated that charge and hydrophobicity play a role in mammalian cellular uptake and intracellular location in promoting the most efficacious clearance of intracellular *Listeria*. Another fascinating pro-drug delivery strategy explored by the Kelley group was designed and investigated using methotrexate (Mtx) to treat intracellular phagosolic mycobacterial infections.⁸⁶ Their intricate three-fold drug design included a cyclohexyl and guanidinium-rich CPP conjugated to Mtx with a cleavable negatively charged shielding peptide to promote endosomal uptake (**Figure 1.6**). Once inside the cell and in the vicinity of intracellular mycobacteria that secrete β -lactamases when active, they showed that the cephalosporin linkage between the shielding peptide and Mtx-CPP was cleaved. The internalized Mtx-CPP successfully accumulated in mammalian cells and eradicated localized intracellular mycobacteria in RAW264.7 cells with minimal observed cellular toxicity.

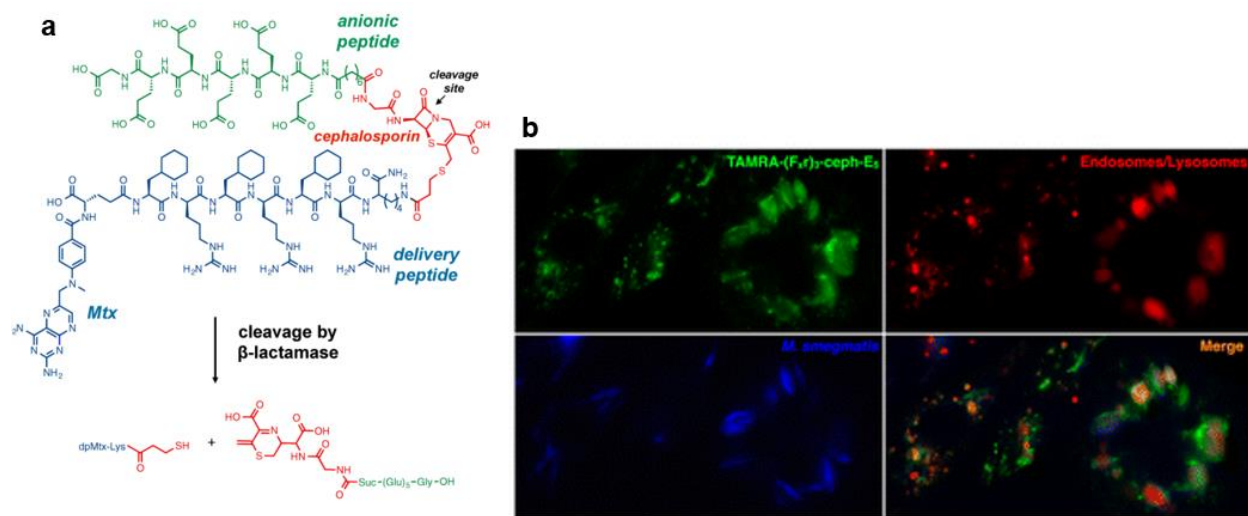


Figure 1.6. (a) Drug design of (F_{xr})₃-Ceph-E₅ conjugate. (b) Wide-field fluorescence microscopy of TAMRA-(F_{xr})₃-Ceph-E₅ (6 μ M) with endosomes/lysosomes (dextran, Alexa Fluor 647; 10 kDa MW) and *M. smegmatis* (labeled with Marina Blue-NHS) after 24 h of incubation in RAW264.7 macrophages. Reprinted (adapted) with permission.⁸⁶ Copyright (2015), American Chemical Society."

More recently, the Kelley group facilely conjugated the same cyclohexyl and guanidinium-rich CPP to nalidixic acid, a quinoline-based Gram-negative antibiotic with poor pharmacokinetics.⁸⁷ In doing so, their conjugates exhibited increased bacterial uptake in *S. aureus* and enhanced

antibacterial activity against sensitive and drug-resistant *S. aureus*. The authors determined that the CPP conjugation allowed nalidixic acid to overcome the intrinsic resistance of *S. aureus* and inhibit bacterial DNA gyrase as the mode of action.

Fosmidomycin, an antibiotic used to treat Gram-negative bacterial and parasitic infections, was improved through amide conjugation to the CPP octa-arginine. Not only was this covalent conjugate and non-covalently bound complex effective against parasitic *Plasmodium* and *Toxoplasma*, the conjugate was effective in accumulation within the parasite or bacteria, and displayed activity against *Mycobacterium*, for which fosmidomycin is ineffective (**Figure 1.7**).⁸⁸ Another antiparasitic drug, miltefosine (MT), was conjugated to the CPP Tat via a cleavable disulfide or non-cleavable thioether linkage. Both conjugates enter and kill both promastigote and intracellular amastigote forms of *Leishmania*. Additionally, the Tat peptide mediated the accumulation of MT in both *Leishmania*, macrophages, and intracellular *Leishmania*, demonstrating an effective strategy in treating intracellular infections.⁸⁹⁻⁹⁰

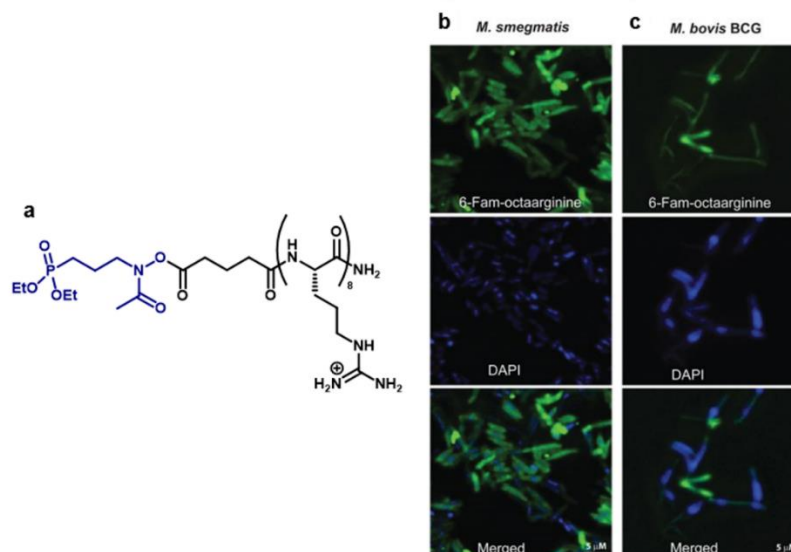


Figure 1.7. (a) Structure of fosmidomycin-R8 (FAM-octaarginine). (b and c) Fluorescent micrographs of *Mycobacterium smegmatis* (b) and *Mycobacteria bovis* BCG (c) showing incorporation of FAM-octaarginine (green). DAPI (blue) is shown as a reference for bacterial DNA. Reprinted with permission.⁸⁸ Copyright 2013, The American Society of Microbiology.

Antimicrobial photodynamic therapy (PDT) has shown promise in treating infections, due to the mechanism of generating singlet oxygen and radical species to kill bacteria. Bourré and coauthors developed a maleimide linked Tat-porphyrin conjugate.⁹¹ Interestingly, Bourré's Tat-

porphyrin conjugate was effective in killing both Gram-negative and Gram-positive bacteria where PDT is typically less effective in Gram-negative bacteria. It was concluded that the activity of this Tat conjugate was the result of the combined effects of membrane destabilization from Tat and induction of toxic reactive oxygen species from the porphyrin photosensitizer.

1.3.4 AMP-CPP conjugates

Antimicrobial peptides and cell penetrating peptides can be combined to afford a therapy with the shared advantages of each. For instance, Wang and coworkers constructed a cell-penetrating peptide and antimicrobial peptide (CPP-AMP) conjugate.⁹² They used N2, an AMP with excellent antimicrobial activity against Gram-negative infections such as *E. coli* and *S. typhimurium*, and conjugated it via an amide linkage to the CPPs, Tat or bLFcin₆. Intracellular uptake was visualized, and both conjugates were localized primarily in endosomes of RAW264.7 macrophage cells (**Figure 1.8**). The N2 peptide alone did not display strong fluorescence, justifying that the CPP-conjugate improved intracellular delivery. Both conjugates also displayed enhanced *in vitro* bactericidal activity against *Salmonella* as compared to N2 alone, and both eradicated over 90% of intracellular *Salmonella* after 3 hours at 10 μ M concentration in macrophage cells. These authors also designed a CPP-AMP conjugate using a cathepsin-cleavable linker conjugating Tat to a similar cyclic AMP, N6.⁹³ They showed enhanced cellular uptake in macrophage cells compared to N6, and over 65% enhanced antimicrobial activity against intracellular *S. typhimurium* in RAW264.7 macrophage cells with almost no hemolysis or cytotoxicity. The survival of *S. typhimurium*-infected mice also improved by over 65% compared to N6.

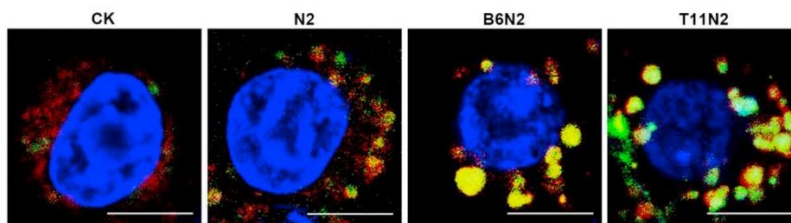


Figure 1.8. FITC-labeled CPPs-N2 uptake in RAW264.7 cells. Partially enlarged details of figures analyzed by confocal microscopy. Scale bar = 7.5 μm . CK: Control of free FITC. Cells were incubated with 5 μM FITC-labeled CPPs-N2 conjugates (green); the nucleus and cell membrane were counterstained with Hoechst 33342 nucleus stain (blue) and FM 4–64 endosomal membrane stain (red), respectively. Reprinted with permission.⁹² Copyright 2018, from Elsevier.

AMPs magainin and M15 were conjugated to polyarginine, R9, through simple amide linkage and peptide elongation.⁹⁴ Both AMP-CPP conjugates displayed broad-spectrum antimicrobial activity with a modest increases against Gram-positive activity and a 4- to 16-fold increase against Gram-negative bacteria compared to AMPs alone. The conjugates also showed an increase in bacterial internalization in *E. coli* and DNA affinity. Thus, this suggests additional antimicrobial functions like internal targeting of the AMP-CPP conjugates, adding to the AMP membrane disruption as the primary mode of action.

Using a similarly straightforward peptide synthesis and elongation strategy, a hydrophobic peptide pheromone (DILIVGG) of *Streptococcus agalactiae* and a CPP (KERKKRRR) were conjugated and investigated by the Lyu and Shan groups.⁹⁵ A series of conjugates with truncated and rearranged amino acids were synthesized to explore the best conjugate with targeted intracellular antibacterial activity. The authors reasoned that adding the hydrophobic pheromone to the cationic CPP, would not only increase the specificity of the conjugate but also increase the amphiphilicity to impart antibacterial character to the conjugate, as amphiphilicity is a common characteristic of AMPs. Thus, they proposed a specific, antimicrobial, and cell penetrating therapy. Multiple conjugates displayed good broad-spectrum antibacterial activity against *S. aureus*, *S. epidermidis*, *S. agalactiae*, *E. coli*, *P. aeruginosa*, and *S. typhimurium*. These peptides also reduced *S. agalactiae* load by over 95% in RAW264.7 macrophage cells and reduced the bacterial load in a mouse *in vivo* model.

Peptide nucleic acids (PNAs) are conjugates that offer a delivery method for antimicrobial gene silencing nucleic acids in order to reach their intracellular targets within mammalian and/or bacterial cells. Through the conjugation of oligonucleotides to CPPs to generate a PNA, enhanced

intracellular and gene-specific killing of bacteria can be accomplished. A phosphorodiamidate morpholino oligonucleotide was amide coupled to a CPP derived from a protein found in human T cells, YARVRRRGPRGYARVRRRGPRRC, via an amide linkage. This conjugate was used to target a highly conserved *Gyrase A* gene of *E. coli* and change gene expression to ultimately kill bacteria.⁹⁶⁻⁹⁷ Furthermore, this conjugate was effective in accumulating within bacteria due to its CPP component and successful in inactivating a broad-spectrum of bacteria with the conserved gene sequences. The Seleem group has used a similar strategy and also used PNAs that silence essential genes within bacteria.⁹⁸ The nucleic acid sequence targeted the *rpoA* gene, which encodes a subunit of RNA polymerase that is essential for *L. monocytogenes* viability. Through conjugation of CPPs (Tat, (RXR)₄XB, and (RFR)₄XB) to the nucleic acid sequence, the authors demonstrated that an *L. monocytogenes* infection could be cleared in *in vitro* culture within 20 minutes at 8 μ M and in infected J774A.1 macrophage cells after 4 hours at 8 μ M. These PNAs also displayed nearly 100% reduction in *L. monocytogenes* in a *C. elegans* model at 16 and 32 μ M.

1.4 Conclusions and Perspective

Nearly a century has passed since the discovery of the first antibiotic. In this “golden era”, optimism for treating diseases reached its peak, and quickly faded as antibiotic resistance overshadowed the practicality of permanently curing transmissible diseases. The modification of existing antibiotic scaffolds by conjugation with cell-penetrating peptides to provide drug-peptide conjugates shows potential in reviving ineffective treatments. Accessing internalized targets, improving pharmacokinetics, and managing resistance are all advantages in employing CPPs.

In this review, we highlighted many examples of antibiotic-CPP conjugates with improved antibacterial activity compared to the unaccompanied antibiotic. Many conjugates displayed activity against a range of drug-resistant bacteria and bacteria within mammalian cells. Traditional antibiotics such as aminoglycosides and glycopeptides were demonstrated to reach intracellular locations with little to no toxicity to mammalian cells. The CPPs not only play a role in accessing previously inaccessible locations, but also may interact with bacterial membranes in a synergistic fashion to improve the bactericidal action.

Increasing the arsenal of antimicrobial drugs from those previously dismissed due to poor internalization or disposition to resistance can be achieved. Additional optimization and translation to clinical trials for these conjugates is highly anticipated for the future. The outlook is bright for

antibiotic-CPP conjugates, and these therapies provide a potential avenue to address challenging bacterial infections.

CHAPTER 2. DESIGN, SYNTHESIS, AND INVESTIGATION OF BROAD-SPECTRUM, DUAL-THERAPEUTIC VANCOMYCIN-CAPH CONJUGATES

2.1 Introduction

Vancomycin is a glycopeptide antibiotic widely used to treat Gram-positive bacterial infections. Since its introduction more than 50 years ago, vancomycin has been regarded as critically important for human health by the World Health Organization.⁹⁹ In addition, it is deemed the “antibiotic of last resort” to treat life-threatening methicillin-resistant *S. aureus* infections. Vancomycin showed great promise for 30 years until a range of vancomycin-resistant strains of bacteria surfaced.¹⁰⁰ Vancomycin-resistant enterococci and methicillin-resistant *S. aureus* are listed as “serious threats” by the CDC due to the limited options in treating these infections.

In drug-sensitive bacteria, vancomycin forms five hydrogen bonds with a pentapeptide on Lipid II in the peptidoglycan layer of a Gram-positive cell wall. Specifically, vancomycin binds the terminal D-Ala-D-Ala sequence (**Figure 2.1 A**). Through this interaction, vancomycin blocks the bacterial transpeptidase enzyme from cross-linking the components of the peptidoglycan to form a fully intact cell wall. The resulting internal pressure bursts the cell wall, thus killing the bacteria. For vancomycin-resistant strains (VanA, VanB, and VanC) of bacteria, the target of vancomycin binding is altered to prevent the tight binding necessary for activity.¹⁰⁰⁻¹⁰¹ A lactic acid (VanA, VanB) or serine (VanC) residue is substituted in place of the terminal alanine residue in the pentapeptide chain. In the altering an amide bond to an ester (VanA, VanB), the loss of a single hydrogen bond and the destabilizing lone pair/lone pair electrostatic interaction introduced with the ester oxygen results in a 1000-fold decrease in binding affinity (**Figure 2.1 B**).¹⁰²⁻¹⁰³ VanC resistance has been documented only in a few clinical isolates. The hydroxymethyl sidechain of D-serine is thought to sterically disrupt the optimal geometry for vancomycin binding, resulting in about a sixfold decrease in affinity for vancomycin.¹⁰⁴⁻¹⁰⁵ The decrease in binding affinity allows the transpeptidase enzyme to displace vancomycin and crosslink the peptidoglycan strands, leading to an intact cell wall. There is an urgent need to produce modified vancomycin antibiotics that can restore or enhance binding affinity to the peptidoglycan or create an alternative route to combat resistance.

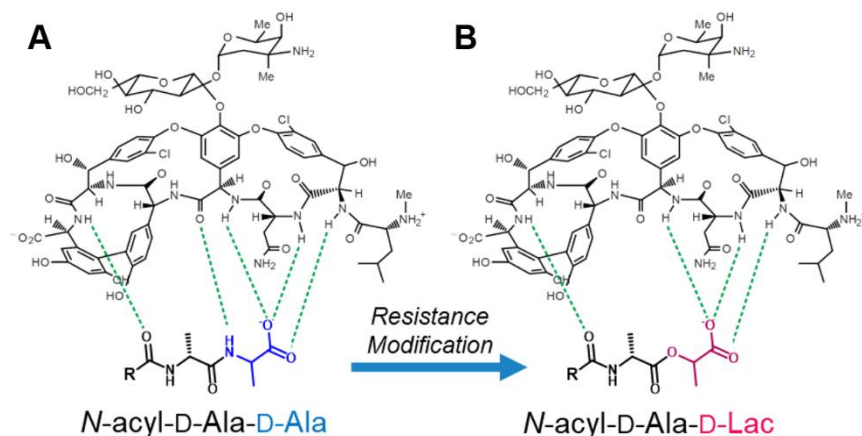


Figure 2.1 The binding affinity of vancomycin to the peptidoglycan pentapeptide (A) decreases due to the loss of a hydrogen bond in resistant bacteria (B).¹⁰⁴

Vancomycin's inability to traverse bacterial membranes accounts for its narrow spectrum of activity against Gram-positive infections. Gram-positive bacteria possess only the cytoplasmic membrane; therefore, vancomycin can target the peptidoglycan layer without crossing a bacterial membrane (**Figure 2.2 A**).¹⁰⁶ Vancomycin is ineffective in treating Gram-negative infections due to the presence of an outer membrane; therefore vancomycin cannot localize with the peptidoglycan target (**Figure 2.2 B**).

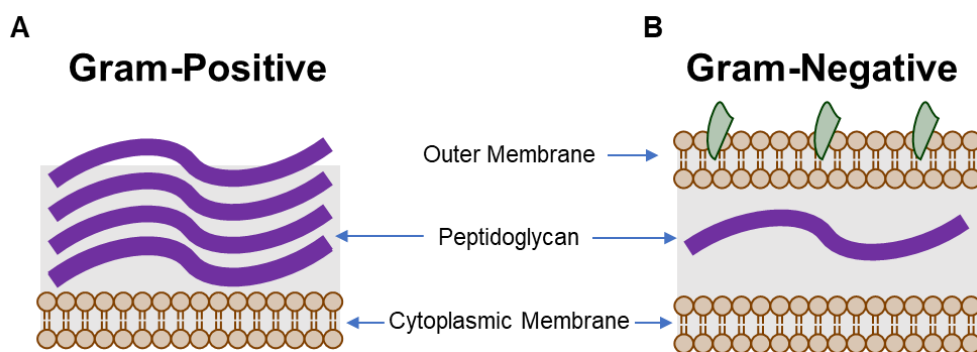


Figure 2.2. The differences in bacterial walls between (A) Gram-positive and (B) Gram-negative bacteria.

Not only is resistance an impediment in treating infections with vancomycin, but mammalian membrane penetration is also an obstacle in vancomycin reaching its full therapeutic potential. Many pathogens causing acute infections are cleared through the host immune system

or by conventional antibiotics. However, a subcategory of intracellular pathogens can establish persistent and chronic infections by residing and proliferating inside host immune cells, primarily macrophages.¹⁰⁷⁻¹⁰⁸ *Staphylococcus aureus* is a well-known Gram-positive pathogen that possesses diverse survival strategies among strains and host cell types, including intracellular survival. Once internalized by phagocytosis, *S. aureus* can replicate in the phagosome or freely in the cytoplasm of its host cells. It induces cell death mechanisms upon escape from the phagosome to the cytoplasm of phagocytes.¹⁰⁹⁻¹¹⁰ Options in treating intracellular *Staphylococcus* infections within mammalian cells are limited due to the poor mammalian membrane permeability of many hydrophilic antibiotics, including vancomycin and other glycopeptides.¹²⁻¹⁵ Some modifications, such as the lipophilic encapsulation of these hydrophilic antibiotics like vancomycin and gentamicin, has been used to bypass the mammalian membrane and deliver drugs to the mammalian intracellular environment.¹¹¹⁻¹¹³ Such modifications can improve delivery, but do not alter bacterial resistance to vancomycin.

Vancomycin exhibits poor efficacy against intracellular methicillin-susceptible, methicillin-resistant, vancomycin-intermediate and vancomycin-resistant *S. aureus*.¹¹⁴⁻¹¹⁵ Therefore, even if an antibiotic is effective against extracellular bacteria, it may be ineffective in clearing intracellular pathogens that may become infectious in time. In order to effectively treat intracellular bacteria, and open the possibility of broad-spectrum activity, increasing the hydrophobicity of vancomycin to permeate mammalian membranes is necessary for improved vancomycin activity. Utility of cell-penetrating agents with vancomycin has also been explored to improve mammalian cell penetration and activity, yet improvements can be made.

2.1.1 Vancomycin modifications and derivatives

Since the discovery of vancomycin resistance in the late-1980s, many approaches have been used to increase the potency of vancomycin toward drug-resistant and/or intracellular Gram-positive bacteria, in addition to increase the spectrum of activity against Gram-negative bacteria.¹¹⁶⁻¹¹⁷ Bringing new drugs to the clinic as resistance develops is a constant battle, but continual progress is ongoing. Over the years, novel glycopeptides such as teicoplanin and the more recent telavancin, dalbavancin, and ortavancin have gained approval and have had longevity in clinic due to their relatively slow development of resistance.¹¹⁸ This progress has been made through the careful study of vancomycin's pharmacokinetic properties and mode of action.

A number of core and peripheral modifications have been made to vancomycin to improve binding to the target pentapeptide and the resistant lactic acid adaptation. Core modifications to the amino acid residues required significant synthetic manipulations but afforded an increase in affinity for terminal D-Ala-D-Lac, thus improving strategies in treating resistant bacteria.^{60, 103, 119-122} Peripheral modifications provide synthetic ease, and the use of the carboxylic acid terminus as a handle for manipulation has provided the opportunity to form an additional direct hydrogen bond with the pentapeptide ligand.¹²³ The Halder group has made use of the C- terminal modification of vancomycin, in addition to using membrane and cell wall interacting components through the addition of lipophilic sugars,¹²⁴ cationic lipids,⁵²⁻⁵⁶ and pyrophosphate binding moieties.⁵⁷⁻⁵⁸ Furthermore, the incorporation of sulfonium-based cationic lipophilic components to the periphery of vancomycin led to enhanced interaction with the negatively charged bacterial cell membrane and increased bactericidal activity through membrane disruption.⁵⁹ Boger and coworkers have built upon the chlorobiphenyl (CBP) modification used in ortavancin to incorporate a quaternary ammonium ion for membrane activity resulting in better bactericidal activity.⁶⁰⁻⁶¹ Other lipophilic and cell surface-interacting vancomycin derivatives have shown increased activity in treating vancomycin resistance as described in Chapter 1.⁶²⁻⁶⁷ Clearly, modifying vancomycin to treat resistant and broad-spectrum bacteria is continuing to broaden our future arsenal of antibiotic treatments.

2.1.2 Cationic amphiphilic polyproline helices (CAPHs)

A novel class of peptides with both antimicrobial and cell-penetrating characteristics have been developed and engineered in the Chmielewski lab to serve as dual therapeutic agents. These cationic amphiphilic polyproline helices (CAPHs) are composed of both cationic and amphiphilic components appended to a polyproline backbone. The polyproline backbone allows for the peptide to adopt a stable, yet flexible, polyproline type II (PPII) secondary structure (**Figure 2.3 A**).¹²⁵ Proline-rich antimicrobial peptides (PrAMPs) present an advantage over traditional AMPs in that many are not membrane lytic.³⁹ PrAMPs can also be translocated within both bacterial and mammalian cells, and they have multiple intracellular targets within bacteria that can slow the emergence of resistance.¹²⁶⁻¹²⁷ Although CAPHs are designed with unnatural proline derivatives, they share some characteristics such as secondary structure, antibacterial properties, and non-lytic membrane proclivity.

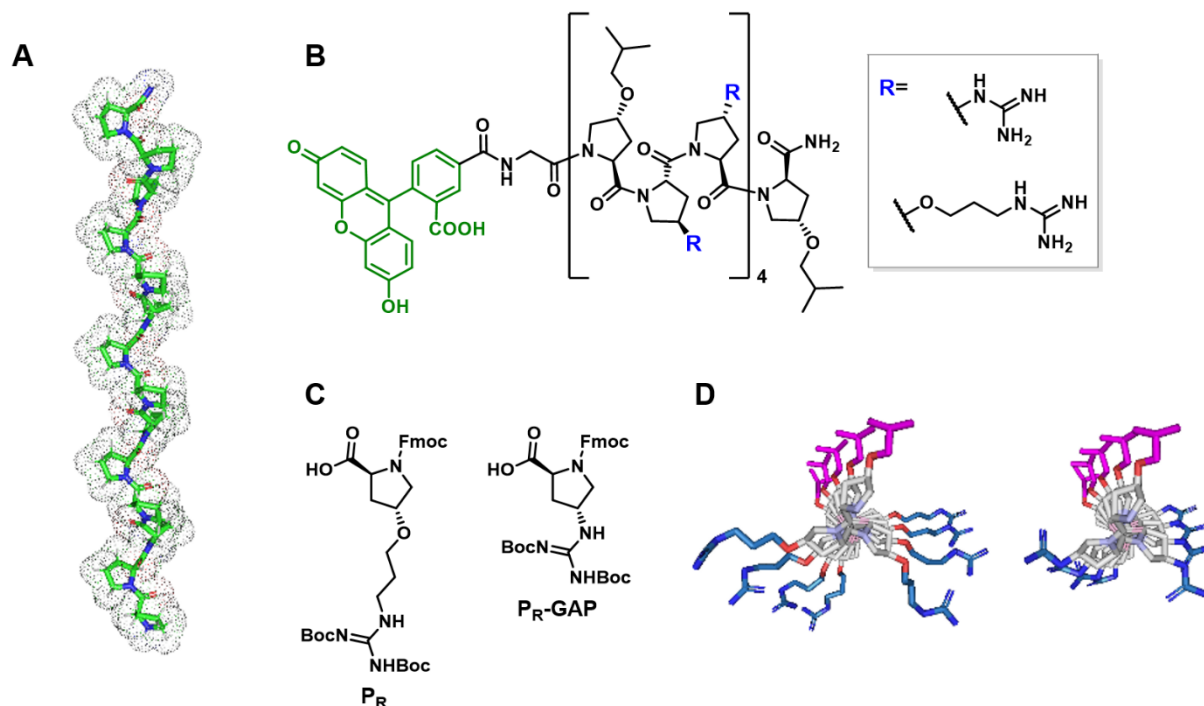


Figure 2.3. (A) Side view of a polyproline helix, PPII secondary structure. (B) Structure of CAPHs with P_R or P_R-GAP cationic amino acids. (C) Cationic amino acids Fmoc-P_R and Fmoc-P_R-GAP. (D) Top view of CAPHs, P14LRR and P14GAP.

Amphiphilicity with cationic and hydrophobic residues on the faces of the CAPH PPII prove to be vital in the specificity and intracellular location of mammalian cellular accumulation. Previous work has revealed that variations in the hydrophobic residues of a CAPH can result in changes to cell selectivity³⁶ and changes in mammalian subcellular location.^{35, 40-42} Changes in the cationic residues generally resulted in changes in mammalian cellular uptake. For instance, the CAPHs with an arginine-like guanidinium modification accumulated in mammalian cells two times greater than the lysine-like primary amine modification.³⁷ By substituting the P_R-GAP amino acid for P_R, consequently shortening the length of the guanidinium group from the proline backbone, a more rigid scaffold was designed (**Figure 2.3 B, C, D**). This **P14GAP** peptide had better cell accumulation than **P14LRR** (up to 7-fold), and this peptide displayed significantly increased antibacterial activity against a range of pathogens. **P14GAP** was also able to completely eradicate two major pathogenic bacteria, *A. baumannii* and *Shigella*, from macrophage cells, as well as drastically decreased the amount of *Listeria*, *MRSA*, and *S. epidermidis* in cyto experiments.⁴² The

dramatic increase in cell uptake and intracellular clearance support the potential of this peptide for antimicrobial therapy.

The length of the CAPHs also play a role in activity. Because CAPHs are composed of a repeating triad, peptide length increases corresponded to the addition of one repeating triad unit where the net charge of the peptide increased by two. Conventionally, the increase in charge of a peptide has resulted in increased mammalian cell uptake while also increasing bacterial membrane disruption and minimum inhibitory concentration (MIC).¹²⁸⁻¹²⁹ The increase in charge for CAPHs ranging from P8LRR-P17LRR yielded consistent corresponding increases in mammalian cell uptake and MICs.^{35, 37, 43, 45} This class of peptides has shown its potential in offering tunable antibiotic activity.

2.1.3 Vancomycin-Cationic Amphiphilic Polyproline Helix (CAPH) Strategy

Conjugating lipophilic and cationic components to vancomycin has recently appeared as a promising strategy in treating difficult bacterial infections by combining different mechanisms of action to overcome resistance. Vancomycin homodimers were some of the first conjugates to address vancomycin resistance. The strategy behind dimerizing vancomycin lies in the back-to-back hydrogen bond-based dimerization that vancomycin adopts in binding its target.¹³⁰ Therefore, multiple adaptations of linker strategies to covalently dimerize vancomycin have shown an increase in binding affinity and resulting increases in potency.¹³¹⁻¹³⁷

The conjugation of vancomycin to other antibiotics has also been explored. For instance, nisin, an antimicrobial peptide that binds pyrophosphate of lipid II, was conjugated to vancomycin with successful increase in potency against resistant bacteria.¹³⁸ Conjugates between vancomycin and β -lactam antibiotics have been investigated to improve potency against sensitive and resistant bacterial strains.¹³⁹⁻¹⁴¹ Conjugation of vancomycin to cathelicidin-related antimicrobial peptides (CRAMPs) provided vancomycin with an additional mode of action against bacteria, disruption of cytoplasmic membrane. A variety of linkers between the vancomycin and CRAMPs was used and many of the compounds showed broad-spectrum activity in addition to antibiofilm formation.¹⁴²

Although vancomycin shows poor mammalian membrane penetration, ortavancin and telavancin glycopeptides show an improvement in clearing intracellular infections.¹¹⁴⁻¹¹⁵ Strategies of conjugating cell-penetrating peptides to vancomycin have been reported (described in Chapter 1). Herein, we wish to describe further strategies in targeting bacteria residing within mammalian

cells. The Chmielewski group has described a reversible kanamycin-P14LRR conjugate that that potently clears intracellular pathogens (described in Chapter 1). Given the success of this kanamycin-CAPH conjugate design, we were poised to investigate new antibiotic-CAPH conjugates with vancomycin replacing kanamycin. This approach would allow, under similar physiological conditions, the delivery of a poorly penetrable antibiotic and antimicrobial peptide to cellular locations without compromising mammalian cell integrity (**Figure 2.4 B**). We hypothesize that the conjugate will be capable of a dual method of killing bacteria. As a conjugate in the extracellular environment, we anticipate bactericidal activity; and, once delivered inside a mammalian cell, we anticipate the reduction of the disulfide linkage to release both antibiotics for dual killing of bacteria.

Herein, we designed a reducible disulfide-linked conjugate, **VanP14S**, and a non-reducible conjugate, **VanMalP14** (**Figure 2.4 A**). The use of two different conjugation approaches will allow us to determine whether intracellular release of vancomycin and **P14LRR-SH** is necessary for the most potent activity in mammalian cells. The **VanMalP14** conjugate linkage was chosen to undergo a facile thiol-maleimide click reaction with **P14LRR-SH**. Moreover, a fluorescent analog, **FIVanP14S**, was synthesized to monitor vancomycin-CAPH activity in fluorescence-based assays. These vancomycin-peptide design modifications had the potential to yield interesting antibiotic activity against a range of pathogens through the combination of a Gram-positive antibiotic, vancomycin, and broad-spectrum antibiotic, **P14LRR**.

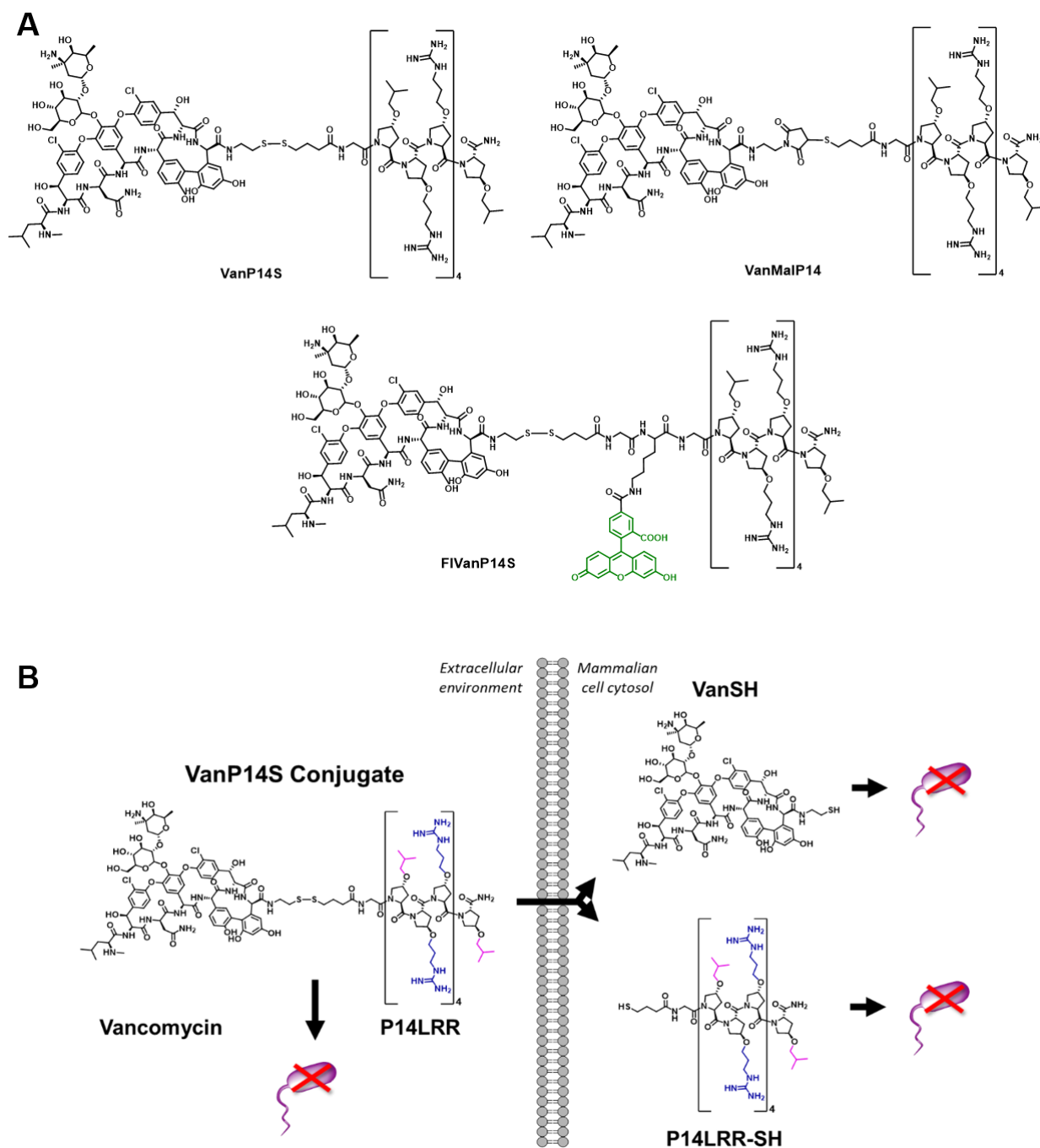
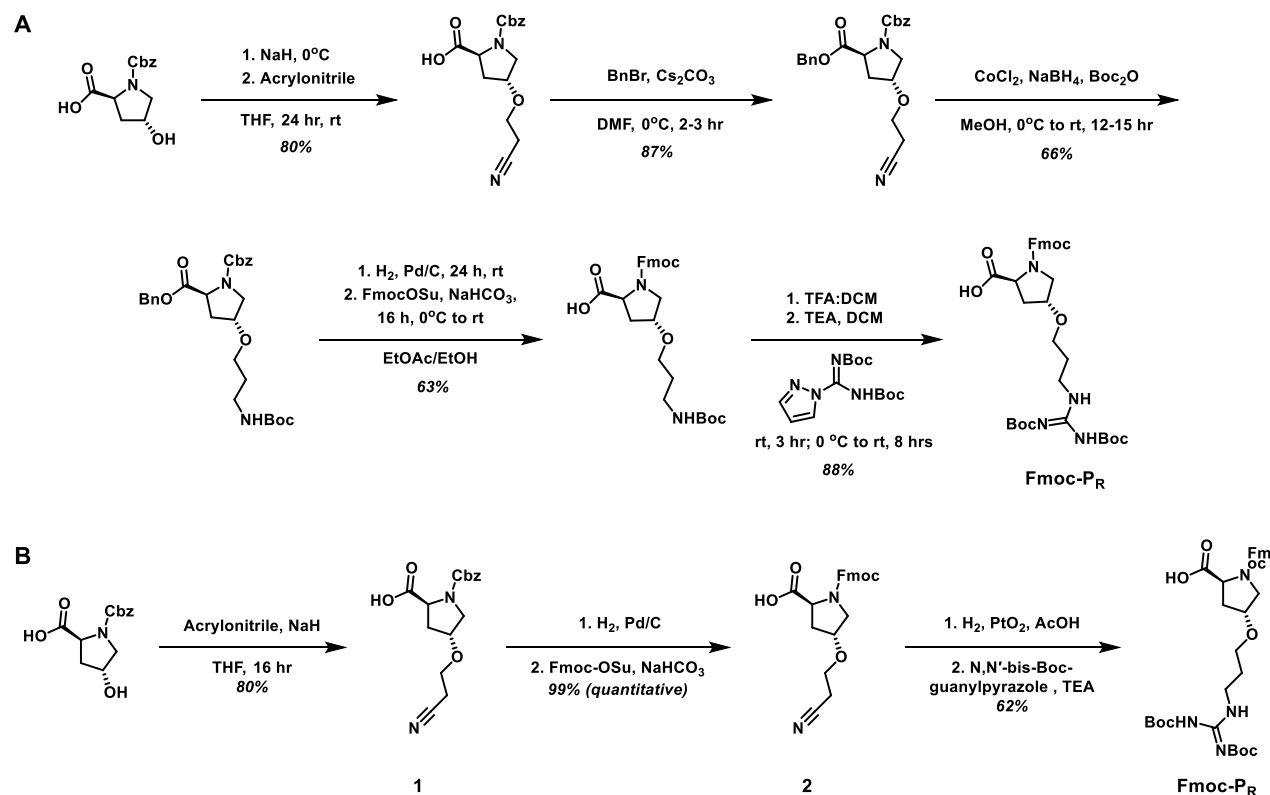


Figure 2.4. (A) Structures of vancomycin-CAPH conjugates. (B) Vancomycin-CAPH strategy in clearing drug-resistant and intracellular bacteria residing in mammalian cells. Once within a reducing environment, the disulfide linker releases vancomycin and CAPH antibiotics.

2.2 Results and Discussion

2.2.1 Synthesis of CAPH amino acids and CAPHs

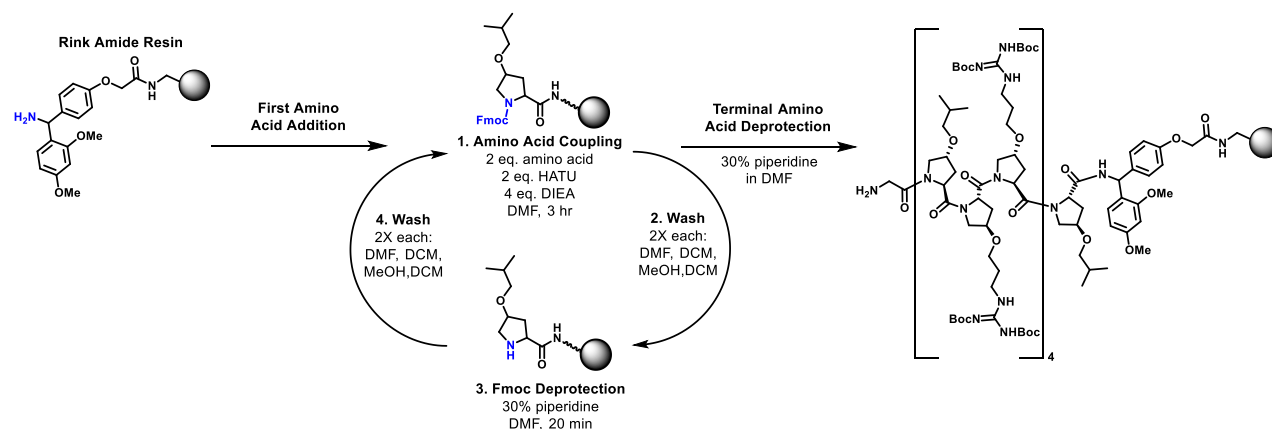
To begin the synthesis of the conjugates described above, we needed the amino acids for the synthesis of the CAPH, **P14LRR-SH**. Two unnatural amino acids have been used previously, Fmoc- P_L and Fmoc- P_R . The synthesis of Fmoc- P_R has been a challenging step in the synthesis of CAPHs since their development.³⁷ Over the years, minor adjustments in protocol have been made to increase the yield of the final amino acid from the original published yield of 19% to 25% using a similar five step reaction scheme (**Scheme 2.1 A**). Recently, I developed a new strategy has improved the synthesis of P_R through fewer steps and an improved yield of 50% overall (**Scheme 2.1 B**).



The synthesis of Fmoc- P_R begins with the O-alkylation of commercially available Z-L-hydroxyproline with acrylonitrile to form ether intermediate **1**. Hydrogenation with palladium on

carbon removed the Cbz protecting group of **1**. The resulting amine was Fmoc protected to afford intermediate **2** in quantitative yield (99% for two steps). Reduction of the nitrile of **2** was achieved using hydrogen and platinum oxide, followed by the addition of N, N'-bis-Boc-1-guanylpurazole to yield the final product, **Fmoc-P_R**. The synthesis of hydrophobic CAPH amino acid, **Fmoc-P_L**, proceeded by the two-step literature procedure³⁷.

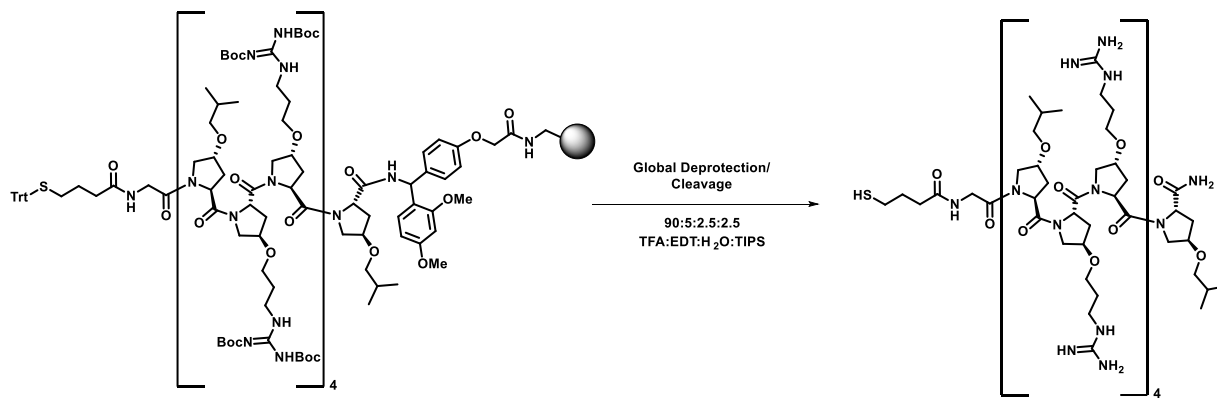
Once both unnatural amino acids were synthesized, Fmoc-based solid phase peptide synthesis (SPPS) was performed to provide the CAPH component of the conjugates. Briefly, **P14LRR-SH** and **FIP14LRR-SH** peptides were synthesized from the C- to N-terminus using a high loading (0.48 mmol/mg) ChemMatrix H-Rink Amide resin. Each amino acid was activated with HATU, coupled to the peptide-bound resin, and the semi-permanent Fmoc group was deprotected with 30% piperidine in DMF. This iterative process was continued until all amino acids were coupled (**Scheme 2.2**).



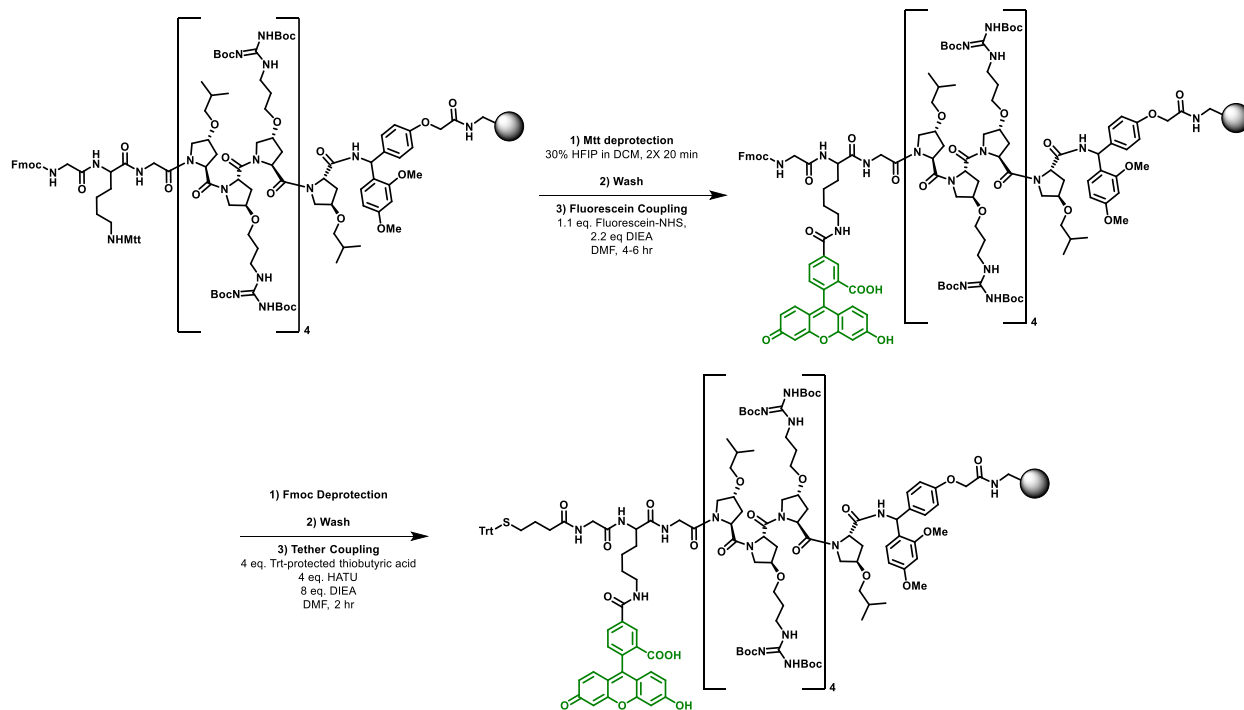
Scheme 2.2. Fmoc-based solid phase peptide synthesis of CAPHs on the Rink amide resin.

After the **P14LRR** peptide was synthesized on resin, the terminal Fmoc was removed and HATU-activated trityl-4-thiobutyric acid was added. Global deprotection and cleavage from the resin was performed using a cocktail of trifluoroacetic acid, ethanedithiol, triisopropylsilane, and water to yield crude peptide (**Scheme 2.3**). **P14LRR-SH** was purified using reverse phase HPLC (RP-HPLC) and the desired product mass was confirmed using matrix-assisted laser desorption ionization – time of flight (MALDI-ToF) mass spectrometry. A fluorescent variant, **FIP14LRR-SH**, was synthesized in order to perform experiments that require fluorescent visualization. Additional steps were taken to add fluorescein once the **P14LRR** peptide was synthesized

(Scheme 2.4). Two additional amino acids, Fmoc-Lys(Mtt)-OH and Fmoc-Gly-OH, were coupled. Then, the Mtt group was selectively deprotected using 30% hexafluoroisopropanol (HFIP) in DCM, followed by the coupling of fluorescein-NHS ester. Addition of trityl-4-thiobutyric acid, followed by global deprotection/cleavage, and purification of peptide by RP-HPLC proceeded similarly to **P14LRR-SH**.



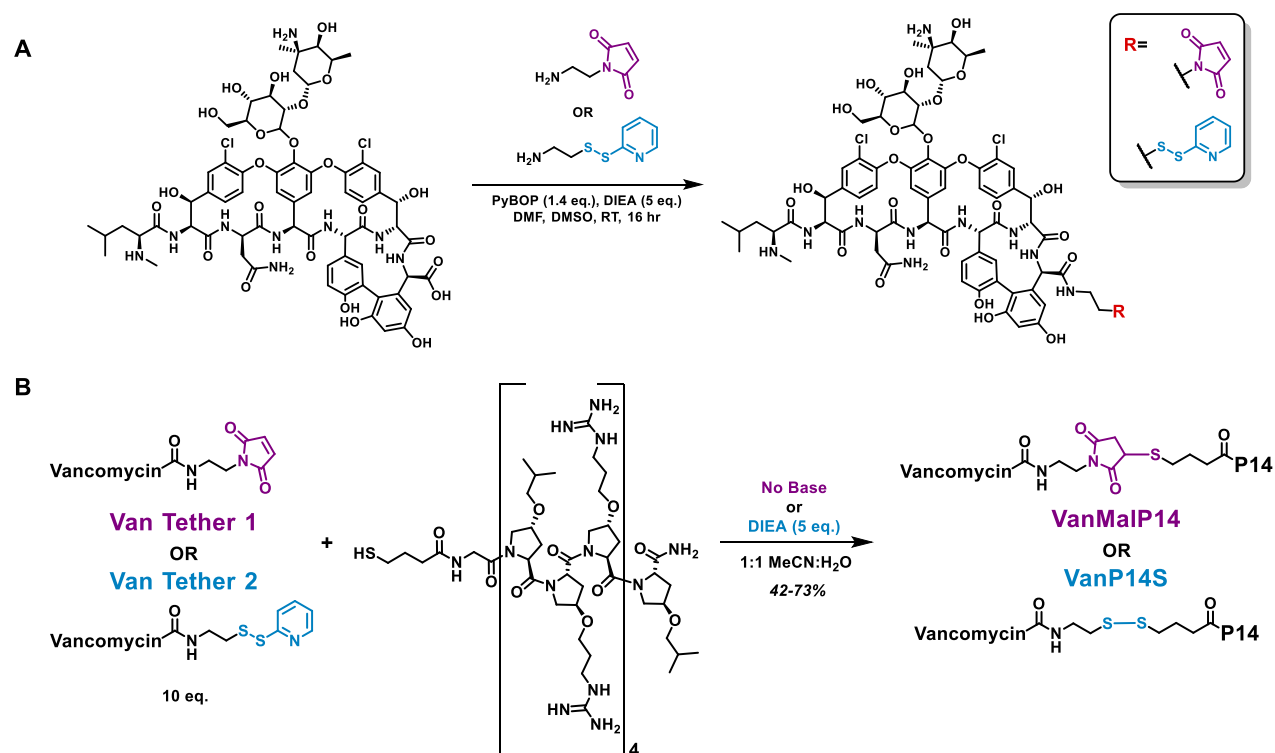
Scheme 2.3. Global deprotection and cleavage from resin with conditions for **P14LRR-SH**.



Scheme 2.4. Addition of fluorescein and thiol tether for synthesis of **FIP14LRR-SH**.

2.2.2 Activating vancomycin and conjugating vancomycin-CAPHs

Vancomycin contains a single carboxylic acid that can be modified without interrupting binding activity to the pentapeptide in the peptidoglycan of Gram-positive bacteria.¹²³ This position was used to covalently link the antibiotic to a CAPH. Two different moieties were chosen to determine whether cleavable or non-cleavable attachment of CAPHs is necessary for more potent activity of the conjugate. A maleimide and an activated disulfide were chosen to link the **P14LRR-SH** peptide to vancomycin to produce the non-cleavable and cleavable conjugates, respectively. Vancomycin hydrochloride was activated for coupling with CAPHs through PyBOP-mediated coupling with amine tethers to generate **VanMalP14** and **VanP14S**, respectively (Scheme 2.5, Figure 2.4 A). A fluorescent conjugate, **FIVanP14S** (Figure 2.4 A), was produced by disulfide exchange in the same manner as **VanP14S**, substituting the CAPH peptide, **FIP14LRR-SH** for **P14LRR-SH**.



Scheme 2.5. (A) Coupling of functionalized amines to vancomycin for the synthesis of Van Tethers. (B) Conjugation of thiol-reactive vancomycin derivatives to **P14LRR-SH**

2.2.3 Synthesis of fluorescent controls

Later on, we will discuss experiments to visualize and explore mammalian cellular uptake and bacterial uptake of the fluorescent conjugate and controls. **FIP14LRR** has been widely used as a control peptide in the Chmielewski group and was synthesized by the same route as previously reported.^{37, 41-42, 44-45} Fluorescent vancomycin controls were synthesized by literature procedure using fluorescein isothiocyanate (FITC) and rhodamine isothiocyanate (RhITC/RITC) (**Figure 2.5**).⁶⁹ Upon completion of the reactions, the desired products were isolated by RP-HPLC. The location of the fluorophore on the secondary amine was determined by MALDI-ToF mass spectrometry by the analysis of the fragmentation pattern (**Figure A 27** and **Figure A 28**). We confirmed by mass spectrometry that the fluorophore did not react with the glucosamine sugar of the glycopeptide.

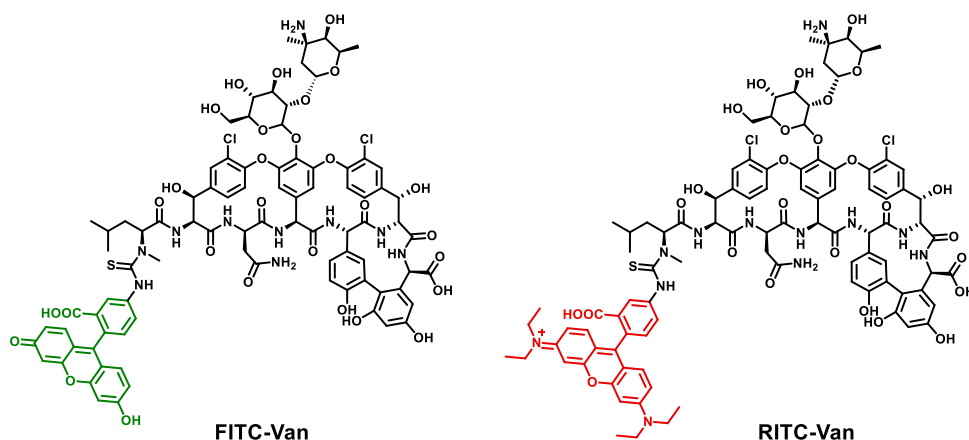


Figure 2.5. Fluorescent vancomycin controls, **FITC-Van** and **RITC-Van**.

2.2.4 Minimum inhibitory concentration against bacterial isolates

First, we wished to determine if these conjugates exhibited superior antibacterial activity compared to controls using a microbroth dilution assay (**Table 2.1**). The tested bacteria included ESKAPE pathogens, Gram-negative and Gram-positive bacteria that are the leading cause of nosocomial and multidrug-resistant infections.¹⁴³ Additionally, a range of susceptible and drug-resistant *S. aureus* were tested. Both conjugates, **VanP14S** and **VanMalP14**, displayed moderate activity against the Gram-negative *E. coli*, *A. baumannii* and *K. pneumonia*. With the same bacteria, Vancomycin, **P14LRR**, and the one-to-one mixture of vancomycin and **P14LRR**, were

not as effective against the same bacteria. This suggests that the conjugation of the two antibiotics appears necessary for bactericidal activity. Both **VanP14S** and **VanMalP14** exhibited similar activity, with a slight improvement of **VanMalP14** over **VanP14S** with *A. baumannii* 19606 and *Klebsiella*.

Table 2.1. Minimum inhibitory concentrations (MICs) of vancomycin conjugates against pathogenic bacteria isolates using the microbroth dilution assay after 20 h of treatment. Values reported in μM .

Bacteria strains (-/+) ^a	Vancomycin	P14LRR	VanP14S	VanMalP14	Van+P14 (1:1)
<i>E. coli</i> ATCC 25922 (-)	>16	16	8	8	16
<i>A. baumannii</i> ATCC 19299 (-)	>16	>16	8	8	>16
<i>A. baumannii</i> ATCC 19606 (-)	>16	16	4	2	16
<i>K. pneumonia</i> ATCC 1706 (-)	>16	>16	4	2	>16
<i>P. aeruginosa</i> ATCC 15442 (-)	>16	>16	16	16	>16
<i>E. cloacae</i> ATCC 1143 (-)	>16	>16	>16	>16	>16
<i>L. monocytogenes</i> ATCC 19111 (+)	0.25	>16	0.5	0.5	0.25
<i>S. aureus</i> ATCC 10537 (+)	0.5	> 16	2	2	1
MRSA USA300 (+)	0.25	>16	1	0.5	0.25
VRSA 10 (+)	>16	>16	1	0.5	>16
VRSA 7 (+)	>16	>16	0.5	0.5	>16
VRSA 5 (+)	>16	>16	1	0.5	>16
VRSA 3A (+)	>16	>16	1	0.5	16
VRSA 1 (+)	>16	>16	0.5	0.5	>16
<i>E. faecium</i> ATCC 700221 (+)	>16	16	0.5	1	8

^a (-/+) denotes Gram-negative or Gram-positive pathogens

Both **VanP14S** and **VanMalP14** exhibit excellent bactericidal activity against Gram-positive bacteria. For *L. monocytogenes*, susceptible *S. aureus* 10537 and MRSA USA300, the conjugates display good activity, however, they display an equivalent potency as vancomycin or a 1 to 1 mixture of vancomycin and **P14LRR**. Astonishingly, **VanP14S** and **VanMalP14** display exceptional activity against a range of vancomycin-resistant *S. aureus* and *Enterococcus* (MIC values of 0.5 to 1 μM), whereas vancomycin and **P14LRR**, or their 1:1 mixture, are not as effective (8, 16, or >16 μM). The activity of conjugates to each other is comparable, with **VanMalP14**

showing slightly better activity in some instances. These data show that both conjugates can overcome resistance that vancomycin typically displays. These data indicate a broad-spectrum activity previously unobserved for vancomycin, and notably improved activity for the broad-spectrum CPP, **P14LRR**.

2.2.5 Time to kill experiments with *S. aureus*

We next investigated the time-to-kill curves of the conjugates against *S. aureus* (**Figure 2.6**). By observing the time to kill bacteria, we can verify whether drugs are bacteriostatic or bactericidal.¹⁴⁴ Rapid killing of bacteria has often been reported for cationic AMPs that disrupt bacterial membrane integrity and function, resulting in bacterial death.¹⁴⁵ Both conjugates, **VanP14S** and **VanMalP14**, exhibited rapid bactericidal killing at 4-6 hours and 2-4 hours as compared to vancomycin at 2X and 5X their MIC values, respectively. At 2X MIC, **VanP14S** kills somewhat faster than **VanMalP14**, while the opposite is observed at 5X MIC. Although slightly different times to eradicate bacteria *in vitro* were observed, the comparable times of the two conjugates may indicate that the reversible and irreversible linkages between vancomycin and CAPH do not play a differing role in antimicrobial action. This increased speed of killing for the conjugates compared to vancomycin may indicate enhanced bacterial membrane disruption, common to AMPs.

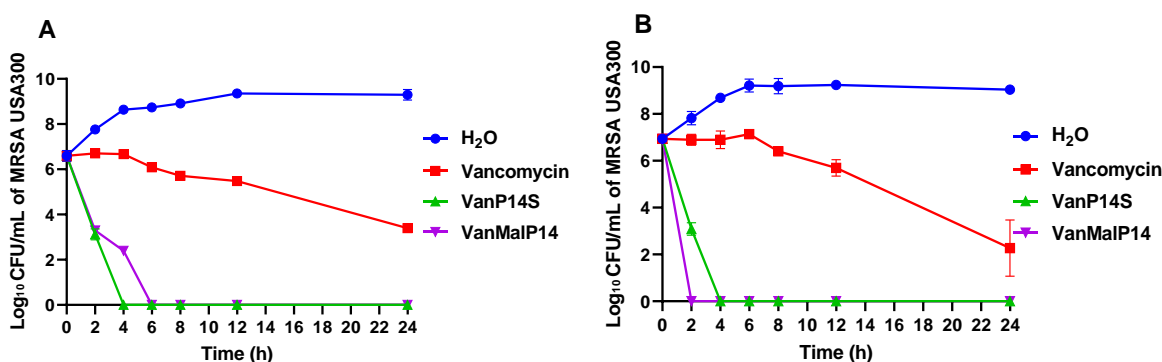


Figure 2.6. Killing kinetics of vancomycin-P14LRR conjugates in comparison to vancomycin at (A) 2X and (B) 5X their MIC values against methicillin-resistant *Staphylococcus aureus* NRS 384 (MRSA USA 300) over a 24-hour incubation period. Water served as a negative control and vancomycin served as a control drug. Error bars represent standard deviation values obtained from triplicate samples used for each test agent.

2.2.6 Cell toxicity studies

Having established excellent antibacterial activity, we next wanted to probe toxicity toward human red blood cells (hRBC) and mammalian macrophage cells (J774A.1). Although the CPP **P14LRR** peptide has reported antimicrobial activity, it does not demonstrate a lytic mechanism of action like many antimicrobial peptides, such as melittin.¹⁴⁶ In this experiment, conjugates were incubated for one hour with hRBCs. If the RBCs were lysed by the compounds, heme would be released and detected in the supernatant by UV-Vis spectroscopy. After one hour of incubation with hRBCs, we confirmed that neither conjugate displayed red blood cell hemolysis at concentrations up to 40 μ M, consistent with the individual non-lytic behaviors of vancomycin and **P14LRR** (Figure 2.7 A).

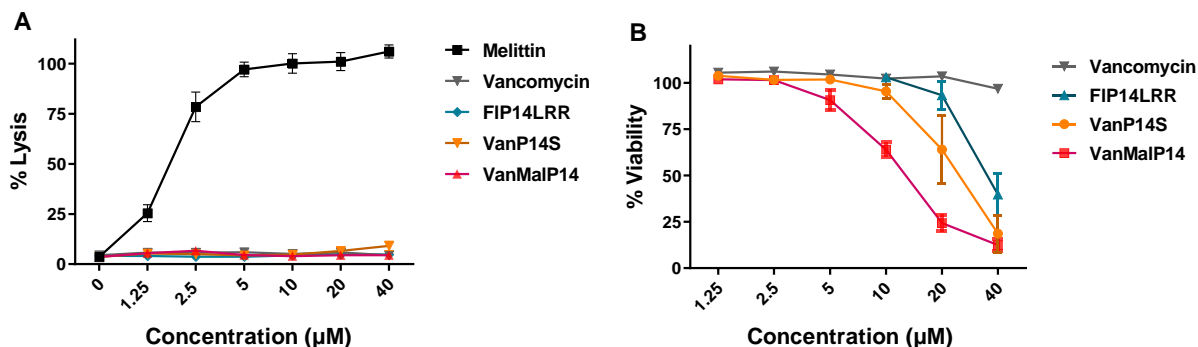


Figure 2.7. (A) Hemolysis assay measuring the release of hemoglobin from hRBC. Melittin was used as a positive control, and values normalized against 0.1% triton X-100. (B) Cytotoxicity measuring J774A.1 cell viability using an MTT assay after 9-hour incubation with compounds.

Because macrophage cells are capable of harboring bacteria in a refuge from external antibiotics, we wish to carry out internalization experiments with a J774A.1 macrophage cell line. Therefore, we evaluated the viability of mammalian cells in the presence of vancomycin-P14LRR conjugates at concentrations ranging from 1.25 μ M to 40 μ M for 9 hours (Figure 2.7 B).¹⁴⁷ The results show increased cytotoxicity of both conjugates compared to vancomycin and P14LRR, but the decreased viability starts at 10 μ M and 20 μ M for **VanMalP14** and **VanP14S**, respectively. Although there is a decrease in viability at 10 μ M and 20 μ M, these concentrations are greater than most Gram-negative bacterial MICs and about 10-fold higher than the selected *S. aureus* and

Enterococcus MICs. Because the cytotoxic concentrations of conjugates remain higher than MICs, toxicity should not be an obstacle in the implementation of these conjugates.

2.2.7 Confocal microscopy to visualize cellular internalization of FIVanP14S conjugate

Although mostly understood as an extracellular pathogen, *Staphylococcus aureus* is a pathogen that has long been known to evade antibiotics through internalization within macrophages and neutrophils.¹⁴⁸⁻¹⁵¹ Once engulfed, professional phagocytes like macrophages can harbor intracellular *S. aureus* within intracellular compartments such as phagosomes and phagolysosomes, where bacteria can replicate, kill host cells, and infect other cells in a continuous infection cycle. Residence and consequential persistence within host cells enables long-term survival. Thus, relapses in illness due to chronic infection result from the clinical failure in eradicating these intracellular reservoirs.¹⁰⁹ In order to effectively clear these intracellular infections, it is ideal for antibiotics to localize within the same intracellular compartments to successfully kill bacteria.

Based on the previous results that **FIP14LRR** was found to penetrate macrophage cells, we wished to study the cellular uptake of the vancomycin-CAPH conjugates. Therefore, the fluorescent vancomycin-P14LRR conjugate, **FIVanP14S**, was incubated at a concentration of 5 μ M with J774A.1 macrophage cells for 1 hour. Following incubation, we further labeled cells with an endosomal marker (Lysotracker red, 300 nM) or a mitochondrial marker (Mitotracker red, 100 nM). Faint green fluorescence is observed in the cytosol; however, most of the green fluorescence appears strong and punctate. When incubated with the endosomal marker, strong overlap of the green **FIVanP14S** and red marker appears as a yellow-orange color (**Figure 2.8 A**). Whereas incubation with the red mitochondrial marker, no overlap is seen and both green and red fluorescence is partitioned (**Figure 2.8 B**). After 1 hour incubation with 10 μ M **FIVanP14S**, the same results are observed; there is strong colocalization with the endosomes, and no colocalization with the mitochondria (data not shown). It can be concluded that **FIVanP14S** primarily colocalizes with endosomal vacuoles at this concentration and incubation time with some cytosolic localization. The results of this study provide evidence that this vancomycin-CPP conjugate can appropriately localize within similar vacuoles as observed for *S. aureus*.

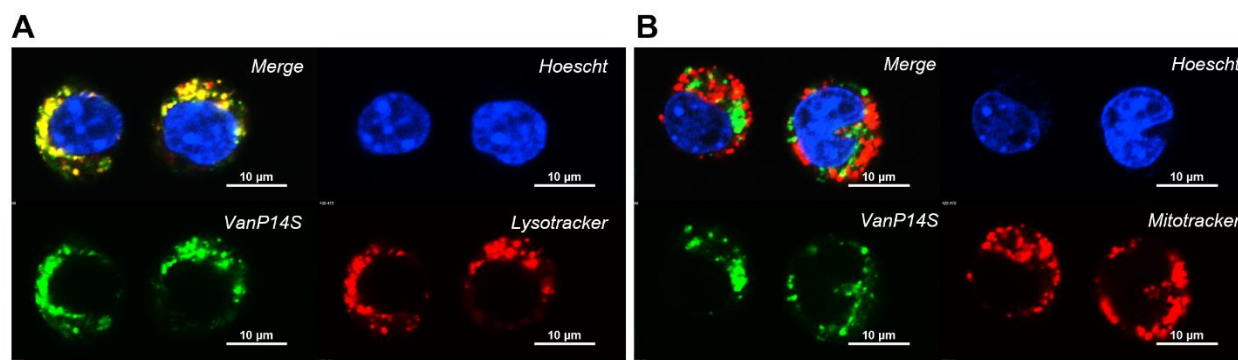


Figure 2.8. Confocal microscopy images of J774A.1 macrophage cells incubated with 5 μ M FIVanP14S for 1 hr. (A) Merge (top left) and separate laser channels with Hoescht 33342 nuclear stain (blue), conjugate (green), and lysotracker lysosomal stain (red). (B) Merge (top left) and separate laser channels with Hoescht 33342 stain (blue), conjugate (green), and mitotracker mitochondrial (red). A yellow-orange color indicates colocalization of conjugate and selected red stain.

2.2.8 Flow cytometry to quantify mammalian cellular accumulation of FIVanP14S

With qualitative knowledge of conjugate location within macrophage cells, we next wanted to quantify the amount of conjugate entering mammalian cells in comparison to the poorly internalized vancomycin control, **FITC-Van**. Insufficient accumulation of therapeutic concentrations of drugs is one of the major obstacles in treating intracellular pathogens. A potential therapy for intracellular bacterial infection must efficiently enter cells and maintain an effective concentration to kill intracellular pathogens. Therefore, we monitored the cell accumulation of **FIVanP14S** in J774A.1 macrophage cells using flow cytometry. Cells were treated with a range of compound concentrations for both 1-hour and 3-hour incubation times. **FIP14LRR**, a previously reported CAPH with better cell penetration than the well-known Tat-peptide, was used as a control.⁴¹ In this experiment, all compounds entered cells, even **FITC-Van** (**Figure 2.9**). **FIP14LRR** showed at least three-fold higher accumulation than **FITC-Van** at all concentrations at both incubation times. Encouragingly at 1-hour incubation, **FIVanP14S** showed about a 6- to 10-fold increase as compared to **FITC-Van** uptake at all concentrations; whereas at a 3-hour incubation, **FIVanP14S** showed about 16- to 19-fold increases in fluorescence as compared to **FITC-Van**. These results demonstrated that **FIVanP14S** had about a 1- to 2-fold increase in uptake from **FIP14LRR** after 1-hour incubation and about a 4-fold increase at 3-hour incubation. Although unexpected, the conjugation of vancomycin to the CAPH CPP led to the increased uptake of **FIVanP14S** compared to **FIP14LRR**. Thomas Dietsche and Chmielewski coworkers have

shown N-terminal modification to the **P14LRR** peptide results in a change of cellular uptake of peptides;⁴⁴ therefore, the addition of vancomycin to the N-terminus of **P14LRR** could explain this change in cellular accumulation. Moreover, the additional cationic groups on vancomycin may also lead to an increase in cellular uptake of the conjugate compared to **P14LRR**. For instance, the extension CAPHs from **P14LRR** to **P17LRR** resulted in an increase in cellular uptake presumed to be the result of the increase in cationic character.^{35, 43}

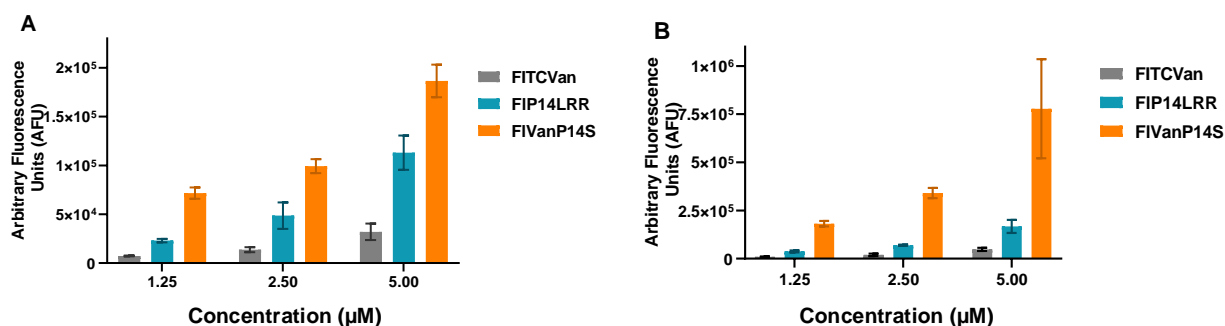


Figure 2.9. Screening the fluorescence of J774A.1 cells incubated with compounds at (A) 1-hour and (B) 3-hour incubation times.

To determine if the conjugates were fully internalized within the cells, or bound to the membrane, we used trypan blue to quench the fluorescence of surface-bound conjugates.¹⁵²⁻¹⁵³ The amount of surface-bound **FIVanP14S** versus internalized **FIVanP14S** was measured by comparing the arbitrary fluorescence units (AFU) of cellular fluorescence quenched by trypan blue to the AFU of cells not quenched with trypan blue after 1- and 3-hour incubations using flow cytometry. After 1-hour incubation with **FIVanP14S**, surface binding accounted for 17% of the total fluorescence at 2.5 μM concentration and less than 3% of the total fluorescence at 5 μM concentration (**Figure 2.10 A, Figure 2.10 B**). After 3 hours of incubation with **FIVanP14S**, increased surface binding was seen, yet the majority of **FIVanP14S** was internalized. A 34% and 38% decrease in fluorescence was seen at 2.5 and 5 μM concentration, respectively (**Figure 2.10 C, Figure 2.10 D**). In all situations, **FIVanP14S** shows less than 40% surface binding as 60% of the conjugate is internalized. Accordingly, **FIVanP14S** translocation into the cell is time-dependent as the fluorescence with trypan blue more than doubles from 1 to 3 hours. More peptide accumulating into the cells, rather than remaining surface-bound, is beneficial for reaching and achieving intracellular

pathogen clearance. Confocal microscopy images confirm similar cell internalization and minimal surface-bound conjugate at 1 hour (**Figure 2.8**).

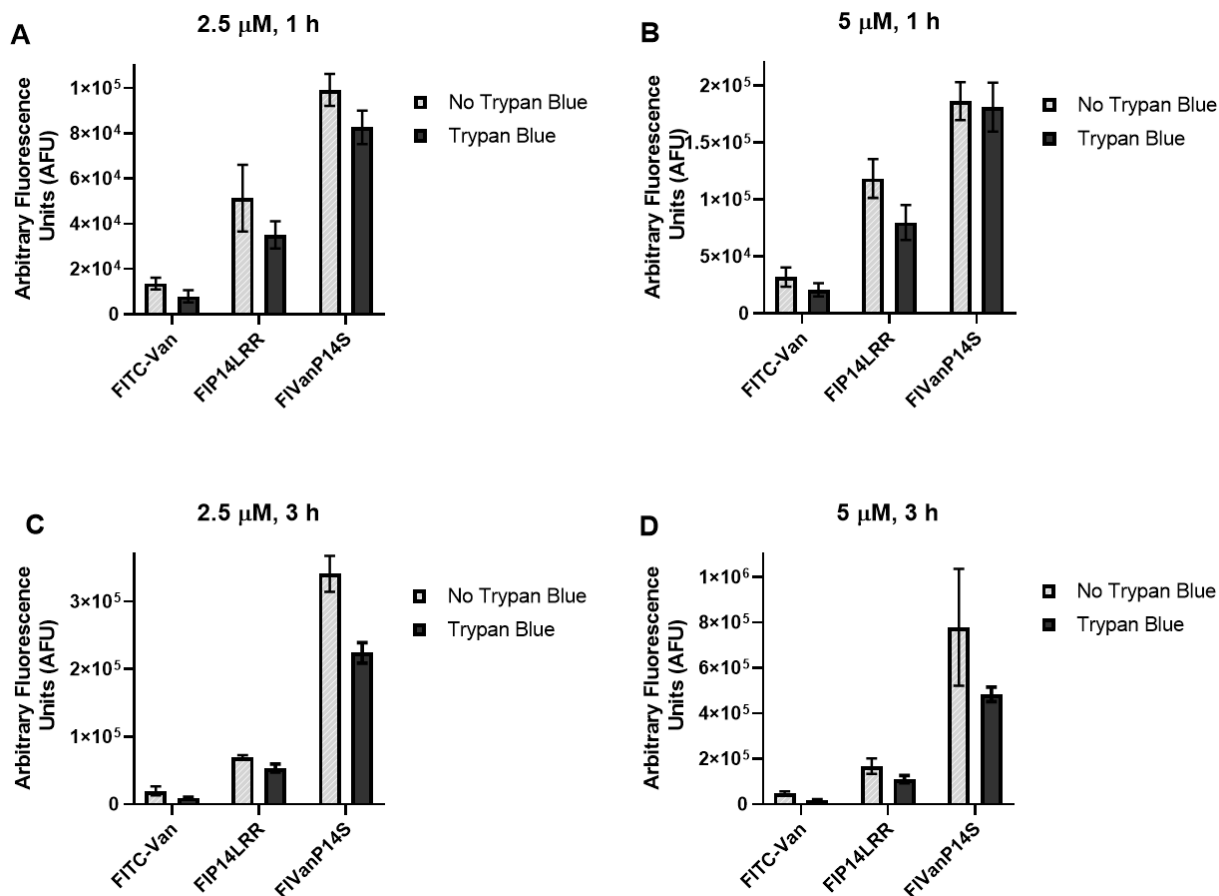


Figure 2.10. Cellular uptake studies of **FIVanP14S**, **FIP14LRR**, and **FITC-Van** at (A, B) 1 h and (C, D) 3 h in J774A.1 cells after incubation at (A, C) 2.5 μ M and (B, D) 5 μ M with trypan blue treatment (black bar) and without trypan blue treatment (gray bar).

2.2.9 Release kinetics of reduced Vancomycin-SH from VanP14S conjugate

VanP14S was designed with a reducible disulfide linkage to release thiol-modified vancomycin and **P14LRR-SH**, two individual antibiotics (**Figure 2.11 A**). Because vancomycin was modified from the C-terminal carboxylic acid, the reduced **Van-SH** is expected to retain equivalent activity as vancomycin. The carboxylic acid on vancomycin does not directly interact with the peptidoglycan target, therefore, the small modification should not detrimentally impact binding.^{123, 154} To evaluate the half-life of the conjugate, **VanP14S** was treated with 10 mM

dithiothreitol (DTT) in PBS (pH 7.4) (**Figure 2.11 B**). **Van-SH** release was monitored using UPLC-MS and the half-life of release ($t_{1/2}$) was calculated to be about 1 minute for a single trial. This rapid reduction prompted us to move to more appropriate physiological conditions. Therefore, **VanP14S** was treated with 10 mM glutathione (GSH) in PBS (pH 7.4) (**Figure 2.11 C**).⁷⁰ **Van-SH** release was monitored using UPLC/MS and the half-life of release ($t_{1/2}$) was calculated to be 150 minutes (2.5 hours) for the experiment performed in duplicate. These studies indicate **VanP14S** is responsive to reducing environments similar to those *in cyto*.

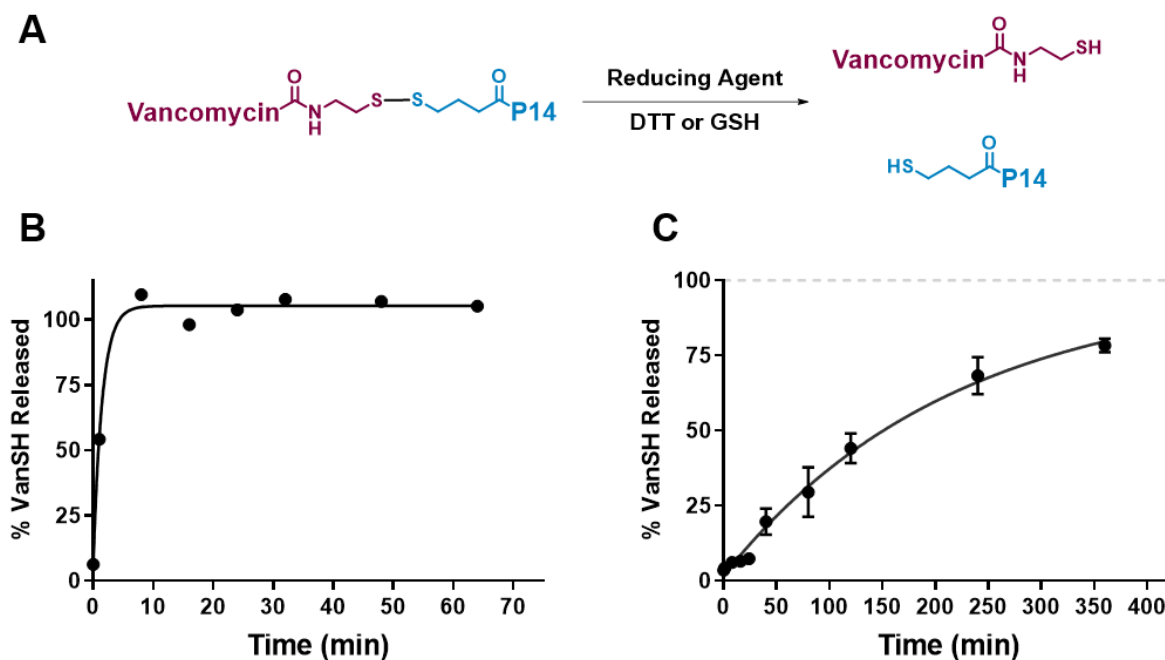


Figure 2.11. (A) Scheme of disulfide reduction and release of vancomycin-SH from VanP14S. (B) Reduction and release of vancomycin-SH using 10 mM DTT in PBS. (C) Reduction and release of vancomycin-SH using 10 mM GSH in PBS.

2.2.10 Mechanism of action studies

Both conjugates, **VanP14S** and **VanMalP14**, displayed similarly potent antibacterial activity. Therefore, we next wanted to probe their mode(s) of action. The mechanism of action common to glycopeptides is the inhibition of bacterial cell-wall synthesis through binding peptidoglycan precursors and hindering the crosslinking of the cell wall to result in cell death. Approved glycopeptides dalbavancin, ortavancin, and telavancin also possess additional interactions to disrupt membrane potential and membrane permeability.¹¹⁸ We first want to explore

potential membrane interactions that these new conjugates may use in their mechanism to kill broad-spectrum and resistant pathogens.

Gram-negative membrane leakage

We first explored whether the inner membrane of Gram-negative bacteria was permeabilized by either **VanP14S** or **VanMalP14** conjugate through a β -galactosidase leakage experiment. Solutions of conjugates at concentrations 2X and 4X their MIC values were used for incubation with *E. coli* ATCC 25922. Melittin, a known membrane-lytic AMP, was used as a positive control. Bacteria were induced using isopropyl- β -D thiogalactopyranoside (IPTG), then incubated with conjugates at various concentrations for 1 hour. Next, a β -galactosidase substrate, 2-nitrophenyl- β -D-galactopyranoside (ONPG), was added. The β -galactosidase enzyme, which is released into the supernatant at concentrations relative to the amount of membrane permeability, cleaves ONPG and releases the colorimetric indicator, O-nitrophenol. The concentration of the indicator can be measured by UV-Vis spectroscopy and is proportional to the amount of membrane permeability (**Figure 2.12**). The absorbance of the indicator was measured over a period of 1 hour. At both concentrations, no significant absorbance was measured for either conjugate compared to negative control. These results indicate that no significant inner membrane leakage was the result of either conjugate at the tested concentrations.

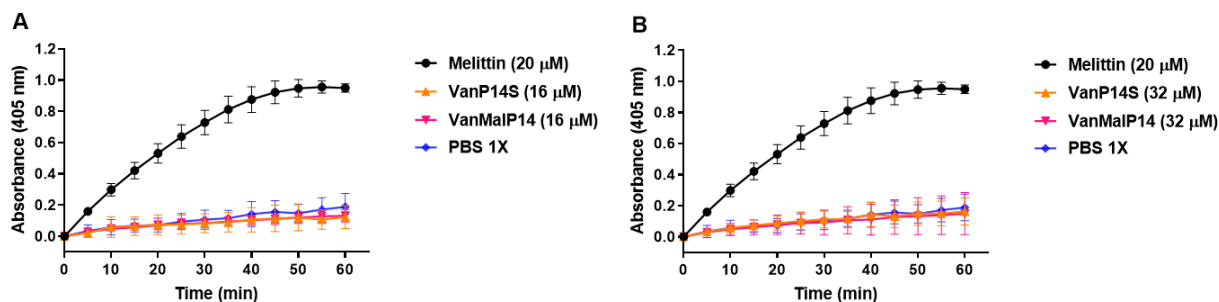


Figure 2.12. Monitoring the release of β -galactosidase over 1 hour from *E. coli* ATCC 25922 after 1 hour incubation with conjugates. Positive control Melittin (20 μ M) and conjugates at (A) 2X and (B) 4X MIC.

Gram-negative outer membrane permeability

Looking into the outer membrane of Gram-negative bacteria, a membrane disruption assay using the fluorescent dye, N-phenyl-1-naphthylamine (NPN) was employed.¹⁵⁵ In the presence of membrane disruption, NPN binds within the lipophilic environment of the membrane and produces a fluorescent signal relative to the amount of membrane disruption. Both **VanP14S** and **VanMalP14** conjugates and **P14LRR** were studied, along with a bacterenecin control, a known cyclic AMP with known outer membrane permeation as a positive control (**Figure 2.13**).¹⁵⁶⁻¹⁵⁷ All values were normalized with 1% Triton X-100, which was assumed to cause total disruption to the membrane. Both conjugates showed equivalent outer membrane disruption in a concentration-dependent manner when incubated at values within MIC range. **P14LRR** also displayed analogous concentration-dependent outer membrane disruption to the conjugates, whereas membrane permeation attributable to vancomycin was minimal. Given that **P14LRR**, **VanP14S**, and **VanMalP14** do not cause inner membrane leakage as discussed previously, the bactericidal activity may not be solely the result of the outer membrane permeation displayed by these compounds. These results may indicate that the conjugates and **P14LRR** cause membrane permeation for access and entry to the peptidoglycan of Gram-negative bacteria.¹⁵⁸⁻¹⁵⁹

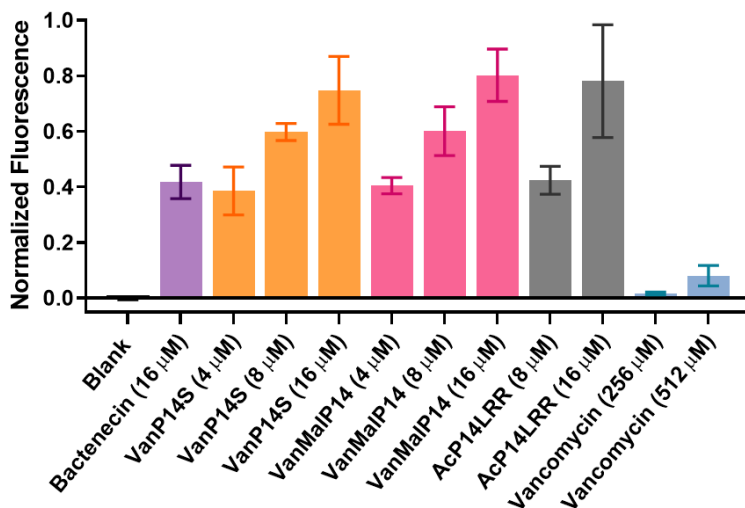


Figure 2.13. Outer membrane permeability assay using NPN as fluorescent probe to monitor the disruption of the outer membrane of Gram-negative bacteria, *E. coli* ATCC 25922.

Gram-positive membrane leakage

Next, we wanted to investigate the effects that the conjugates had on the membrane of Gram-positive bacteria such as *S. aureus*. **VanP14S** and **VanMalP14** were incubated with MRSA USA300 for 30 minutes (**Figure 2.14**). Water and vancomycin were used as negative controls; and lysostaphin was used as a positive control because of its known induction of membrane permeability through the degradation of the bacterial cell wall.¹⁶⁰ After the incubation period, a UV-Vis NanoDrop spectrometer was used to quantify the concentration of released DNA (ng/ μ L) by analysis of the 260 and 280 nm absorbing components of nucleic acids.¹⁶¹ The concentration of released DNA is proportional to the extent of membrane damage. Both conjugates equally display 20% membrane leakage compared to the lysostaphin control, with the negative control value subtracted. Although the conjugate concentrations used were at 5X their MIC values, both conjugates show analogous results that correspond to a small, yet significant, amount of membrane damage. Like other membrane-insertive lipidated-glycopeptide analogs, it may be that the amphiphilic CAPH component of the conjugates is interacting with the membrane and causing disruption at high concentrations. This could be a secondary interaction in addition to the known peptidoglycan binding of vancomycin, leading to enhanced bacteria death.

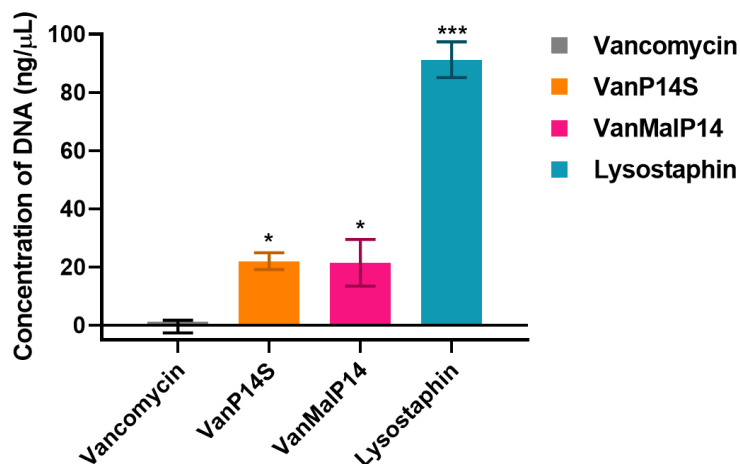


Figure 2.14. Leakage of nucleic acids resulting from vancomycin-P14 conjugate (5X MIC) incubation with MRSA USA300. Data were analyzed by one-way ANOVA. H₂O was negative control and Lysostaphin (20 μ g/mL) was positive control. Asterisks (*/**/****) correspond to adjusted P-values ($\alpha = 0.05$) of < 0.033/< 0.002/< 0.001.

Membrane depolarization

The membrane potential of the cytoplasmic membrane is crucial to bacterial survival. Dissipation of the transmembrane potential resulting from pore formation would lead to an electrical failure of a bacterium's primary defense. Dissipation could also result from other methods of increasing ion permeability.¹⁶² To examine whether **VanP14S** or **VanMalP14** can disrupt the bacterial membrane potential of the cytoplasmic membrane, we used the fluorescent voltage-sensitive probe 3,3'-dipropylthiadicarbocyanine iodide (diSC₃₋₅). This dye is self-quenching upon intercalating within the bacterial membrane. Upon dissipation of the membrane potential, the dye is released and produces a fluorescent signal proportional to the amount of membrane potential dissipation. In this experiment, bacteria were incubated with diSC₃₋₅ for 1 hour. Then, concentrations of **VanP14S**, **VanMalP14**, **P14LRR**, and vancomycin were added to the bacteria and the fluorescence of the diSC₃₋₅ release was obtained on a spectrometer. Triton X-100 (1%) served as a positive control as it digests bacterial membranes (**Figure 2.15**).

Among the tested compounds, **VanMalP14**, **P14LRR** and vancomycin, did not produce significant fluorescence signals from depolarization of the cytoplasmic membrane in the Gram-negative *E. coli* (**Figure 2.15 A**). However, a low level of dissipation of the membrane potential was observed for **VanP14S**, with about 5% dissipation compared to the Triton X positive control at all three concentrations. For Gram-positive *S. aureus*, both conjugates, **VanP14S** and **VanMalP14**, exhibited a concentration-dependent dissipation of the membrane potential (**Figure 2.15 B**). **VanP14S** showed 9%, 17%, and 18% dissipation at 4, 8, and 16 μ M, respectively. **VanMalP14** showed 12%, 18%, and 20% dissipation at 4, 8, and 16 μ M, respectively. Upon analysis of these results, the conjugates show minimal dissipation of membrane potential in the Gram-negative *E. coli* at concentrations near the MIC values. Moreover, **VanP14S** and **VanMalP14** both show low, but significant levels of membrane dissipation potential at concentrations near reported MIC values in the Gram-positive *S. aureus*. These data validate that disrupting membrane potential, especially in Gram-positive bacteria, may be an aspect in the mode of killing from these conjugates.

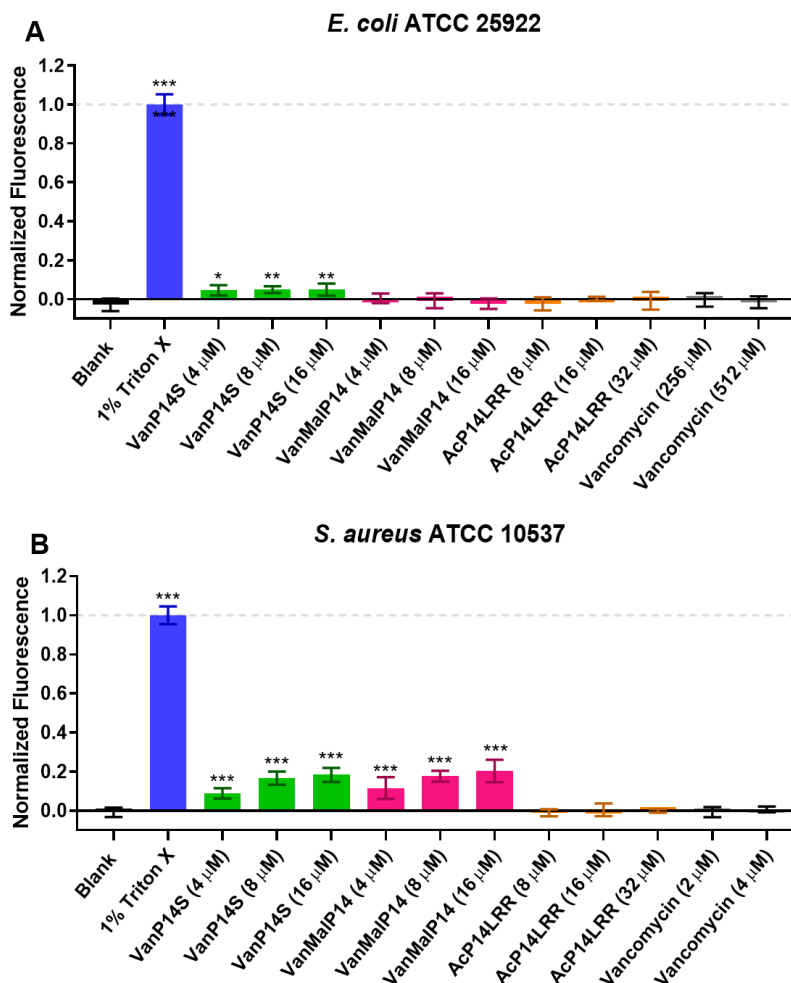


Figure 2.15. Cytoplasmic membrane potential disruption assay using disc₃-5 as fluorescent probe to monitor the quantity of potential dissipation in (A) Gram-negative *E. coli* and (B) Gram-positive *S. aureus* bacteria. Data were analyzed by one-way ANOVA. Asterisks (*/**/****) correspond to adjusted P-values ($\alpha = 0.05$) of < 0.033 / < 0.002 / < 0.001 .

2.2.11 Confocal microscopy to visualize bacterial internalization of compounds

In order to evaluate the bacterial localization of the **FIVanP14S** conjugate and controls, **FITC-Van** and **FIP14LRR**, we utilized confocal imaging. Previously, fluorescent vancomycin has been shown to localize in the cell wall of non-replicating Gram-positive bacteria, and in the septa during cell division, owing to vancomycin binding to lipid II of the peptidoglycan layer of the cell wall.^{69, 163}

First, we investigated the fluorescent uptake in Gram-negative *E. coli* ATCC 25922 after 10 minute and 1 hour incubation times with 20 μ M **FITC-Van**, **FIP14LRR**, and **FIVanP14S**.

Only minimal fluorescence was observed for **FITC-Van** with *E. coli* even up to 20 μ M, consistent with vancomycin's inability to penetrate through the outer membrane (data not shown). However, incubation of *E. coli* with 20 μ M **FIP14LRR** showed rapid internalization to cytoplasm as previously described (**Figure 2.16 A, B**).⁴⁵ At the same concentration and times, **FIVanP14S** shows much less consistent results. After 10 minutes of incubation, some bacteria show minimal uptake and spotty uptake around the cell wall (**Figure 2.16 C**). After 1 hour of incubation, more internalization is visible yet most bacteria with fluorescence exhibit inconsistent and spotty uptake (**Figure 2.16 D**). These results are reasonable, with respect to the mechanism of action studies whereby **VanP14S** showed outer membrane permeability without displaying inner membrane perturbation. Perhaps the conjugate both binds peptidoglycan and also slowly translocates to the bacterial cytoplasm through both membranes due to sufficient permeability of the CAPH.

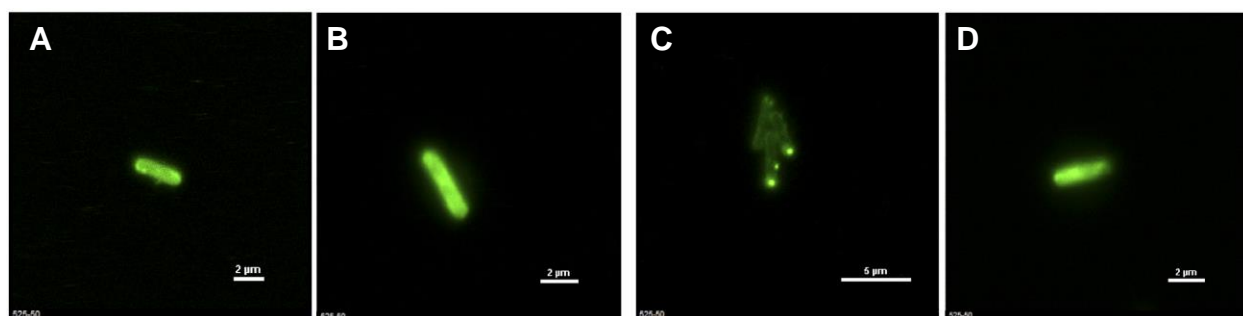


Figure 2.16. Confocal microscopy imaging of *E. coli* ATCC 25922 with 20 μ M concentration of (A, B) **FIP14LRR** after (A) 10 minutes and (B) 1 hour incubation, and (C, D) **FIVanP14S** after (C) 10 minutes and (D) 1 hour incubation.

We next investigated the fluorescent uptake in Gram-positive *S. aureus* ATCC 10537 after 10 minute and 1 hour incubation times with 5 μ M **FITC-Van**, **FIP14LRR**, and **FIVanP14S** (data not shown). All three compounds displayed strong cell wall fluorescence and minimal internal fluorescence after both incubation times. Interestingly, the **FIP14LRR** cell-penetrating peptide does not visually internalize within *S. aureus* to the same extent as within *E. coli*. **FITC-Van** was consistent with the literature, where vancomycin targets and localizes in the cell wall in Gram-positive bacteria.¹⁶³ Next, we wanted to investigate any visual differences in bacterial uptake location between the compounds. Therefore, we incubated a red fluorescent vancomycin, **RITC-Van**, with either green fluorescent **FIP14LRR** or **FIVanP14S** for 10 minutes. After incubation of 5 μ M **RITC-Van** with **FIP14LRR**, we observe both compounds localize around the cell wall of

S. aureus ATCC 10537 after 10 minutes (**Figure 2.17**). The colocalization of the green **FIP14LRR** and red **RITC-Van** is expressed by a yellow-orange color in the merge of both channels (**Figure 2.17 A**). Performing the same experiment with 5 μ M **RITC-Van** and 5 μ M **FIVanP14S**, yielded analogous results (**Figure 2.18**). We observe both compounds localize around the cell wall of *S. aureus* ATCC 10537 after 10 minutes. The colocalization of the green **FIVanP14S** and red **RITC-Van** is expressed by a yellow-orange color in the merge of both channels (**Figure 2.18 A**). **FIVanP14S**, analogous to vancomycin, was observed to localize within the cell wall and the septa of replicating cells. These data provide insight that the **FIVanP14S** conjugate may target the peptidoglycan and/or cell membrane in its mode of action in killing Gram-positive bacteria.

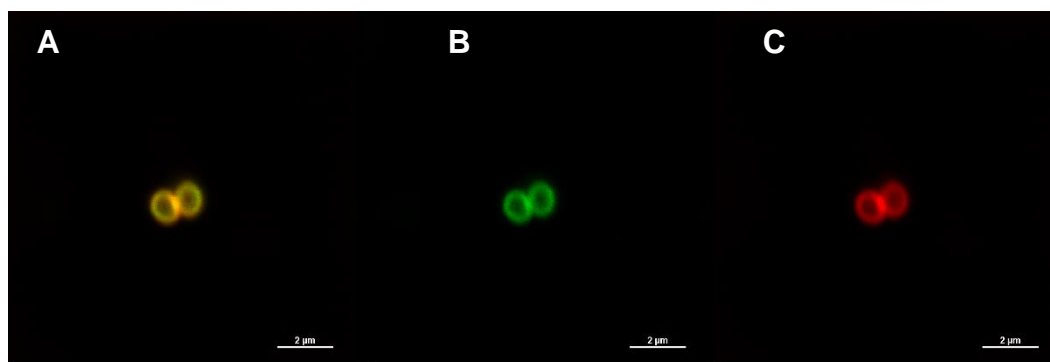


Figure 2.17. Confocal microscopy imaging of *S. aureus* ATCC 10537 with 5 μ M **RITC-Van** and 5 μ M **FIP14LRR** after 10 minutes of incubation. (A) Overlay of green and red channels, yellow-orange color represents colocalization. (B) Green channel representing **FIP14LRR** fluorescence. (C) Red channel representing **RITC-Van** fluorescence.

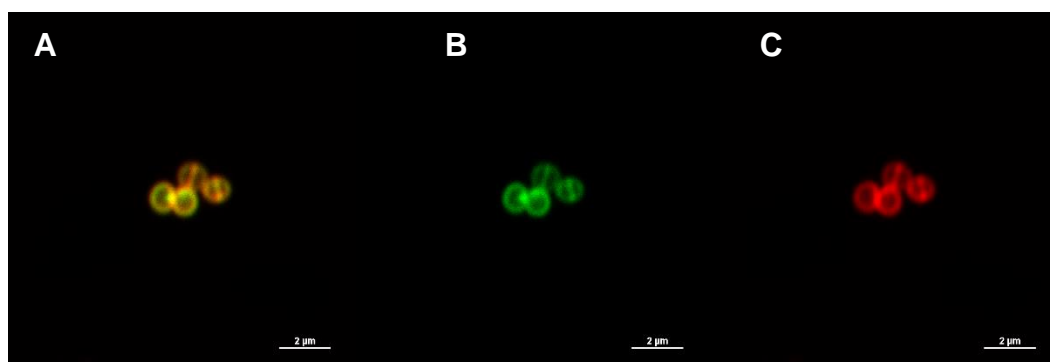


Figure 2.18. Confocal microscopy imaging of *S. aureus* ATCC 10537 with 5 μ M **RITC-Van** and 5 μ M **FIVanP14S** after 10 minutes of incubation. (A) Overlay of green and red channels, yellow-orange color represents colocalization. (B) Green channel representing **FIVanP14S** fluorescence. (C) Red channel representing **RITC-Van** fluorescence.

2.2.12 Flow cytometry to quantify bacterial accumulation of compounds

To thoroughly examine the spatial localization of **FIVanP14S** with both *E. coli* and *S. aureus*, we used flow cytometry with and without trypan blue (TB) to quench the surface-bound fluorophores.^{69, 163-165} If bacterial membranes remain intact, the membrane impermeable TB will quench only surface-bound fluorescence from **FITC-Van**, **FIP14LRR** or **FIVanP14S**. Therefore, we can determine the extent of internalized fluorescence. The amount of surface-bound **FIVanP14S** versus internalized **FIVanP14S** was measured by comparing the arbitrary fluorescence units (AFU) of cellular fluorescence quenched by trypan blue to the AFU of cells not quenched with trypan blue after 10 minute and 1-hour incubations using flow cytometry (**Figure 2.19**). In *E. coli*, 88% and 94% of the external fluorescence was quenched by TB for **FIVanP14S** at 10 minute and 1-hour times, respectively. The fluorescence intensities for **FIP14LRR** and **FIVanP14S** were comparable to each other, and much stronger than **FITC-Van** for both incubation times; consistent with **FITC-Van** confocal imaging and poor penetration into *E. coli*. Longer incubation time increased intensity of fluorescence for **FIP14LRR** and **FIVanP14S**, demonstrating a time-dependent uptake in *E. coli*.

In *S. aureus*, we observed a 95% and 97% decrease in fluorescence quenching by TB for 10 minute and 1 hour incubation times with **FIVanP14S**, respectively. **FIP14LRR** also exhibited an 86% and 92% decrease in fluorescence with TB quenching after 10 minutes and 1-hour incubations. This is consistent with confocal imaging showing strong fluorescence in the cell wall for both **FIP14LRR** and **FIVanP14S**. Additionally, **FIVanP14S** showed 4-fold higher increase in uptake than **FIP14LRR** at both 10 minute and 1-hour incubation times without trypan blue. **FIP14LRR** showed 12- and 8-fold higher increases in uptake than **FITC-Van** at 10 minute and 1-hour incubation times without trypan blue. These results are consistent with a proposed mechanism of enhanced association of the **FIVanP14S** conjugate with the lipid II of the peptidoglycan and/or membrane.

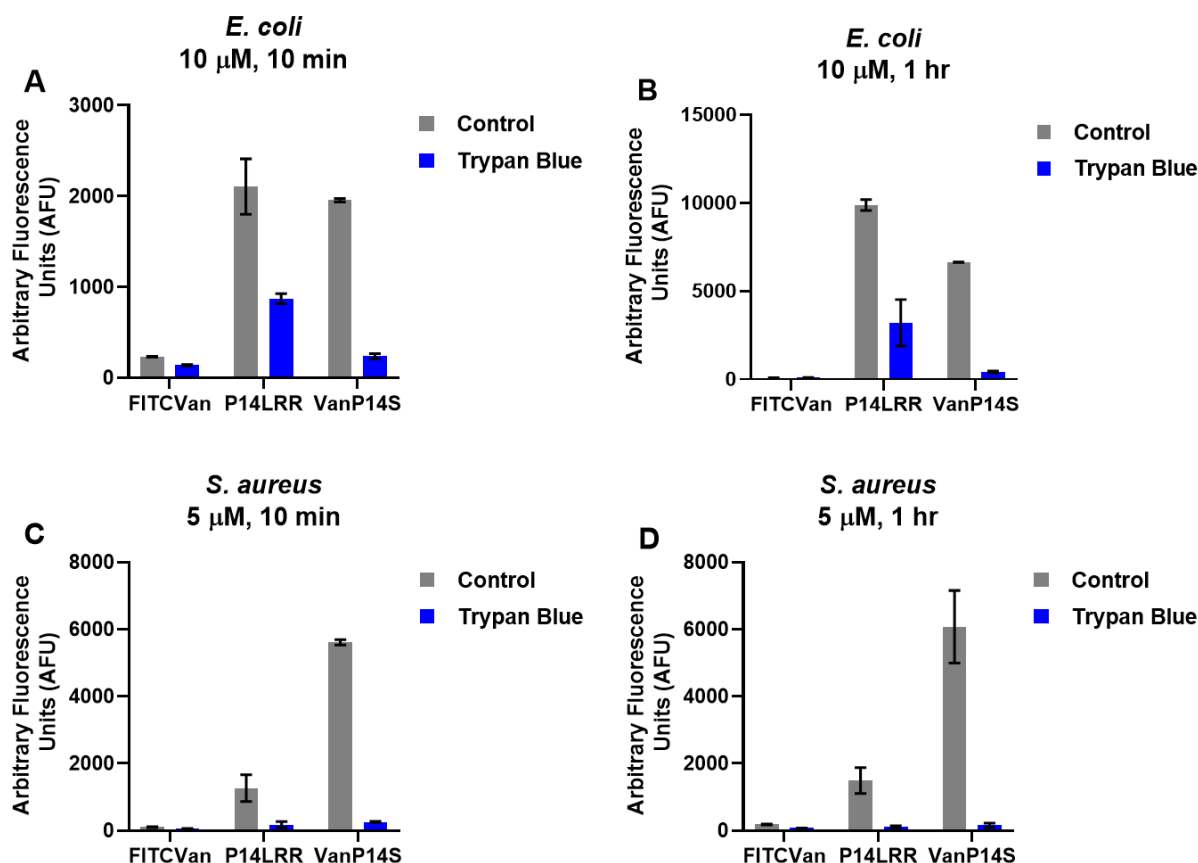


Figure 2.19. Bacterial uptake studies of **FITC-VanP14S**, **FIP14LRR**, and **FITC-Van** in (A, B) 10 μ M concentrations with *E. coli* and (C, D) 5 μ M concentrations with *S. aureus* after (A, C) 10 minute and (B, D) 1 hour incubation periods. 2 mg/mL trypan blue treatment (blue bar) and without trypan blue treatment (gray bar).

2.2.13 Vancomycin-P14GAP conjugates

From previous research in the Chmielewski group, a modified CAPH had shown improved antibacterial activity and cellular uptake in comparison to the **P14LRR** peptide. This cell-penetrating peptide, **P14GAP**, consists of the same polyproline backbone with modification to the cationic amino acid.⁴² The cationic guanidinium moiety on the modified amino acid, Fmoc-P_{GAP}, is directly connected to the proline, allowing the charged groups to be held closer to the backbone, than with the Fmoc-P_R of **P14LRR**. We sought to exploit the increased antibacterial activity of **P14GAP** and generate a vancomycin-P14GAP conjugate with improved features compared to the vancomycin-P14LRR conjugate. We used an identical strategy to synthesize **VanP14GAPS** and **FIVanP14GAPS** as that for **VanP14S** and **FIVanP14S** (Figure 2.20).

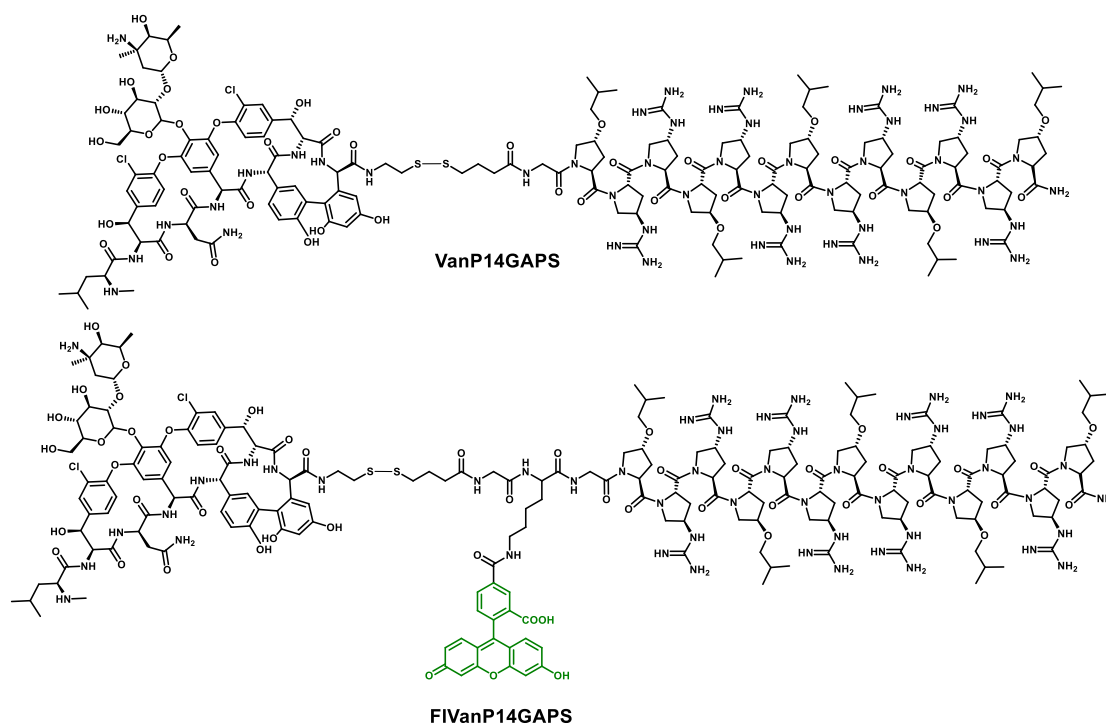


Figure 2.20. Structures of **VanP14GAPS** (top) and **FIVanP14GAPS** (bottom)

Minimum inhibitory concentrations against pathogenic isolates

First, antibacterial activity was determined against a broad selection of pathogenic bacteria (Table 2.2). **VanP14GAPS**, **P14GAP**, vancomycin and gentamicin were incubated with bacteria in logarithmic growth phase for 24 hours, and minimum inhibitory concentrations were visible determined by assessing the growth. The **VanP14GAPS** conjugate exhibits good to moderate activity against almost all pathogens tested, including most of the notable ESKAPE pathogens.¹⁶⁶ Similar to **VanP14S** (Table 2.1), **VanP14GAPS** also displays excellent activity against *S. aureus* strains and moderate activity against a selection of Gram-negative pathogens. **VanP14GAPS** exhibits similar MICs to **VanP14S** in *E. faecium* 700221, *S. aureus* NRS 384 (USA300), *A. baumannii*, *L. monocytogenes* 19111, VRSA 5, and VRSA 10. **VanP14GAPS** showed an enhanced MIC for ESKAPE pathogen *E. cloacae* 1143 at 8 μ M, while **VanP14S** was >16 μ M. A decrease in potency was observed for *P. aeruginosa* 1706 with **VanP14S** and **VanP14GAPS** displaying MICs of 4 μ M and >16 μ M, respectively. When comparing to **P14GAP** from an independent experiment, **VanP14GAPS** provides comparable bactericidal activity in *S. aureus*

and *E. faecium* (0.5 – 2 μ M MICs). **P14GAP**, however, was significantly more potent than **VanP14GAPS** for all Gram-negative strains tested.

Table 2.2. Minimum inhibitory concentrations (MICs) of vancomycin conjugates against pathogenic bacteria isolates using the microbroth dilution assay after 20 h of treatment. Values of **VanP14GAPS** and **P14GAP** reported in μ M; and values of vancomycin and gentamicin controls reported in μ g/mL.

Bacteria strain	VanP14GAPS	P14GAP ^a	Vancomycin	Gentamicin
<i>ESKAPE Pathogens</i>				
<i>E. faecium</i> 700221	1	1	>32	>32
<i>S. aureus</i> NRS 384	0.5	2	1	≤ 0.25
<i>K. pneumoniae</i> 1706	>16	8	>32	1
<i>A. baumannii</i> 1605	8	0.5	>32	>32
<i>P. aeruginosa</i> 50573	16	2	>32	1
<i>E. cloacae</i> 1134	8	2	>32	≤ 0.25
<i>Selected Staphylococcus and Gram-positive strains</i>				
<i>S. aureus</i> NRS 383	0.5	2	0.5	>32
<i>S. aureus</i> NRS 382	1	2	2	0.5
<i>S. aureus</i> NRS 4220	1	2	1	≤ 0.25
<i>S. aureus</i> ATCC 6538	1	1	1	≤ 0.25
<i>S. epidermidis</i> NRS101	1	1	2	16
VRSA 5	1	2	>32	≤ 0.25
VRSA 10	1	2	>32	≤ 0.25
<i>L. monocytogenes</i> 19111	0.5	2	0.5	0.5
<i>Selected Gram-negative strains</i>				
<i>P. aeruginosa</i> 48982	8	4	>32	>32
<i>P. aeruginosa</i> 31040	16	2	>32	32
<i>P. aeruginosa</i> 31041	8	2	>32	>32
<i>A. baumannii</i> 1747	2	0.5	>32	≤ 0.25
<i>A. baumannii</i> 19606	4	1	>32	16
<i>S. flexneri</i> 1a	2	1	>32	≤ 0.25
<i>S. enteritidis</i>	4	1	>32	≤ 0.25
<i>E. coli</i> 25922	8	1	>32	≤ 0.25
<i>S. trphimurium</i> LT2	4	1	>32	≤ 0.25

^a values were taken from an independent analysis⁴²

Cell toxicity studies

When compared to the known lytic AMP, melittin, and established non-lytic AMP/CPP, **P14LRR**, **VanP14GAPS** displays significantly more lysis in hRBCs than the individual components, **P14GAP** and vancomycin (**Figure 2.21 A**). Although the lysis does not occur to the same extent as positive control melittin, we begin to see an increase of lytic behavior at 2.5 μM with 8% lysis, increasing to 15% lysis at 5 μM , and upwards at increasing concentrations. No lysis was seen for **VanP14S** at any of the tested concentrations (**Figure 2.7 A**). While **VanP14GAPS** displays unfortunate hRBC lytic characteristics, the conjugate shows similar low toxicity toward macrophage (J774A.1) cells at 5 μM as compared to **VanP14S** (**Figure 2.21 B**), and most of the population of cells are viable at concentrations up to 10 μM after a 9-hour incubation. These results show the cell viability is comparable to the **VanP14S** conjugate with a significant drop in viability at 20 μM , a concentration well above MIC values.

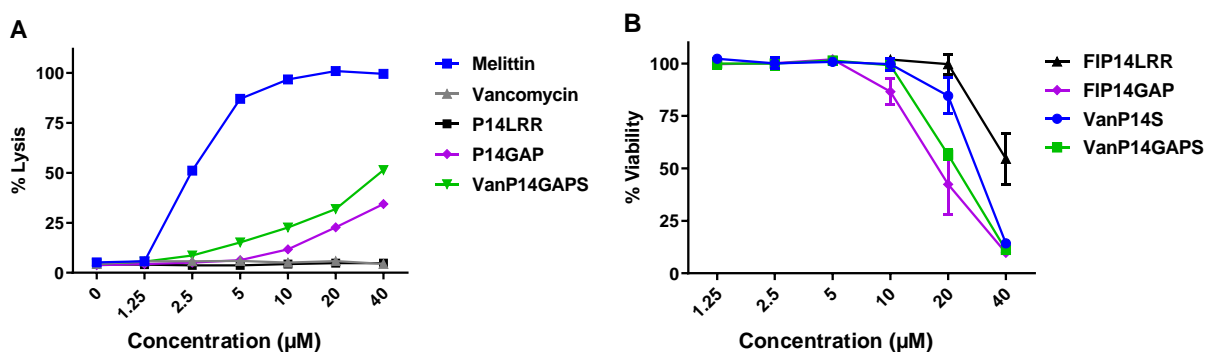


Figure 2.21. (A) Hemolysis assay measuring the release of hemoglobin from hRBC. Melittin was used as a positive control, and values were normalized against 0.1% triton X-100. (B) Cytotoxicity measuring the cell viability using an MTT assay after 9-hour incubation with compounds.

Confocal microscopy to visualize cellular internalization of FIVanGAPS conjugate

FIP14GAP showed increased membrane binding compared to **FIP14LRR**.⁴² To visualize internalization of **FIVanP14GAPS**, and determine localization within mammalian cells, confocal microscopy was applied. Therefore, the fluorescent vancomycin-P14GAP conjugate, **FIVanP14GAPS**, was incubated at a concentration of 5 μM with J774A.1 macrophage cells for 1 hour. Following incubation, we further labeled cells with an endosomal marker (Lysotracker red,

300 nM) or a mitochondrial marker (Mitotracker red, 100 nM). **FIVanP14GAPS** appears to show localization within the endosomes of macrophage cells after 1-hour incubation of 5 μ M concentration (**Figure 2.22 A**). No mitochondrial association is visible (**Figure 2.22 B**). Both **FIP14GAP** and **FIVanP14S** were also localized within endosomes at 5 μ M concentration and timepoint previously determined. After 1 hour incubation with 10 μ M **FIVanP14GAPS**, the same results are observed; there is strong colocalization with the endosomes, and no colocalization with the mitochondria (data not shown). Interestingly, **FIVanP14GAPS** also shows an increase in membrane association from **FIVanP14S**, based on the faintly visible green outline in the cell images. In conjugating vancomycin with the **P14GAP** peptide that showed strong membrane association, the vancomycin-P14GAP conjugate displays a mixture of membrane binding and internal endosomal localization, owing to the predicted mode of endosomal cellular entry.

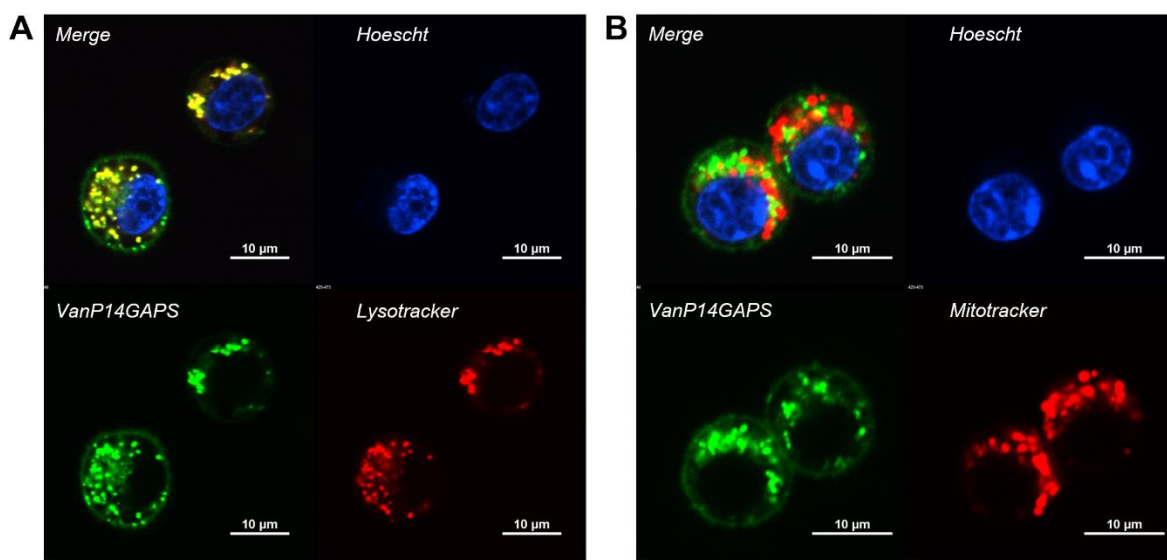


Figure 2.22. Confocal microscopy images of J774A.1 macrophage cells incubated with 5 μ M FIVanP14GAPS for 1 hr. (A) Merge (top left) and separate laser channels with Hoescht 33342 stain (blue), conjugate (green), and lysotracker lysosomal stain (red). (B) Merge (top left) and separate laser channels with Hoescht 33342 nuclear stain (blue), conjugate (green), and mitotracker mitochondrial (red). A yellow-orange color indicates colocalization of conjugate and selected red stain.

Flow cytometry to quantify cellular accumulation of FIVanGAPS

To quantify the internalization of **FIVanP14GAPS** compared to that of **FIVanP14S** and the individual components of the conjugate (vancomycin and **P14GAP**), we measured cellular

fluorescence by flow cytometry. **FIVanP14GAPS**, **FIVanP14S**, **FIP14GAP**, **FIP14LRR**, and **FITC-Van** were incubated with macrophage J774A.1 for 1-hour and 3-hour incubation times; followed by flow cytometry to read cell fluorescence (**Figure 2.23**). Both conjugates, **FIVanP14GAPS** and **FIVanP14S** showed an increase in fluorescence at longer incubation times, exhibiting time-dependent uptake. Both conjugates also exhibited significantly lower (about 3-fold) cell fluorescence than **FIP14GAP** at 5 μ M concentrations, but significantly higher (60-80%) fluorescence than the reference **FIP14LRR** cell-penetrating peptide at both incubation times. **FIVanP14GAPS** showed about a 30% increase in fluorescence after 1 hour compared to **FIVanP14S** at all concentrations, but these conjugates displayed analogous fluorescence after 3-hour incubation. This indicates that **VanP14GAPS** may be initially faster at associating with or internalizing into the cell.

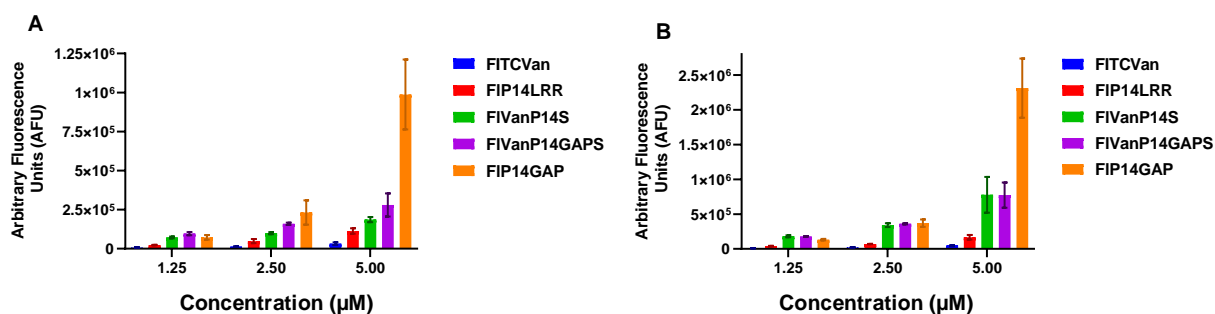


Figure 2.23. Screening of arbitrary fluorescence in J774A.1 cells incubated with compounds at (A) 1-hour and (B) 3-hour incubation times.

To quantify the amount of conjugate that associates and binds to the mammalian cellular membrane, trypan blue was used to quench external fluorescence of the surface-bound conjugates (**Figure 2.24**). **FIP14GAP** was reported to have increased membrane association with mammalian cells compared to **FIP14LRR**, so we wished to investigate whether the conjugate would show a similar association.⁴² Fairly comparable fluorescence was observed in all cases with and without trypan blue. At both 2.5 and 5 μ M concentrations and both time points, over 75% of the **FIVanP14GAPS** conjugate is internalized. Although the confocal images appear as though there may be more significant membrane binding of **FIVanP14GAPS** than **FIVanP14S**, the flow cytometry data show comparable decreases in fluorescence with the addition of trypan blue. These

data indicate that cellular uptake of both vancomycin conjugates is comparable, with only moderate increases in uptake for **FIVanP14GAPS** over **FIVanP14S**.

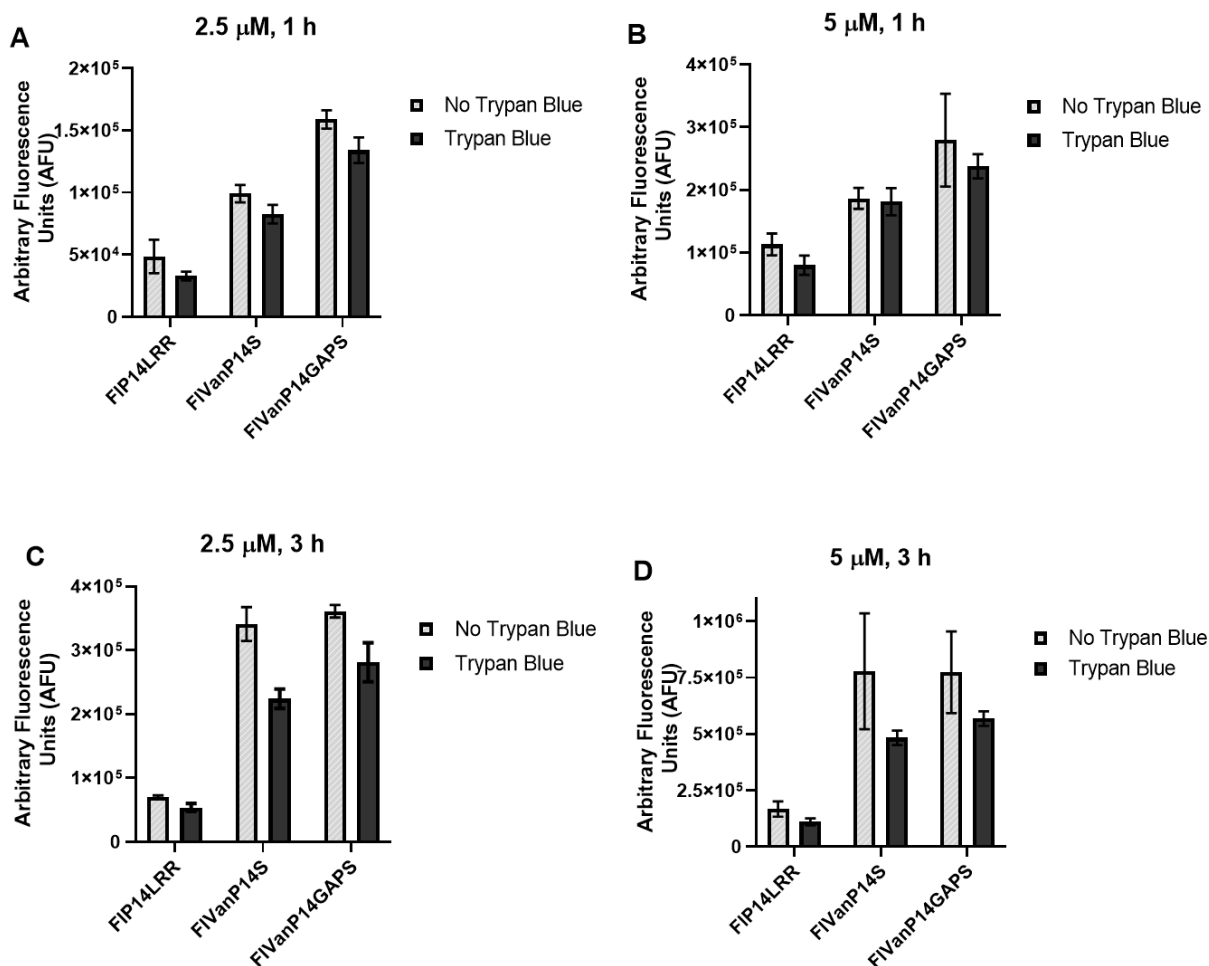


Figure 2.24. Cellular uptake studies of **FIP14LRR**, **FIVanP14S**, and **FIVanP14GAPS** at (A, B) 1 h and (C, D) 3 h in J774A.1 cells after incubation at (A, C) 2.5 μ M and (B, D) 5 μ M with trypan blue treatment (black bar) and without trypan blue treatment (gray bar).

2.3 Conclusions

We have demonstrated that vancomycin-CAPHs conjugates have excellent antimicrobial activity against sensitive and drug-resistant Gram-positive bacteria, in addition to moderate activity against certain Gram-negative. Although outer membrane disruption was determined in Gram-negative bacteria, no inner membrane leakage or membrane depolarization was observed.

Feasibly, the conjugate may act on Gram-negative bacteria by disturbing the outer membrane to access the peptidoglycan or by non-lytic penetration of the cell wall to access internal targets.

Both **VanMalP14** and **VanP14S** show faster killing kinetics than vancomycin with *S. aureus*. In addition, the conjugates exhibit some membrane disruption as identified through membrane leakage and membrane potential experiments. The fluorescent **VanP14S** is shown to localize within the cell wall of *S. aureus*, analogous to vancomycin and **P14LRR** individually; however, much stronger association was observed with 5-fold increases in bacterial fluorescence as compared to **FIP14LRR** alone. Although we have not addressed pentapeptide binding affinity directly, we can propose with the fluorescence data that the addition of a cationic cell-penetrating peptide, vancomycin may show increased antibacterial activity by tighter binding to lipid II in the peptidoglycan. The positive charges in **P14LRR** may enhance vancomycin's cell wall affinity through interactions negatively charged teichoic acid components dispersed in the cell wall and with the negatively charged phospholipid head groups in the membrane as described in literature.⁶⁹ The **P14LRR** cell-penetrating peptide could insert within the membrane and establish an anchoring effect as described in a similar literature example (**Figure 2.26**).⁷⁰ One or more of these interactions could promote **P14LRR** interacting with the cytoplasmic membrane and acting as an anchor for vancomycin to bind its pentapeptide target more efficiently.

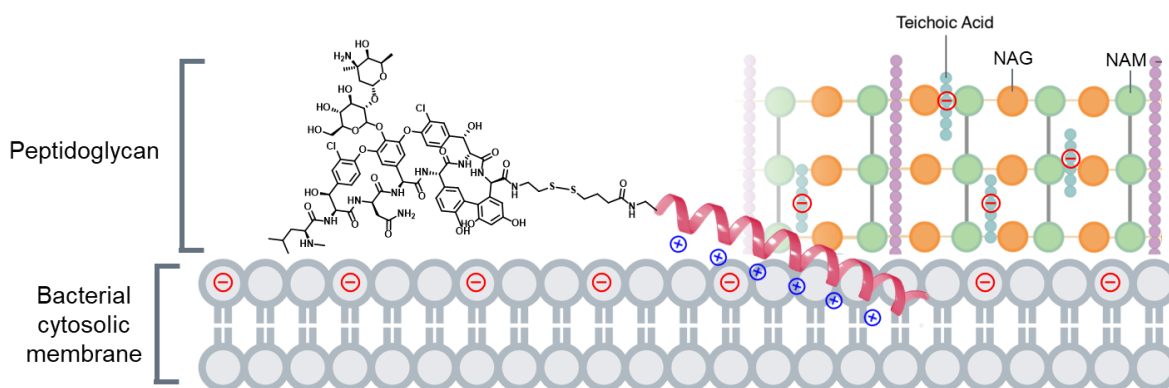


Figure 2.25. Proposed model of vancomycin-CAPHs acting on a bacterial membrane.

We also provided evidence to support that **VanP14S** should be effective as an antibiotic against intracellular pathogens, a subset of infections previously inaccessible for vancomycin. The conjugates, **VanMalP14** and **VanP14S**, showed minimal cytotoxicity and hemotoxicity at

concentrations relevant to the observed MIC values, with enhanced cellular uptake as compared to **P14LRR**, and with intracellular endosomal localization. Additionally, it was shown that **VanP14S** is capable of cleaving in a reducible environment, supporting the assertion that a CAPH could deliver vancomycin intracellularly, allowing both vancomycin and the peptide to synergistically act as independent antibiotics.

An additional conjugate was synthesized, **VanP14GAPS**, to determine if the exchange of a more potent antibiotic CPP, **P14GAP**, would result in a further improved vancomycin conjugate. Although **P14GAP** showed more potency as an antibiotic peptide compared to **P14LRR**, the vancomycin conjugate displayed comparable potency against the same *S. aureus* and enterococcus strains. Although there was some increase in mammalian cellular uptake, red blood cell hemolysis was significantly greater for **VanP14GAPS** than **VanP14S**. Based on these preliminary results, there appears to be no significant advantage in pursuing **VanP14GAPS** over the more promising **VanP14S** or **VanMalP14** conjugates.

2.4 Future Directions

Progress has been achieved with our vancomycin-CAPH conjugates, yet additional experiments are needed. Specifically, we must determine if the conjugates can clear intracellular pathogens. Up to this point, we have shown that the conjugates are excellent antimicrobials and able to accumulate within mammalian cells, but must unequivocally determine that they can accomplish both concurrently. Answering the question of whether the reducibility of the conjugate is essential is also a goal. Additionally, it would be prudent to elucidate the mode of action of these conjugates beyond that of which we have already determined, such as using lipid II binding assays. Undisputedly concluding that these conjugates show tighter binding to the peptidoglycan of bacteria than vancomycin would confirm the mode of action, with the CAPH peptide functioning as an anchor within the cell wall or as a membrane-interacting component.

2.5 Materials and Methods

2.5.1 Materials

Starting material for unnatural amino acids, Fmoc-protected natural amino acids, and coupling reagents were purchased from Chem Impex (Wood Dale, IL) or Ana Spec, Inc. (Fremont,

CA). H-Rink Amide ChemMatrix resin for peptide synthesis was purchased from PCAS Biomatrix Inc. (Quebec, Canada). Vancomycin hydrochloride was purchased from Millipore Sigma (Burlington, MA). Sterile DMEM supplemented with L-glutamine and Penicillin-Streptomycin were purchased from VWR (Batavia, IL). Buffers (1X PBS) and 10% fetal bovine serum (FBS) used in cell culture were purchased from Corning Inc. (Corning, NY) and Atlanta Biologicals, Inc. (Research and Diagnostic Systems, Inc., Minneapolis, MN), respectively. All bacteria and cell lines for culture were purchased from ATCC (Manassas, VA). All other chemicals and reagents were purchased commercially and were used without further purification unless mentioned from Sigma Aldrich (St. Louis, MO), Alfa Aesar (Haverhill, MA) or Thermo Fischer (Waltham, MA).

Peptides and conjugates were purified using a Waters Delta Prep 4000 HPLC equipped with a Phenomenex C18 semi-preparative column. CAPHs characterization was performed using matrix associated laser desorption ionization-time of flight (MALDI-TOF) mass spectrometry using an Applied Biosystem Voyager-DE TM BioSpectrometry workstation and analytical RP-HPLC using Waters Delta Prep 4000 HPLC equipped with a C18 reverse phase analytical column (5 μ m, 4.6 mm x 250 mm; Phenomenex Luna). Cell uptake was measured using BD sciences Acurri and analyzed using BD software. Bacterial fluorescence uptake was measured on BD Fortessa LSR flow cytometry cell analyzer and analyzed using BD software. Absorbance readings were obtained using microplate reader TECAN SpectraFluor Plus or TECAN Infinite F PLEX. Confocal images were obtained using Nikon A1R-MP inverted confocal fluorescence microscope equipped with 60X (mammalian cell culture) or 100X (bacterial culture) oil objective. NIS Elements software was used to process images.

2.5.2 Synthesis of Z-trans-(2-cyanoethyl)hydroxyproline (1)

To a solution of NaH (1.58 g, 65.9 mmol) in THF (150 mL) with 4Å mol sieves at 0 °C and under an atmosphere of nitrogen was added an ice-cooled solution of Cbz-hydroxyproline (5.0 g, 18.8 mmol) in THF (50 mL). The mixture was stirred at 0 °C for 1 h. To this mixture was added acrylonitrile (5.0 mL, 93.4 mmol), and the reaction was allowed to warm to room temperature and stirred for 24 h. The reaction mixture was cooled to 0 °C, and water (100 mL) was added to quench the excess NaH. The THF was removed in vacuo, and 10% HCl was added to bring the solution to a pH of 1. The mixture was vacuum filtered and the resulting solution was extracted with EtOAc, the organic layer was dried over anhydrous MgSO₄, and the solvent was removed *in vacuo*. The

desired product was purified by silica gel column chromatography (95% CH₂Cl₂, 4% MeOH, 1% AcOH) to provide **1** as a colorless oil in 80% yield.

MS (ESI): 319 m/z (M + H⁺).

¹H NMR (300 MHz, CDCl₃): δ 7.34 (m, 5H), 5.15 (m, 2H), 4.54 (m, 1H), 3.65 (m, 3H), 2.59 (t, J = 5.6 Hz, 2H), 2.40 (m, 1H), 2.15 (m, 2H), 2.39 (m, 2H).

2.5.3 Synthesis of Z-trans-(2-cyanoethyl)hydroxyproline (**2**)

To a solution of **1** (600 mg, 1.9 mmol) in 20 mL ethanol was added Pd/C (60 mg, 10% wt/wt). The solution was stirred under 1 atmosphere of hydrogen for 3 h. The solution was gravity filtered through filter paper, the solvent was removed *in vacuo*, and the residue was used in the next step without further purification. The resulting material was solubilized in 10 mL DI water and was cooled to 0 °C, followed by the addition of sodium bicarbonate (475 mg, 5.7 mmol). To this cooled mixture, was added a solution of Fmoc-OSu (699 mg, 2.1 mmol) in 10 mL acetone dropwise, and the resulting slurry was allowed to warm to room temperature and stirred for 20 hours. The reaction was treated with 10% HCl to a pH of 1 and extracted with EtOAc. The organic layers were dried over anhydrous Na₂SO₄, and the solvent was removed *in vacuo*. The desired product was purified by silica gel column chromatography (96% CH₂Cl₂, 3% MeOH, 1% AcOH) to provide **2** as a colorless oil in 99%/quantitative yield.

HRMS (APCI) calculated for C₂₃H₂₂N₂O₅ [M+H⁺] 407.1601 m/z, found 407.1598

¹H NMR (400 MHz, CDCl₃): δ 10.14 (br s, 1H), 7.73 (dd, J = 24 Hz, 8 Hz, 2H), 7.55 (m, 2H), 7.34 (m, 4H), 4.40-4.51 (m, 3H), 4.24-4.36 (m, 1H), 4.13 (m, 1H), 3.51-3.75 (m, 4H), 2.55 (q, J = 8 Hz, 2H), 2.33-2.48 (m, 1H), 2.11-2.26 (m, 1H).

¹³C NMR (100 MHz, CDCl₃): δ 177.2*, 175.6, 155.7, 154.5*, 143.6, 141.2, 129.0, 127.7, 127.0, 124.9, 119.9, 117.4, 67.9, 63.7, 57.9, 57.2*, 51.4, 47.0, 36.6*, 34.7, 19.0 (*indicates minor rotamer)

2.5.4 Synthesis of Fmoc-Pr(Boc)₂

To a solution of **2** (700 mg, 1.7 mmol) in 20 mL methanol was added 1 mL AcOH, then PtO₂ (70 mg, 10% wt/wt). The solution was stirred under 1 atmosphere of hydrogen overnight. The solution was filtered through celite, and the solvent was removed *in vacuo*, and the residue was washed with toluene to remove excess AcOH. The residue was used in the next step without

further purification. The resulting material was solubilized in 15 mL CH₂Cl₂ and was cooled to 0 °C. To this cooled mixture, a solution of triethylamine (0.720 mL, 5.2 mmol) in 5 mL CH₂Cl₂ was added dropwise, then N, N'-Bis-Boc-1-guanylpurazole (695 mg, 2.2 mmol). The resulting slurry was allowed to warm to room temperature and stirred 20 hours. The reaction was washed with saturated sodium bicarbonate, the organic layer was dried over anhydrous Na₂SO₄, and the solvent was removed *in vacuo*. The desired product was purified by silica gel column chromatography (96% CH₂Cl₂, 3% MeOH, 1% AcOH) to provide Fmoc-P_R(Boc)₂ as a white solid in 62% yield. $[\alpha]_D^{22} = -30.6$ (*c* 1.0 CHCl₃).

HRMS calcd for C₃₄H₄₄N₂O₇ [M + H⁺] 653.3187, found 653.3185

¹H NMR (300 MHz, CDCl₃): δ 7.79 (t, J = 7.5 Hz, 2H), 7.62 (t, J = 7.5 Hz, 2H), 7.37 (m, 5H), 4.48 (m, 3H), 4.13 (m, 2H), 3.57 (m, 3H), 3.25 (m, 2H), 2.46-2.20 (m, 2H), 1.78 (m, 2H), 1.50 (s, 18H).

¹³C NMR (75 MHz, CDCl₃): δ 177.2*, 175.9, 163.0, 156.0, 155.8*, 154.75, 153.1, 144.0, 143.8*, 143.7, 141.3, 141.2*, 128.3, 127.8, 127.7*, 127.2, 127.1, 125.1, 120.0, 119.9*, 83.4, 79.8, 68.2, 68.1*, 68.0, 58.2, 57.6*, 51.6*, 51.3, 47.1, 39.8, 36.8, 35.1, 28.8, 28.3, 28.1* (*indicates minor rotamer).

2.5.5 Fmoc-GAP(Boc)₂ synthesis⁴²

To Fmoc-L-Pro(4-NHBoc)-OH(2S,4R) (200 mg, 0.44 mmol) in a 25 mL round bottom flask with a stir bar was added a trifluoroacetic acid (TFA) and dichloromethane (DCM) solution (1:1, 2 mL). The reaction was stirred for 3 h at room temperature. The TFA/DCM solution was removed *in vacuo* and dried under high vacuum for 12 h. The residue was solubilized in DCM (2 mL) with sonication and N,N'-di-tert-butoxycarbonyl-1H-pyrazole-1 (150.9 mg, 0.49 mmol) was added to the reaction flask followed by triethylamine (TEA) (249 μL, 1.77 mmol). The resulting solution was stirred for 12 h at room temperature. The organic solvent was removed *in vacuo* and the crude material was purified by flash chromatography on silica gel with an eluent composed of 93% DCM and 7% methanol (MeOH). The desired fractions were collected and the solvent was removed *in vacuo* to yield a white solid, Fmoc-GAP(Boc)₂ in 50% yield.

HRMS [ESI] C₃₁H₃₉N₄O₈ 595.2762, observed: 595.2756

^1H NMR (400 MHz, CD_3CN): δ 11.48 (br s, 1H), 8.28 (br s, 1H), 7.81 (d, J = 8 Hz, 2H), 7.62 (m, 2H), 7.40 (m, 2H), 7.33 (m, 2H), 4.51 (dd, J = 4 Hz, 8 Hz, 1H), 4.18-4.43 (m, 4H), 3.72 (m, 1H), 3.31 (m, 1H), 2.25-2.38 (m, 2H), 1.42 (q, J = 8 Hz, 18H).

^{13}C NMR (100 MHz, CD_3CN): δ 163.34, 155.81, 154.03, 152.79, 143.87, 141.05, 127.62, 127.08*, 125.00, 119.88, 83.34, 78.82, 67.19, 51.23, 49.28, 48.39*, 46.92, 35.74, 34.49, 27.38, 27.06* (*indicates minor rotamer).

2.5.6 Synthesis of trityl-4-thiobutyric acid

4-butyrothiolactone (1 g, 9.6 mmol) was added to a solution of NaOH (1.96 g, 49 mmol) in 7.5 mL DI water. The mixture was heated to reflux and stirred for 15 minutes. Upon cooling to room temperature, the reaction mixture was poured directly into a separatory funnel with 7 mL of concentrated HCl. The mixture was extracted with diethylether (3X), dried over MgSO_4 , and solvent removed *in vacuo*. In a second round bottom flask, trityl chloride (2.45 g, 1.5 eq.) was added to a round bottom flask and dissolved in 6 mL DMF. The intermediate product (0.7 g) was dissolved in 1 mL DMF and transferred to the solution of trityl-chloride. The flask was rinsed two additional times with 1 mL DMF to ensure complete transfer of intermediate. The final reaction volume was 9 mL. The reaction was left to stir at room temperature for 48 hours. The resulting mixture was poured into a beaker of 75 mL of 10% sodium acetate and the product precipitated. The solid product was isolated by vacuum filtration, dissolved in acetone, and dried overnight to yield 1.5 g of pure, white solid in 42% yield. No additional purification steps followed.

^1H NMR (400 MHz, CDCl_3): δ 7.18-7.42 (m, 15H), 2.32 (t, J = 8 Hz, 2H), 2.23 (t, J = 8 Hz, 2H), 1.68 (quintet, J = 8 Hz, 2H)

2.5.7 Solid Phase Peptide Synthesis of P14LRR CAPHs

A 10 mL peptide synthesis flask was charged with 100 mg H-rink amide resin (0.48 mmol/g loading) for the synthesis of **P14LRR-SH** and **FIP14LRR-SH**. The resin was manually crushed and washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). The resin was swelled in 5 mL DMF with agitation for 30 mins. The first amino acid (2 eq.) was activated by sonication with HATU (2 eq.) and DIEA (4 eq.) in 4 mL DMF, then added to the resin in the peptide flask. Reaction was agitated for 3 hr then washed with 5 mL of DMF (2X), DCM

(2X), MeOH (2X), DCM (2X), DMF (2X). The Fmoc protecting group was deprotected by adding 4 mL of 30% piperidine in DMF, and mixing for 20 minutes. The solution was drained, and resin washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). This cycle of coupling and deprotecting continued until all amino acids were coupled. The process was monitored by Kaiser¹⁶⁷ and chloranil¹⁶⁸ tests, and verified using test cleavages and MALDI-ToF mass spectrometry.

Synthesis of P14LRR-SH

Upon synthesis of the 100 mg resin scale of **P14LRR** peptide, the resin was split and half (about 50 mg) resin was used. The trityl-4-mercaptopbutyric acid (26.1 mg, 3 mmol) was activated with HATU (27.4 mg, 3 mmol) and diisopropylethylamine (25 μ L, 6 mmol) in 4 mL DMF. This mixture was added to the peptide-bound resin and agitated at room temperature for 3 hours. The reaction solution was drained and the resin and washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). **P14LRR-SH** was then globally deprotected and cleaved from the resin using 5 mL cleavage cocktail of 90: 5: 2.5: 2.5, TFA: EDT: TIPS: H₂O. The cleavage cocktail with the peptide-bound resin was agitated for 3 hours. The crude mixture was collected into a preweighed plastic falcon tube. The resin was washed with 5 mL DCM and collected into the preweighed plastic falcon tube. A second 5 mL aliquot of cleavage cocktail was added to the flask and agitated for an additional 30 minutes and drained into the collecting tube. The resin was washed with 5 mL DCM and collected. The crude mixture was concentrated *in vacuo*. Cold diethyl ether (15 mL) was added to the crude mixture, which led to a precipitate. After centrifugation, the supernatant was decanted and the precipitate was dried under vacuum overnight. The crude **P14LRR-SH** was resuspended in deionized water (approximately 10 mg/mL) and was purified on RP-HPLC, using a C18 column (Phenomenex, USA) with a gradient of 25-65% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 21.5 mg (33% yield). The mass was confirmed using MALDI-TOF (Expected Mass: 2718 Da, Observed Mass: 2718 Da). Analytical HPLC trace shown in **Figure A 1**.

Synthesis of FIP14LRR-SH

Peptide synthesis proceeded as described previously with the following changes. Synthesis of **FIP14LRR-SH** on a 50 mg scale of H-link amide resin was performed. After the final Fmoc-P_L, the addition of a glycine (28.5 mg, 4 mmol), an MTT-protected lysine (60.0 mg, 4 mmol), and a second glycine (28.5 mg, 4 mmol) were activated with HATU (36.5 mg, 4 mmol) and DIEA (33.4 μ L, 8 mmol); and coupled for 2 hours each. The MTT group was deprotected with the addition of 4 mL 30% HFIP in DCM for 20 minutes (2X). A solution of fluorescein-NHS (12.5 mg, 1.1 mmol eq) and diisopropylethylamine (8.6 μ L, 2.2 mmol) in 4 mL DMF was coupled for 4 hours and washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). The terminal glycine was deprotected by adding 4 mL of 30% piperidine in DMF, mixed for 20 minutes, and washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). Trityl-4-mercaptobutyric acid (34.8 mg, 4 mmol), activated with HATU (36.5 mg, 4 mmol) and DIEA (33.4 μ L, 8 mmol) in 4 mL DMF, was added to peptide-bound resin and mixed for 3 hours. The reaction solution was drained and the resin was washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). **FIP14LRR-SH** was then globally deprotected and cleaved from the resin using 5 mL cleavage cocktail of 90: 5: 2.5: 2.5, TFA: EDT: TIPS: H₂O. The cleavage cocktail with the peptide-bound resin was agitated for 3 hours. The crude mixture was collected into a preweighed plastic falcon tube. The resin was washed with 5 mL DCM and collected into the preweighed plastic falcon tube. A second 5 mL aliquot of cleavage cocktail was added to the flask and agitated for an additional 30 minutes and drained into the collecting tube. The resin was washed with 5 mL DCM and collected. The crude mixture was concentrated *in vacuo*. Cold diethyl ether (15 mL) was added to the crude mixture, which led to a precipitate. After centrifugation, the supernatant was decanted and the precipitate was dried under vacuum overnight. The crude **FIP14LRR-SH** was resuspended in deionized water (approximately 10 mg/mL) and was purified on RP-HPLC, using a C18 column (Phenomenex, USA) with a gradient of 30-60% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 10.5 mg (13.4% yield). The mass was confirmed using MALDI-TOF mass spectrometry (Expected Mass: 3261 Da, Observed Mass: 3261.5 Da). Analytical HPLC trace shown in **Figure A 2**.

2.5.8 Solid Phase Peptide Synthesis of P14GAP CAPHs

A 15 mL peptide synthesis flask was charged with 200 mg H-rink amide resin (0.45 mmol/g loading) for the synthesis of **P14GAP-SH** and **FIP14GAP-SH**. The resin was manually crushed and washed with 7 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). The resin was swelled in 7 mL DMF with agitation for 30 mins. The first amino acid (2.5 eq.) was activated by sonication with HATU (2.5 eq.) and DIEA (5 eq.) in 7 mL DMF, then added to the resin in the peptide flask. Reaction was agitated for 3 hr then washed with 7 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). The Fmoc protecting group was deprotected by adding 7 mL of 30% piperidine in DMF, and mixing for 20 minutes. The solution was drained, and resin washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). This cycle of coupling and deprotecting continued until all amino acids were coupled. After the final Fmoc-P_L, the addition of a glycine (107 mg, 4 mmol) was activated with HATU (136.9 mg, 4 mmol) and DIEA (125 μ L, 8 eq); and coupled for 2 hours each. The process was monitored by Kaiser¹⁶⁷ and chloranil¹⁶⁸ tests, and verified using test cleavages and MALDI-ToF mass spectrometry.

Synthesis of P14GAPSH

Upon synthesis of the 200 mg resin scale of **P14GAP** peptide, the resin was split and half (about 100 mg) resin was used. The trityl-4-mercaptopbutyric acid (65.25 mg, 4 mmol) was activated with HATU (68.4 mg, 4 mmol) and diisopropylethylamine (64.2 μ L, 8 mmol) in 4 mL DMF. This mixture was added to the peptide-bound resin and agitated at room temperature for 2 hours. The reaction solution was drained and the resin and washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). **P14GAP-SH** was then globally deprotected and cleaved from the resin using 5 mL cleavage cocktail of 90: 5: 2.5: 2.5, TFA: EDT: TIPS: H₂O. The cleavage cocktail with the peptide-bound resin was agitated for 3 hours. The crude mixture was collected into a preweighed plastic falcon tube. The resin was washed with 5 mL DCM and collected into the preweighed plastic falcon tube. A second 5 mL aliquot of cleavage cocktail was added to the flask and agitated for an additional 30 minutes and drained into the collecting tube. The resin was washed with 5 mL DCM and collected. The crude mixture was concentrated *in vacuo*. Cold diethyl ether (15 mL) was added to the crude mixture, which led to a precipitate. After

centrifugation, the supernatant was decanted and the precipitate was dried under vacuum overnight. The crude **P14GAP-SH** was resuspended in deionized water (approximately 10 mg/mL) and was purified on RP-HPLC, using a C18 column (Phenomenex, USA) with a gradient of 20-75% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 25.1 mg (25% yield). The mass was confirmed using MALDI-TOF (Expected Mass: 2255 Da, Observed Mass: 2255 Da). Analytical HPLC trace shown in **Figure A 9**.

Synthesis of FIP14GAPSH

Peptide synthesis proceeded as described previously with the following changes. Synthesis of **FIP14GAP-SH** on a 50 mg scale of H-rink amide resin was performed. After the final addition of Fmoc-glycine, an MTT-protected lysine (56.2 mg, 4 mmol), and a second glycine (26.7 mg, 4 mmol) were activated with HATU (34.2 mg, 4 mmol) and DIEA (31.2 μ L, 8 mmol); and coupled for 2 hours each. The MTT group was deprotected with the addition of 4 mL 30% HFIP in DCM for 20 minutes (2X). A solution of fluorescein-NHS (11.7 mg, 1.2 mmol eq) and diisopropylethylamine (8.6 μ L, 2.4 mmol) in 4 mL DMF was coupled for 5 hours and washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). The terminal glycine was deprotected by adding 4 mL of 30% piperidine in DMF, mixed for 20 minutes, and washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). Trityl-4-mercaptopbutyric acid (32.6 mg, 4 mmol), activated with HATU (34.2 mg, 4 mmol) and DIEA (31.2 μ L, 8 mmol) in 4 mL DMF, was added to peptide-bound resin and mixed for 2 hours. The reaction solution was drained and the resin was washed with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X). **FIP14GAP-SH** was then globally deprotected and cleaved from the resin using 5 mL cleavage cocktail of 90: 5: 2.5: 2.5, TFA: EDT: TIPS: H₂O. The cleavage cocktail with the peptide-bound resin was agitated for 3 hours. The crude mixture was collected into a preweighed plastic falcon tube. The resin was washed with 5 mL DCM and collected into the preweighed plastic falcon tube. A second 5 mL aliquot of cleavage cocktail was added to the flask and agitated for an additional 30 minutes and drained into the collecting tube. The resin was washed with 5 mL DCM and collected. The crude mixture was concentrated *in vacuo*. Cold diethyl ether (15 mL) was added to the crude mixture, which led to a precipitate. After centrifugation, the supernatant was decanted and the precipitate was dried under vacuum overnight. The crude **FIP14GAP-SH** was resuspended

in deionized water (approximately 10 mg/mL) and was purified on RP-HPLC, using a C18 column (Phenomenex, USA) with a gradient of 20-75% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 6.1 mg (9.7% yield). The mass was confirmed using MALDI-TOF mass spectrometry (Expected Mass: 2798 Da, Observed Mass: 2798 Da). Analytical HPLC trace shown in **Figure A 10**.

2.5.9 Synthesis of Vancomycin tethers

Synthesis was adapted from literature procedure.⁷⁰ Vancomycin (100 mg, 69 mmol) was dissolved in 1 mL DMSO. A solution of N-(2-aminoethyl)maleimide (Cayman Chemical, 10.2 mg, 58 mmol) or 2-(pyridyldithio)ethylamine (Biosynth Carbosynth, 12.8 mg, 57 mmol) in 1 mL DMF was added dropwise to the vancomycin solution. PyBOP (benzotriazol-1-yl-oxytripyrrolidinophosphonium hexafluorophosphate, (41.9 mg, 81 mmol) was dissolved in 100 μ L DMF and added to the vancomycin solution. The solution stirred at room temperature for 20 hours. A sample of the reaction was analyzed on a Waters Acquity UPLC with a SQD2 mass spectrometer to visualize the disappearance of vancomycin and appearance of the desired product, N-(2-aminoethyl)maleimido-vancomycin (Van Tether 1) or 2-(2-pyridinyl)dithioethylamino-vancomycin (Van Tether 2). Upon consumption of starting material, the product was precipitated with 5 mL diethylether and supernatant decanted. A second aliquot of 5 mL diethylether was added and decanted to wash the precipitate. The crude product was carried on to next steps without further purification.

2.5.10 Synthesis and purification of Vancomycin-P14 conjugates

VanP14S

Activated 2-(2-pyridinyl)dithioethylamino-vancomycin (Van Tether 2) (59.4 mg, 100% conversion, 37 mmol) was dissolved in 2 mL 1:1 water: acetonitrile. P14LRR-SH (10 mg, 4 mmol) was dissolved in 1 mL water and added to the vancomycin reaction flask. Diisopropylethylamine (3.2 μ L, 18.4 mmol) was added and reaction stirred overnight. The crude mixture was purified by RP-HPLC using a C18 column (Phenomenex, USA) with a gradient of 5-60% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by

UV at 214 nm. Final product mass was 10.3 mg (51% yield). The mass was confirmed using MALDI-TOF mass spectrometry (Expected Mass: 4223 Da, Observed Mass: 4224 Da). Analytical HPLC trace shown in **Figure A 3**.

FlVanP14S

Activated 2-(2-pyridinyl)dithioethylamino-vancomycin (Van Tether 2) (52 mg, 100% conversion, 32 mmol) was dissolved in 2 mL 1:1 water: acetonitrile. **FIP14LRR-SH** (10.5 mg, 3.2 mmol) was dissolved in 1 mL water and added to the vancomycin reaction flask. Diisopropylethylamine (2.8 μ L, 16.1 mmol) was added and reaction stirred overnight. The crude was purified using C18 column (Phenomenex, USA) with a gradient of 10-80% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 11.2 mg (73% yield). The mass was confirmed using MALDI-TOF mass spectrometry (Expected Mass: 4766 Da, Observed Mass: 4768 Da). Quantification of concentration was determined by UV-vis spectrometry and Beer's Law at 495 nm wavelength and 70,000 molar extinction coefficient. Analytical HPLC trace shown in **Figure A 5**.

VanMalP14

Activated N-(2-aminoethyl)maleimido-vancomycin (Van Tether 1) (54 mg, 100% conversion, 34 mmol) was dissolved in 500 μ L DMSO and added to 1.5 mL acetonitrile. **P14LRR-SH** (8 mg, 3 mmol) was dissolved in 1 mL water and added to the vancomycin reaction flask and reaction stirred overnight. The crude was purified using C18 column (Phenomenex, USA) with a gradient of 5-70% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 5.3 mg (42% yield). The mass was confirmed using MALDI-TOF mass spectrometry (Expected Mass: 4286 Da, Observed Mass: 4289 Da). Analytical HPLC trace shown in **Figure A 4**.

VanP14GAPS

Activated 2-(2-pyridinyl)dithioethylamino-vancomycin (Van Tether 2) (52 mg, 100% conversion, 32 mmol) was dissolved in 2 mL 1:1 water: acetonitrile. Approximately 10.0 mg of

P14GAP-SH (4.4 mmol) in 1 mL DI water was added to the vancomycin reaction flask (2254). Diisopropylethylamine (3.9 μ L, 22.4 mmol) was added and reaction stirred for 48 hours. The crude was purified using C18 column (Phenomenex, USA) with a gradient of 5-70% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 9.2 mg (55% yield). The mass was confirmed using MALDI-TOF mass spectrometry (Expected Mass: 3758 Da, Observed Mass: 3762 Da). Analytical HPLC trace shown in **Figure A 11**.

FlVanP14GAPS

Activated 2-(2-pyridinyl)dithioethylamino-vancomycin (Van Tether 2) (52 mg, 100% conversion, 32 mmol) was dissolved in 2 mL 1:1 water: acetonitrile. **FIP14GAP-SH** (6.1 mg, 2.2 mmol) was dissolved in 1 mL water and added to the vancomycin reaction flask. Diisopropylethylamine (1.9 μ L, 10.9 mmol) was added and reaction stirred for 48 hours. The crude was purified using C18 column (Phenomenex, USA) with a gradient of 5-60% acetonitrile (0.1% TFA) in water (0.1% TFA) with a flow rate of 12 mL/min over 60 minutes and was visualized by UV at 214 nm. Final product mass was 5.6 mg (58% yield). The mass was confirmed using MALDI-TOF mass spectrometry (Expected Mass: 4301 Da, Observed Mass: 4308 Da). Quantification of concentration was determined by UV-vis spectrometry and Beer's Law at 495 nm wavelength and 70,000 molar extinction coefficient. Analytical HPLC trace shown in **Figure A 12**.

2.5.11 Synthesis of Vancomycin-SH

Vancomycin-SH was used to calibrate a standard curve for the disulfide reduction of **VanP14S**. To a 1-dram glass vial, 940 μ L of crude Van Tether 2 in DI water (approximately 50 mg, 30.9 mmol) was added. Following the addition of 1 mL of acetonitrile, the solution was degassed for 30 minutes with N₂. Dithiothreitol (47.7 mg, 309 mmol) was added and the solution was stirred at room temperature for 20 hours. Completion of the reaction was confirmed by ESI-MS. The reaction mixture was filtered and purified by RP-HPLC (C18 column), 10-70% acetonitrile in water with 0.1% TFA over 60 min. The appropriate fractions were isolated and

lyophilized to afford 33.7 mg of white powder in 72% yield. Analytical HPLC trace shown in **Figure A 8**.

2.5.12 Synthesis of fluorescent vancomycin compounds

FITC-Vancomycin

The procedure was adapted from literature procedure.⁶⁹ Vancomycin (16.1 mg, 0.011 mmol) was dissolved in 2.5 mL of carbonate/bicarbonate buffer (pH 10) in a 1-dram vial charged with a stir bar. Fluorescein isothiocyanate (FITC) (25 mg, 0.064 mmol, 6.0 molar equiv) was dissolved in 100 μ L DMSO and added to the reaction vial. The solution was stirred at 4 °C overnight. The reaction mixture was filtered and purified by RP-HPLC (C18 column), 20-85% acetonitrile in water with 0.1% TFA over 60 min. The appropriate fractions were isolated and lyophilized to afford an orange powder (6.2 mg, 26% yield). Two peaks were isolated by HPLC, accounting for the mixed isomers of FITC starting material. The mass was confirmed using MALDI-ToF. Expected Mass: $[M+H]^+$ 1839 Da; Observed Mass: $[M+Na]^+$ 1861 Da. Regiochemistry confirmed by loss of vancosamine sugar(s) 1696, 1533 Da. Analytical HPLC trace shown in **Figure A 6**.

RITC-Vancomycin

Procedure was adapted from literature protocol.⁶⁹ Vancomycin (11.2 mg, 0.008 mmol) was dissolved in 2.5 mL of carbonate/bicarbonate buffer (pH 10) in a 1-dram vial charged with a stir bar. Rhodamine isothiocyanate (RhITC) (25 mg, 0.047 mmol, 6.0 molar equiv) was dissolved in 100 μ L DMSO and added to the reaction vial. The solution was stirred at 4 °C overnight. The reaction mixture was filtered and purified by RP-HPLC (C18 column), 15-85% acetonitrile in water with 0.1% TFA over 60 min. The appropriate fractions were isolated and lyophilized to afford a red powder (5.8 mg, 29% yield). Two peaks were isolated by HPLC, accounting for the mixed isomers of RhITC starting material. The mass was confirmed using MALDI-ToF. Expected Mass: $[M+H]^+$ 1950 Da; Observed Mass: $[M+H]^+$ 1950 Da. Regiochemistry confirmed by loss of vancosamine sugar(s) 1787, 1645 Da. Analytical HPLC trace shown in **Figure A 7**.

2.5.13 VanP14S disulfide reduction in the presence of DTT

VanP14S (25 μ M) was incubated at 37 °C with 10 mM DTT in degassed phosphate buffered saline (PBS 1X, pH 7.4) containing 50 μ M 6-quinolinecarboxylic acid as the internal standard in 1.0 mL total volume. When monitoring VanSH release, an aliquot of the reaction mixture was taken at different time points and directly analyzed by UPLC-MS. The time points were analyzed using RP-UPLC with a C18 column consisting of 2-50% acetonitrile in water with 0.1% formic acid (0.5 ml/min, column temperature of 40 °C). The peaks corresponding to the m/z for quinolinecarboxylic acid (+1: 174), VanSH (+1: 2113), and P14LRR-SH (+5: 545) were detected and extracted using MassLynx software. This experiment was run a single time, and the percentage of release vs time was fitted using Graph Pad Prism 7 to generate half-life.

2.5.14 VanP14S disulfide reduction in the presence glutathione

VanP14S (25 μ M) was incubated at 37 °C with 10 mM glutathione (GSH) in degassed phosphate buffered saline (PBS 1X, pH 7.4) containing 50 μ M 6-quinolinecarboxylic acid as the internal standard in 1.0 mL total volume. When monitoring VanSH release, an aliquot of the reaction mixture was taken at different time points and directly analyzed by UPLC-MS. The time points were analyzed using RP-UPLC with a C18 column consisting of 2-50% acetonitrile in water with 0.1% formic acid (0.5 ml/min, column temperature of 40 °C). The peaks corresponding to the m/z for quinolinecarboxylic acid (+1: 174), VanSH (+1: 2113), and P14LRR-SH (+5: 545) were detected and extracted using MassLynx software. This experiment was run in duplicate, and the percentage of release vs time was fitted using Graph Pad Prism 7 to generate half-life.

2.5.15 Minimum Inhibitory Concentration (MIC)

The minimum inhibitory concentrations (MICs) of the conjugates and control drugs were determined using the broth microdilution method, according to guidelines outlined by the Clinical and Laboratory Standards Institute (CLSI)¹⁶⁹ against clinically-relevant bacterial strains. Bacterial strains were grown aerobically overnight on tryptone soy agar plates at 37°C. Afterwards, a bacterial solution equivalent to 0.5 McFarland standard was prepared and diluted in cation-adjusted Mueller-Hinton broth (CAMHB) (except *E. faecium* and *L. monocytogenes*) to achieve a bacterial concentration of about 5×10^5 CFU/mL. *E. faecium* and *L. monocytogenes* were diluted

in tryptone soy broth (TSB) to achieve a bacterial concentration of about 5×10^5 CFU/mL. conjugates and control drugs were added in the first row of the 96-well plates, and serially diluted with the corresponding media containing bacteria. Plates were then, incubated aerobically at 37 °C for 18-20 hours before recording the MIC results. MICs reported are the minimum concentrations of the compounds and control drugs that could completely inhibit the visual growth of bacteria.

2.5.16 Time-kill kinetics assay of vancomycin-P14 conjugates

In order to evaluate the mode of killing of vancomycin-P14 conjugates, a standard time kill assay was performed against MRSA USA 300 as described previously.¹⁷⁰⁻¹⁷¹ MRSA USA300 cells in logarithmic growth phase (OD₆₀₀ ~1.00) were diluted to ~10⁶ colony-forming units (CFU/mL) and exposed to concentrations equivalent to either 2X or 5X MIC (in triplicate) of VanP14S, VanMalP14 and vancomycin in tryptic soy broth. Water (the solvent of conjugates) was used as a negative control. Aliquots (100 µL) were collected from each treatment after 0, 2, 4, 6, 8, 12, and 24 hours of incubation at 37 °C and subsequently serially diluted in PBS. Bacteria were then, transferred to tryptic soy agar plates and incubated at 37 °C for 18-20 hours before viable CFU/mL was determined.

2.5.17 *E. coli* membrane disruption assessment by β-galactosidase leakage

Peptide solutions were prepared in deionized water (DIW) to make concentrations at 2X and 4X the determined MIC values for peptides and conjugates against *E. coli* ATCC 25922. Five stationary colonies of *E. coli* ATCC 25922 were inoculated in 50 mL Mueller Hinton Broth (MHB) and grown at 37 °C for about 2 hours until the OD₅₉₅ reached 0.1. Following, 5 mL of isopropyl-β-D thiogalactopyranoside (IPTG) (24 mg/10 mL PBS) was added to bacteria culture for final concentration of 1 mM. The bacteria was induced until an OD₅₉₅ between 0.3 and 0.5 was reached. A 10 mL aliquot of the bacteria in the IPTG/MHB media was removed and centrifuged at 3500 rpm for 5 min and spent media was decanted. The pelleted bacteria was then washed twice with 10 mL of fresh MHB and resuspended in 10 mL of MHB. The bacteria were plated in a 96-well plate at 90 µL per well, followed by the addition of 10 µL of the peptide solution. The bacteria were incubated with the peptide treatment for 1 hour. The 96-well plate was centrifuged for 5 min at 3500 rpm, and 80 µL of the supernatant was transferred to a new 96-well plate. Then, 20 µL of

a freshly prepared 2-nitrophenyl- β -D-galactopyranoside (ONPG) (40 mg/10 mL of PBS) solution was added to each well for a final concentration of 0.8 mg/mL. The 405 nm absorbance values were read every 5 min over the course of 1 h using a TECAN microplate reader to monitor the levels of o-nitrophenol from the β -galactosidase release from *E. coli*. The leakage was calculated based on the absorbance value obtained for treated samples with a time zero absorbance value subtracted from each time point. For all experiments melittin was used as a positive control, PBS 1X was used as a negative control, and Fl-P14LRR was included as a reference. Data was obtained in duplicates from at least two independent experiments.

2.5.18 *E. coli* outer membrane disruption assay

The effect of peptides on the bacterial outer membrane permeability was characterized by measuring the uptake of N-phenyl-1-naphthylamine (NPN) adapted from a previous study.¹⁷² Stationary colonies of *E. coli* ATCC 25922 were inoculated in Mueller Hinton Broth (MHB) and grown at 37 °C to mid-exponential phase with optical density at 595 nm between 0.5 and 0.6. An aliquot was centrifuged at 3500 rpm for 5 minutes and the supernatant was removed. Bacteria were washed with 5 mM HEPES, 5 mM glucose buffer, pH 7 and centrifuged at 3500 rpm for 5 minutes, the supernatant was removed, and the bacteria were resuspended in buffer. Bacteria were plated in a black-walled 96-well plate (Greiner Bio One, Catalog No.07-000-088) to a final OD₅₉₅ of 0.5 and NPN was added to a final concentration of 10 μ M. Fluorescence was measured with excitation of 360 nm and emission at 415 nm on a TECAN Infinite F PLEX plate reader with readings every 2 minutes. A baseline was established by reading the wells for 4 minutes. Sample concentrations of 10 μ L peptides, 1% Triton X-100 (positive control), and DIW (negative control) were added and fluorescence measured for a total experiment time of 30 minutes. Data were normalized by using Triton X as 100% permeation of the outer membrane, and by taking the raw values at 16 minutes where the fluorescence stabilized. $(F_{\text{obs}} - F_0) / (T_{100} - T_0)$ where F_{obs} is the fluorescence at 16 minutes, F_0 is fluorescence at time zero, T_{100} is fluorescence of Triton X-100 at 16 minutes, and T_0 is fluorescence of Triton X at time zero. An ordinary one-way ANOVA for multiple comparisons was used to test statistical significance ($P < 0.05$) between results for each treatment and the untreated samples. Data were obtained in duplicates from at least two independent experiments.

2.5.19 MRSA cell membrane disruption assessment

Release of 260 and 280 nm absorbing components of nucleic acids was utilized in order to investigate the effect of the vancomycin-P14 conjugates on the integrity of the MRSA cell membrane, as previously described.¹⁷³⁻¹⁷⁴ A logarithmic phase culture of MRSA USA300 was incubated with 5X MIC concentrations of **VanP14S** or **VanMalP14** (in triplicate) at 37°C for 30 minutes. Untreated MRSA cells (H₂O) and vancomycin-treated cells (in triplicates) served as negative controls. Lysostaphin (20 µg/mL, in 50 mM Tris-HCl, pH 8.00) was used as a positive control due to its ability to disrupt the staphylococcal cell membrane.¹⁷⁵⁻¹⁷⁶ The concentration of released DNA (ng/µL) was determined using the standard nucleic acid quantification function of a NanoDrop spectrophotometer (ThermoScientific). Data are presented as the concentration of released DNA (ng/µL) for each test agent compared to that for the positive control (lysostaphin). A one-way ANOVA with post-hoc Dunnett's test for multiple comparisons was used to test statistical significance ($P < 0.05$) between results for each treatment and the untreated samples (H₂O).

2.5.20 Membrane depolarization assay

The cytoplasmic membrane depolarization activity of the peptides were characterized with the membrane potential-sensitive dye 3,3'-dipropylthiadicarbocyanine iodide (diSC₃₋₅) adapted from a previous study.¹⁷² Stationary colonies of *E. coli* ATCC 25922 or *S. aureus* ATCC 10537 were inoculated in Mueller Hinton Broth (MHB) and grown at 37 °C to mid-exponential phase with an optical density at 595 nm between 0.5 and 0.6. An aliquot was centrifuged at 3500 rpm for 5 minutes and supernatant removed. Bacteria were washed with 5 mM HEPES, 20 mM glucose, 0.1 M KCl buffer, pH 7 and centrifuged at 3500 rpm for 5 minutes. The supernatant was removed and the bacteria were resuspended in buffer. Bacteria were plated in a black-walled 96-well plate (Greiner Bio One, Catalog No.07-000-088) to a final OD₅₉₅ of 0.1 and diSC₃₋₅ was added to a final concentration of 1 µM. The bacteria were incubated at 37 °C for 1 hour. Fluorescence was measured with an excitation wavelength of 620 nm and emission at 670 nm on a TECAN Infinite F PLEX plate reader with readings every 30 seconds. A baseline was established by reading the wells for 1 minute. Sample concentrations of 10 µL peptides, 1% Triton X-100 (positive control), and DIW (negative control) were added and fluorescence measured for a total experiment time of

5 minutes. Data were normalized using the raw data at 4 minutes where the fluorescence stabilized. $(F_{\text{obs}} - F_0) / (T_{100} - T_0)$ where F_{obs} is the fluorescence at 4 minutes, F_0 is fluorescence at time zero, T_{100} is fluorescence of Triton X at 4 minutes, and T_0 is fluorescence of Triton X at time zero. An ordinary one-way ANOVA for multiple comparisons was used to test statistical significance ($P < 0.05$) between results for each treatment and the untreated samples. Data were obtained in duplicates from at least two independent experiments.

2.5.21 Flow cytometry of bacterial uptake

The procedure was adapted from literature procedure.⁶⁹ Stationary colonies of *E. coli* ATCC 25922 or *S. aureus* ATCC 10537 were inoculated in Mueller Hinton Broth (MHB) and grown at 37 °C to mid-exponential phase with an optical density at 595 nm near 0.5. A 15 mL aliquot was removed and centrifuged at 3500 rpm for 5 minutes. The supernatant was removed, and the pellet was resuspended in 15 mL fresh MHB. The bacterial culture was separated into 500 µL aliquots in Eppendorf tubes. Bacteria were centrifuged at 6000 rpm for 5 minutes, the media was removed, and the bacteria were resuspended in 300 µL of conjugate concentrations in PBS 1X. Bacteria were incubated for 10 minutes or 1 hour at 37 °C, then centrifuged at 6000 rpm for 5 minutes and washed with PBS 1X twice. Bacterial pellets were finally resuspended in 1 mL PBS 1X or 1 mL PBS with 2 mg/mL trypan blue. The fluorescence of the bacteria cells was measured using a BD Fortessa LSR flow cytometry cell analyzer (BD Biosciences). All samples were run in duplicate. The mean arbitrary fluorescence values of gated cells were measured and recorded upon excitation of the fluorophore, fluorescein, using the 488 nm laser. For each experiment, a negative control of cells that were not incubated with fluorescent compound (PBS only) was also analyzed.

2.5.22 Confocal microscopy of live bacteria

Stationary colonies of *E. coli* ATCC 25922 or *S. aureus* ATCC 10537 were inoculated in Mueller Hinton Broth (MHB) and grown at 37 °C to mid-exponential phase with optical density at 595 nm between 0.5 and 0.6. An aliquot was centrifuged at 3500 rpm for 5 minutes and the supernatant was removed. Bacteria were resuspended in fresh MHB and washed a second time. After resuspending with fresh MHB, 90 µL bacteria were plated and 10 µL peptide concentrations were added to a 96-well plate. The plate was incubated for 10 min or 1 hr at 37 °C. Aliquots of

bacteria were centrifuged at 12,000 rpm for 2 minutes and washed 2X with PBS 1X buffer. Bacteria were fixed with 4% paraformaldehyde in PBS for 30 minutes at room temperature. Bacterial samples were loaded on a #1.5 poly-lysine coverslip (Neuvitro) and imaged using Nikon A1R Multiphoton inverted confocal microscope using a 100X oil objective with 488 nm laser line employed.

2.5.23 Hemotoxicity of conjugates on human red blood cells

Fresh human red blood cells (hRBCs) (Innovative Research, cat # IWB3CPDA1) were collected by centrifugation at 2000 rpm for 5 min followed by washing three times with 5 mL PBS 1X, pH 7.4. The supernatant of the final wash was aspirated and 200 μ L of the cell pellet was added to 4.8 mL of PBS to make a 4% suspension (v/v), with a total volume of 5 mL. To a 96-well plate, 50 μ L of the hRBC solution was added, followed by 50 μ L of peptide treatment prepared in PBS to achieve a 2-fold dilution of peptide and a final suspension of 2% (v/v) of hRBCs. The plate was incubated at 37 °C under 5% CO₂ for 1 h. The plate was subsequently centrifuged at 1200 rpm for 5 min at 4 °C. Next 75 μ L aliquots of the supernatants in each well were carefully transferred to a new 96-well plate. The release of hemoglobin was monitored by measuring the absorbance at OD₄₀₅ every 5 minutes for 1 hour with a TECAN micro-plate reader. As controls, hRBCs were treated with PBS 1X as a negative control, 0.1% Triton X-100 as a positive control, and melittin (Sigma M2272) as a positive control. The percent of hemolysis was calculated based on the 100% release with 0.1% Triton X-100. Data were obtained in duplicates from two independent experiments.

2.5.24 Cell culture

J774A.1 macrophage cell line was cultured in DMEM supplemented with 10% Fetal Bovine Serum, 1% L-glutamine, 1% penicillin-streptomycin. The cells were grown at 37 °C under a controlled humidified atmosphere with 5% carbon dioxide. Cells were sub-cultured biweekly.

2.5.25 *In vitro* cell viability

Analysis of the cytotoxicity of compounds against J77A.1 macrophage cells was carried out using methylthiazolyldiphenyl-tetrazolium bromide (MTT, Sigma M2128).¹⁴⁷ Macrophage

cells were seeded into a 96-well plate at 100 μ L and density of 20,000 or 25,000 cells/well in complete DMEM media. Cells were incubated overnight (60-80% confluency) at 37 °C under a 5% CO₂ atmosphere. The spent media was aspirated, and the cells were washed with 100 μ L PBS. Then, 100 μ L of peptide treatment made in DMEM growth media was added to the cells at concentrations ranging from 1.25- 40 μ M. After 9 hours of incubation under 5% CO₂ atmosphere at 37 °C, the treatment media was aspirated and the cells were washed with 100 μ L of PBS. Then, 100 μ L of fresh complete DMEM media was added, followed by the addition of 10 μ L of MTT solution (5 mg /mL MTT in PBS) to each well. The cells were incubated for an additional 2 hours. The MTT solution was aspirated and 100 μ L of DMSO was added to each well and agitated for 5 minutes to dissolve the formed formazan crystals. Samples were run in duplicates and each experiment was repeated at least twice. The mean absorbance for each sample was measured and recorded at 590 nm on a TECAN micro-plate reader. Percent viability was determined by taking ratio of treated cells to untreated cells.

2.5.26 Mammalian cell flow cytometry

Macrophage J774A.1 cells were plated in 500 μ L of complete DMEM media at 150,000 cells/well in round bottom tubes (BD Biosciences) and incubated overnight at 37°C under 5% CO₂ atmosphere. The cells were centrifuged at 1200 rpm for 7 minutes at 4 °C, and the spent media was aspirated using a Pasteur pipet and vacuum filter. The cells were then treated with 300 μ L of peptide at concentrations ranging from 2.5- 10 μ M prepared in complete DMEM media and were incubated for the desired incubation time at 37°C. Upon completion of the incubation period, the cells were centrifuged, and the spent media was aspirated. The cells were then resuspended in 300 μ L of cold PBS 1X and the fluorescence of the cells was measured using an Accuri flow cytometer (BD Biosciences). All samples were run in duplicate, and each experiment was repeated at least twice. The mean arbitrary fluorescence values of gated cells were measured and recorded upon excitation of the fluorophore, fluorescein, using the 488 nm laser. For each experiment, a negative control of cells that were not incubated with fluorescent compound (DMEM only) was also analyzed.

Flow cytometry with Trypan Blue

After incubation with compounds, cells were centrifuged, and media aspirated as described in protocol above. The cells were then resuspended in 300 μ L of 1 mg/mL trypan blue in PBS 1X without additional washing. Fluorescence was measured and analyzed as described.

2.5.27 Confocal microscopy in live cells

High resolution imaging of mammalian cells and subcellular localization of the peptides was performed in the J774A.1 macrophage cell line using a Nikon A1R Multiphoton inverted confocal microscope with a 60X oil objective with 405 nm (blue), 488 nm (green), and 572 nm (red) 488 nm laser lines employed. J774A.1 cells were seeded into a 4-well Ibidi μ -slide 4-well culture chamber (Cat. No. 80426) at a density of 125,000 cells/well in 500 μ L of complete DMEM media. The cells were grown overnight (60% confluency) in a humidified 5% CO₂ atmosphere at 37°C. Cells were washed with 500 μ L PBS, then, 400 μ L of peptide treatment prepared in growth media was added to the wells at desired concentrations. The cells were incubated for 1 hour or 3 hours in the presence of the peptide treatment. Following incubation, the cells were washed with 500 μ L of PBS. Then, 400 μ L of either 100 nM Mitotracker, 1 μ M Hoescht 33342 solution prepared in growth media or 300 nM LysoTracker, 1 μ M Hoescht 33342 prepared in growth media were added to the wells and the cells were incubated for 30 minutes. After incubation, the cells were washed with 500 μ L of PBS. Next, 500 μ L of complete DMEM media was added to each well prior to imaging. Live cells were then imaged as above.

CHAPTER 3. CONTINUING THE PURSUIT OF AN IMPROVED DUAL-ANTIBIOTIC WITH A LINEZOLID-CAPH CONJUGATE

3.1 Introduction

Linezolid is a synthetic antibiotic grouped into the newer class of oxazolidinone antibiotics. It has been approved by the U.S. Food and Drug Administration (FDA) for use against resistant Gram-positive cocci, such as vancomycin-resistant *Enterococcus*, methicillin-resistant *S. aureus*, and penicillin-resistant *Pneumococci*. Linezolid disrupts bacterial growth by targeting protein synthesis machinery and inhibiting the initiation process. The target site of inhibition occurs earlier in the initiation process than other protein synthesis inhibitors, such as chloramphenicol, clindamycin, aminoglycosides, and macrolides.¹⁷⁷⁻¹⁷⁸ Given linezolid has remarkable pharmacokinetic properties with almost 100% oral bioavailability, it is an ideal therapy against numerous Gram-positive infections located throughout the body.¹⁷⁹ Additionally, linezolid is equally active against methicillin-susceptible and -resistant staphylococci, against vancomycin-susceptible and VanA, VanB or VanC resistant enterococci, and against susceptible pneumococci or penicillin- and/or macrolide- resistant strains.¹⁸⁰⁻¹⁸² Advantageously, linezolid has equivalent mammalian intracellular activity as extracellular activity, although inferior *in cyto* killing to vancomycin derivatives, telavancin and ortavancin.¹⁸³ Given the disadvantages of vancomycin therapy detailed previously (chapter 2), linezolid is one of the best alternative therapies in treating Gram-positive infections.¹⁸⁴⁻¹⁸⁵

An estimated 10 million people fell ill with tuberculosis (TB) in 2019, with close to 1.5 million associated deaths.¹⁸⁶ Recently, linezolid has been included in the standard regimen of treatment for TB, multidrug-resistant tuberculosis (MDR TB), and extensively drug-resistant tuberculosis (XDR TB).¹⁸⁷⁻¹⁸⁸ Although linezolid is an effective second-line treatment for TB, the long-term usage in the TB treatment regimen carries considerable risk of toxicity and adverse effects.^{187, 189-190}

The promising outlook of combination therapy, observed in recent TB treatment, has inspired the treatment of other difficult infections through combination of linezolid and a secondary drug.¹⁹¹ In combination with rifampicin, linezolid has shown an additive activity for susceptible *S. aureus* strains and inhibited rifampicin-resistant variants; and this combination appeared to be an attractive therapy against methicillin-resistant *S. aureus* strains.¹⁹² Although

there are increases in bactericidal activity and reduction in the emergence of drug-resistant mutants of linezolid in combination with rifampicin, some observations have noted synergism, indifference, or slight antagonism between the two drugs.¹⁹²⁻¹⁹⁵ Antagonism among combinatorial drugs is not ideal as it typically results in the reduction in the effectiveness of one or more of the drugs; however, information gained from antagonistic drugs may aid in developing strategies to hinder the onset of resistance.¹⁹⁶

Resistance to linezolid has remained infrequent since its approval and clinical use.¹⁹⁷ Because linezolid inhibits protein synthesis by binding to the peptidyl transferase center on the ribosome, almost all known resistance mechanisms involve mutations to the linezolid binding site.¹⁹⁸⁻¹⁹⁹ Although linezolid is currently the only drug approved to treat vancomycin-resistant enterococci, the threat of linezolid- and vancomycin-resistant *Enterococcus faecium* (LR-VRE) represents a major challenge for infection control.²⁰⁰⁻²⁰¹ Antibiotic resistance among *Staphylococcus epidermidis* is also an emerging problem with linezolid-resistant *S. epidermidis* (LRSE) strains being detected in Europe since 2014.²⁰² Such isolates have limited or no therapeutic option.

In order to mitigate these disadvantages in linezolid therapy, we utilized our antibiotic-CAPH conjugate design to yield **LnzP14** and its fluorescent equivalent, **FLnzP14** (**Figure 3.1 A**). We have developed an alternate mean of attaching the tether to linezolid, which only has an amide group available for conjugation. An acyl-amide strategy was undertaken, with a disulfide within the tether. In a reducing environment, the resulting linezolid thiol should rearrange to regenerate linezolid (**Figure 3.1 B**). Through this strategy using a reducible disulfide linker, we expect to deliver a combination therapy with superior activity against susceptible, resistant, and intracellular bacteria, including enterococcus, staphylococcus, and mycobacteria species. In arming this conjugate with two effective antibiotics, there is an additional potential to lower the dosage of linezolid and reduce toxicity accompanied by long-term usage. Lastly, this dual-antibiotic strategy may broaden the activity of linezolid to Gram-negative bacteria, that have intrinsic resistance to linezolid by efflux pumps.²⁰³⁻²⁰⁴ Through conjugation with a cell-penetrating peptide, this conjugate may possibly evade efflux pump mechanisms and effectively accumulate within bacterial cells to therapeutic concentrations.

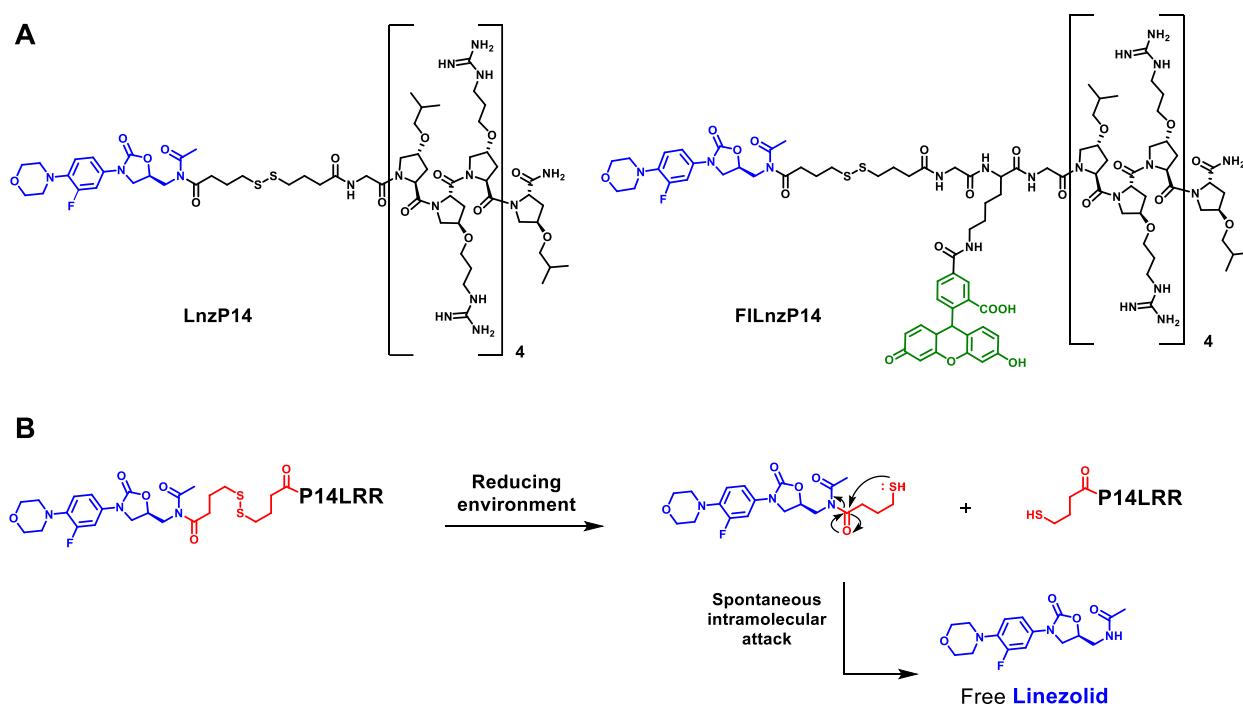
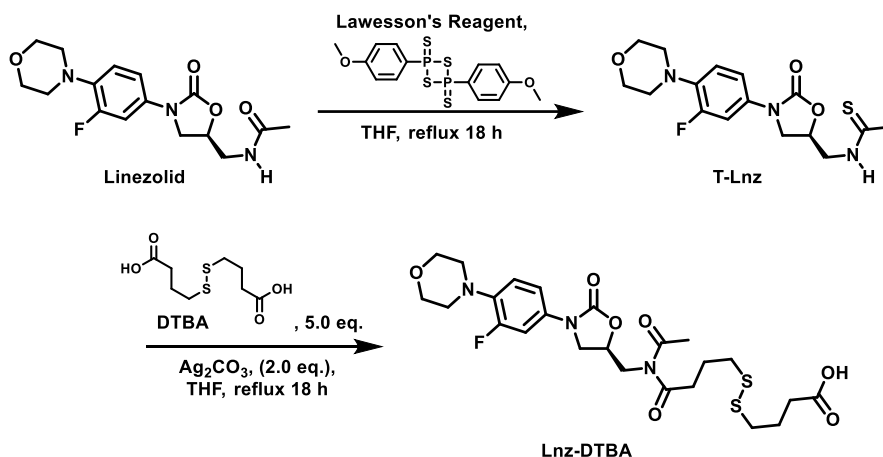


Figure 3.1. (A) Structures of linezolid-CAPHs conjugates. (B) Reduction of the **LnzP14** conjugate and regeneration of free linezolid.

3.2 Results and Discussion

3.2.1 Synthesis of thioamide linezolid (**T-Lnz**) and dithiobutyric acid linezolid (**Lnz-DTBA**)

The synthesis of the acylated linezolid derivative was carried out in the Chmielewski group by the visiting scholar, Paulo Pitasse Santos. In order to begin synthesis of the linezolid-CAPH conjugates, linezolid was first activated according to literature procedure.²⁰⁵ Using Lawesson's reagent, the amide moiety in linezolid was transformed to a thioamide, **T-Lnz**, in an excellent 91% yield (**Scheme 3.1**). Next, **Lnz-DTBA** was synthesized by an Ag^{I} -promoted coupling reaction of thioamides and carboxylic acids. The **T-Lnz** intermediate and 4,4'-bis-dithiodibutyric acid (DTBA) were refluxed in the presence of silver (I) carbonate to afford the linezolid imide, **Lnz-DTBA**, in good yield, 69% (**Scheme 3.1**).

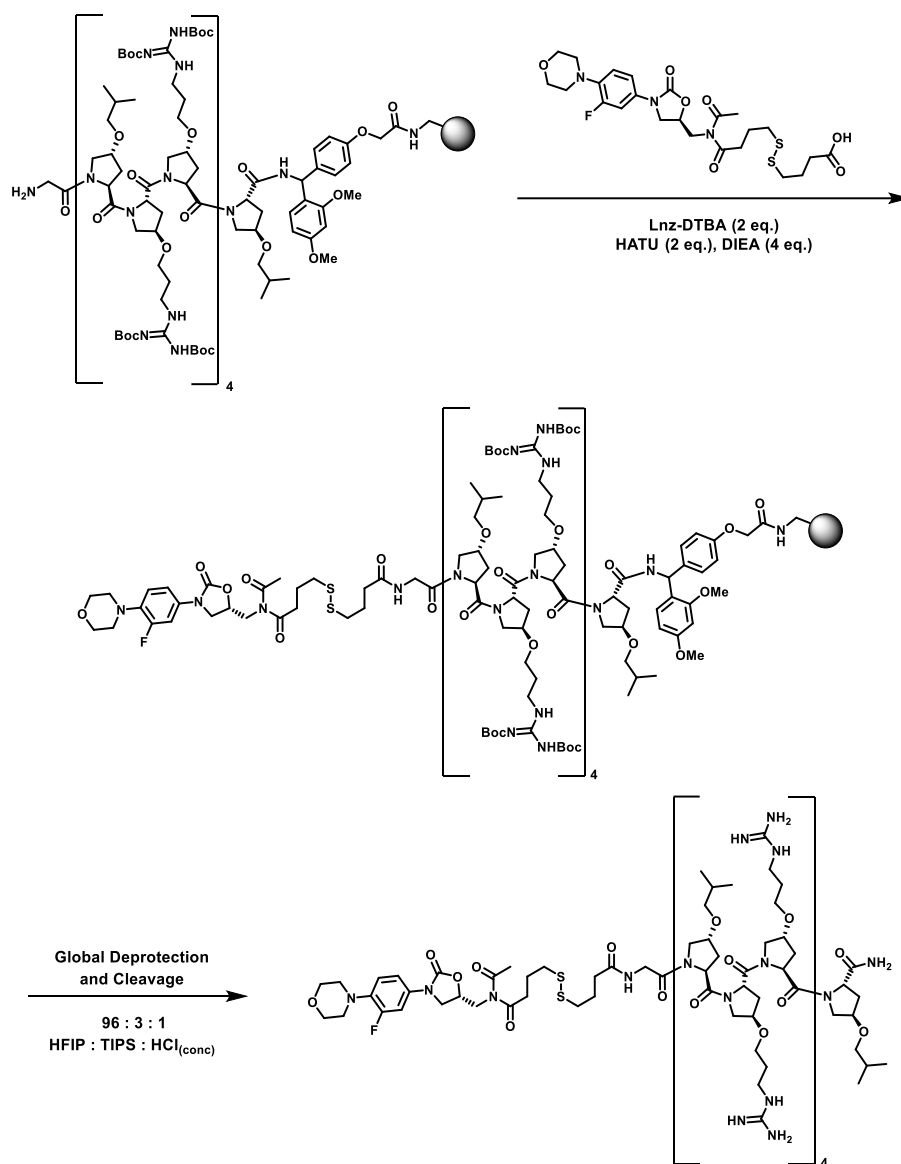


Scheme 3.1. Synthesis of tether-activated **Lnz-DTBA** as carried out by Paulo Pitasse Santos.

3.2.2 Synthesis of resin-bound conjugates **LnzP14** and **FILnzP14**

Peptide synthesis was carried out by Paulo Pitasse Santos and proceeded as previously described (Chapter 2, **Scheme 2.2**). Briefly, solid phase peptide synthesis (SPPS) afforded resin-bound **P14LRR** from Rink amide ChemMatrix resin (0.045 mmol/g). Amino acids were activated with HATU and DIEA in DMF and coupled to resin-bound peptide. Fmoc was deprotected with 30% piperidine in DMF. Once **P14LRR** was synthesized, the terminal Fmoc was removed and HATU-activated **Lnz-DTBA** was coupled to afford **LnzP14** on resin. Global deprotection and cleavage from the resin was performed using a cocktail of 96:3:1 HFIP:TIPS:HCl_{conc.} (final HCl concentration of 0.12 N) to yield crude conjugate (**Scheme 3.2**). These conditions were utilized to avoid hydrolysis of the imide moiety between linezolid and tether. **LnzP14** was purified using RP-HPLC and the desired product was confirmed using MALDI-ToF mass spectrometry.

The fluorescent variant, **FILnzP14**, was synthesized similarly with minor modifications by Paulo. Fmoc-Lys(Mtt)-OH and Fmoc-Gly-OH were coupled to the **P14LRR** on resin, the Mtt group was selectively deprotected using 30% hexafluoroisopropanol (HFIP) in DCM, followed by the coupling of fluorescein-NHS ester to the free primary amine. The Fmoc on the terminal glycine was removed and HATU-activated **Lnz-DTBA** was coupled to afford **FILnzP14** on resin. Global deprotection and cleavage from the resin was performed as described above. **FILnzP14** was purified using RP-HPLC and the desired product was confirmed using MALDI-ToF mass spectrometry.



Scheme 3.2. Synthesis of **LnzP14** as carried out by Paulo Pitasse Santos.

3.2.3 Minimum inhibitory concentrations against *E. coli*

To begin our investigation into the activity of the **LnzP14** conjugates, we determined the minimum inhibitory concentrations using a microbroth dilution assay. Gram-negative *E. coli* ATCC 25922 bacteria were grown and incubated at a range of concentrations from 0.125- 16 μ M with the melittin control,²⁰⁶ conjugate, and 1:1 mixture of peptide and linezolid for 20 hours. The minimum inhibitory concentration (MIC) was determined using a microplate reader as the lowest concentration of drug at which no growth was observed (**Table 3.1**). Both conjugates, **LnzP14** and **FILnzP14**, exhibited the same moderate antibacterial activity with MIC values of 4 μ M. This is

an excellent increase in bactericidal activity as linezolid shows poor activity against *E. coli* ATCC 25922 with MIC greater than 256 µg/mL and other strains of *E. coli* greater than 128 µg/mL.²⁰⁷⁻²⁰⁸ Additionally, the 1:1 mixture of CAPH and linezolid provided the same 4 µM MIC as the conjugates, with a slight improvement in activity compared to the CAPH alone. These data support the conjugation of CAPH to linezolid can broaden the scope of activity for linezolid, a Gram-positive antibiotic, to Gram-negative bacteria like *E. coli*.

Table 3.1. Minimum inhibitory concentrations (MICs) of linezolid conjugates against pathogenic bacteria isolates using the microbroth dilution assay after 20 h of treatment. Values reported in µM.

MIC (µM) ^a	
Melittin	2
LnzP14	4
FlLnzP14	4
FIP14LRR	8
1:1 FIP14LRR: Lnz	4

^aconducted by Paulo Pitasse Santos

3.2.4 Bacterial lysis study using β-galactosidase assay

There are different mechanisms that antimicrobial peptides use to achieve antimicrobial activity, including targeting vital bacterial proteins or through lysing the bacterial membrane.²⁰⁹⁻²¹⁰ The P14LRR CAPH does not lyse bacteria and linezolid targets intracellular machinery as previously described. However, we wish to determine whether **LnzP14** targets the bacterial membrane in a lytic mechanism. Once the MICs were determined, we proceeded with the previously described β-galactosidase (β-Gal) assay to determine if **LnzP14** demonstrated lytic behavior in *E. coli*. At 2-times and 4-times the MIC value, **LnzP14** did not show significant lysis compared to the lytic melittin control (**Figure 3.2**). We can conclude that even at concentrations well above MIC values, **LnzP14** does not show bacterial membrane lysis as its bactericidal mechanism of action.

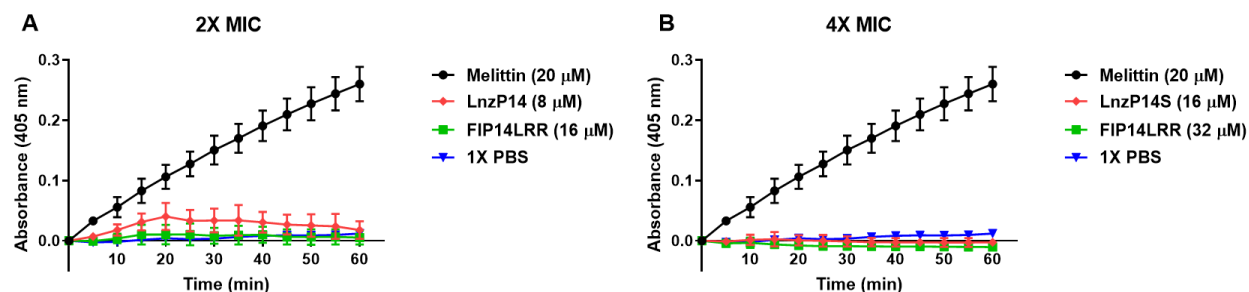


Figure 3.2. Monitoring the release of β -galactosidase over 1 hour from *E. coli* ATCC 25922 after 1 hour incubation with compounds. Positive control Melittin (20 μ M) and conjugate at (A) 2X and (B) 4X MIC

3.2.5 Cell toxicity studies

With the **LnzP14** and **FILnzP14** conjugates displaying bactericidal activity, we next needed to probe toxicity of the conjugates toward human red blood cells (hRBCs) and mammalian macrophage cells (J774A.1). We used melittin, a control AMP that demonstrates a lytic mechanism of action, and **P14LRR** as a standard for non-lytic behavior. The compounds and conjugate were incubated the with red blood cells for 1 hour. If the compounds lysed the red blood cells, heme would be released, and a signal would be observed on the spectrophotometer proportionally to the amount of lysis. Our results confirm that the **LnzP14** conjugate does not show significant lysis of red blood cells, analogous to linezolid or **P14LRR**, even at concentrations up to 40 μ M (**Figure 3.3 A**). Therefore, it can be concluded that the **LnzP14** conjugate may not use a lytic mode of action that many AMPs display.

We chose to explore the mammalian cell toxicity of our conjugates with J774A.1 macrophage cells because phagocytes can engulf and harbor bacteria, and we wish to carry out conjugate internalization and *in cyto* bacterial clearance experiments on the same cell lines. Therefore, **LnzP14** and **FILnzP14**, and **FIP14LRR** CAPH were incubated with mammalian cells for 9 and 24 hours at concentrations ranging from 0.5 to 32 μ M, as carried out by Paulo Pitasse Santos. Through the addition of 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT), we could monitor the metabolic activity of the cells in converting MTT to formazan, a colorimetric indicator of cell viability that can be quantified on a spectrophotometer. After 9 hours, **LnzP14** did not show any visible cytotoxicity at concentrations up to 32 μ M (data not shown). Our 24 hour results indicate that **LnzP14** and fluorescent variant, **FILnzP14**, show similar toxicity with greater than 75% viability at 2 μ M concentration (**Figure 3.3 B**). **FIP14LRR** shows less

toxicity with the viability decreasing at concentrations 8 μM and greater. Unfortunately, these results indicate that the cell viability of macrophage cells begins to decrease at the corresponding minimum inhibitory concentrations (MICs) of the conjugates. However, the MICs must be determined for a range of bacteria, including *in cyto* MICs, as only *in vitro* *E. coli* data has been presented.

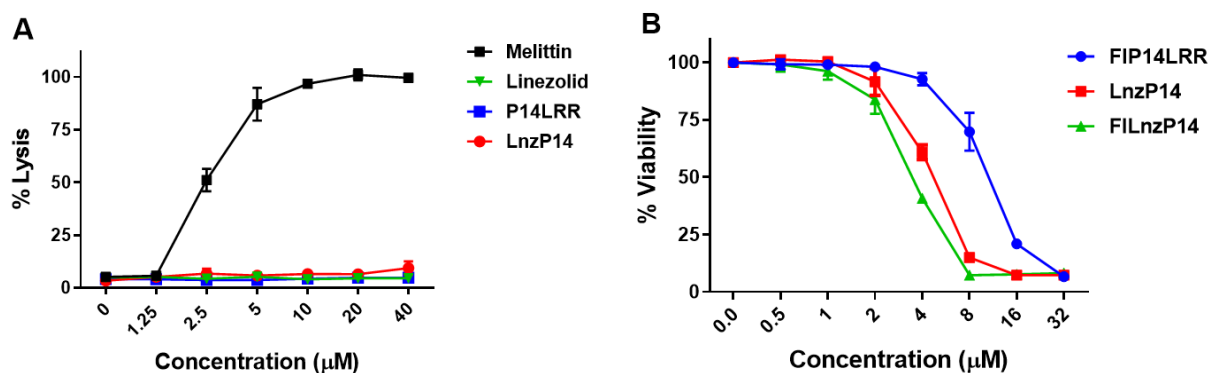
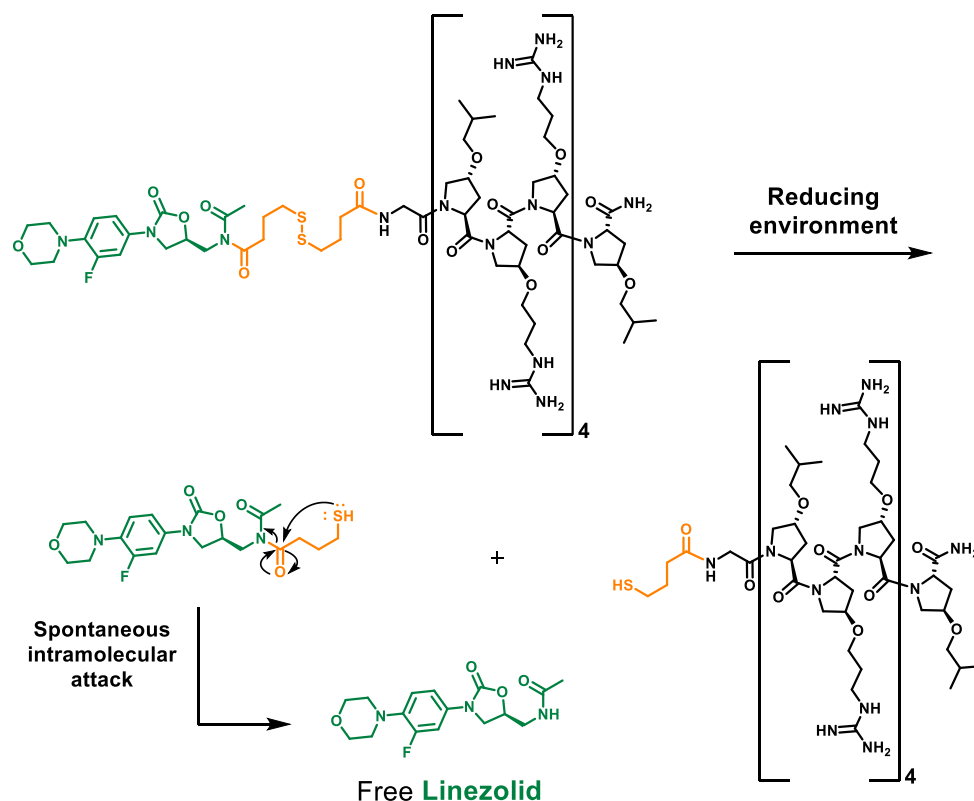


Figure 3.3. (A) Hemolysis assay measuring the release of hemoglobin from hRBC. Melittin was used as a positive control and values were normalized against 0.1% triton X-100. (B) Cell toxicity measuring the cell viability using an MTT assay after 24 h treatment of compounds, carried out by Paulo Pitasse Santos.

3.2.6 Release kinetics of linezolid from the LnzP14 conjugate

The inclusion of the disulfide linker between linezolid and **P14LRR** is a notable characteristic in the conjugate design. In the internal environment of the cell, the reductant glutathione is present in concentrations ranging from 1- 10 mM.²¹¹ Once penetrating into the mammalian cell intracellular environment, the disulfide linkage of **LnzP14** can reduce to release **P14LRR-SH** and linezolid-SH (**Scheme 3.3**). After the disulfide is reduced, the free thiol on linezolid can intramolecularly attack the imide and release free linezolid. To evaluate the half-life ($t_{1/2}$) of the conjugate, we incubated **LnzP14** with two different reducing agents, dithiothreitol (DTT, 10 mM) and glutathione (GSH, 5 mM). Linezolid release from the conjugate was monitored using UPLC-MS for up to 90 minutes in DTT (carried out by Paulo Pitasse Santos) or 18 hours in GSH (**Figure 3.4**).



Scheme 3.3. Reduction of **LnzP14** and release of linezolid and **P14LRR-SH**.

For the DTT reduction of **LnzP14** and release of free linezolid, the half-life of the conjugate was determined to be 2.5 minutes (**Figure 3.4 A**). Ultimately, 82% of linezolid was released from the initial concentration of **LnzP14**. To examine this release under more physiological conditions, we changed reducing agents to GSH and lowered its concentration to reflect the typical intracellular environment more closely. Using 5 mM GSH, the half-life of the conjugate was calculated to be 2.5 hours and about 60% of linezolid was released from the initial concentration of **LnzP14** (**Figure 3.4 B**). The incomplete release of linezolid could be explained mechanistically. Once the conjugate is reduced to produce **P14LRR-SH** and Lnz-SH, the free thiol on linezolid may intramolecularly attack the acetyl position of the imide. Although unfavored, this rearrangement would not result in the release of free linezolid. Although 100% release of linezolid was not observed, these studies indicate that **LnzP14** is responsive to reducing environments comparable to *in cyto*.

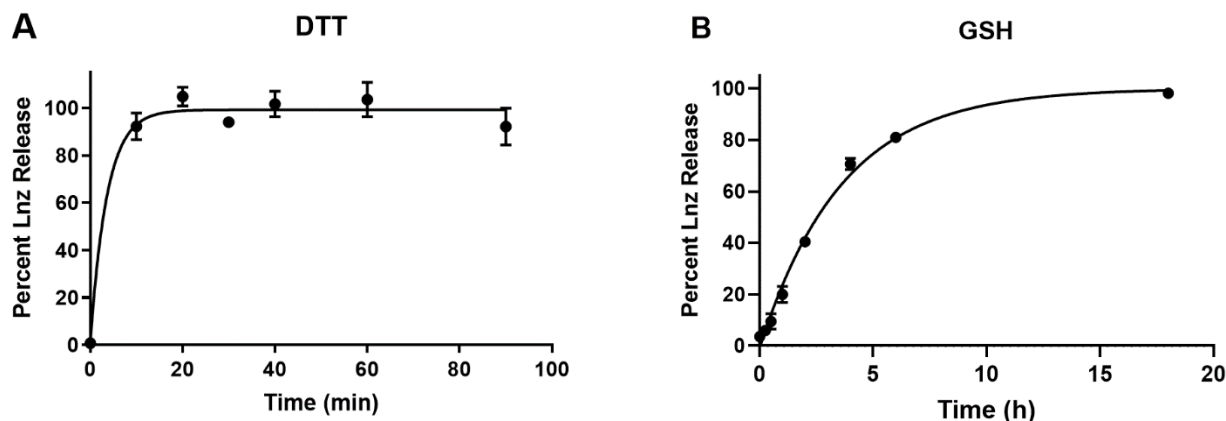


Figure 3.4. (A) Adjusted reduction and release of linezolid from **LnzP14** with 10 mM DTT in PBS buffer, as carried out by Paulo Pitasse Santos. (B) Adjusted reduction and release of linezolid from **LnzP14** with 5 mM GSH in PBS buffer.

3.2.7 Flow cytometry to quantify cellular internalization of **FILnzP14**

After concluding that the **LnzP14** conjugate can reduce within the intracellular environment, we sought to quantify intracellular accumulation of the conjugate in macrophage cells. Using J774A.1 macrophage cells, 1.25 and 2.5 μM concentrations of **FILnzP14** and **FIP14LRR** were incubated with cells for a period of 1 hour and 3 hours. These concentrations were chosen because cell viability was near 100%. After incubation, the cells were sorted by flow cytometry and cellular fluorescence was quantified. Trypan blue was used to quench surface-bound fluorescence, and quantify internal fluorescence related to the internalization of fluorescent peptide and conjugate.

First, the accumulation of both **FIP14LRR** and **FILnzP14** is concentration and time dependent, as more fluorescence is observed after longer incubation times and higher concentrations (**Figure 3.5**). Next, **FILnzP14** has significantly higher accumulation than **FIP14LRR** at both concentrations and time points (**Table 3.2**). At 1.25 μM , **FILnzP14** has 3-4 times more fluorescence than **FIP14LRR**; and at 2.5 μM , **FILnzP14** has 6-10 times more fluorescence than **FIP14LRR**. Additionally, 1 hour incubation times showed that there is a minimal level of the peptides associated with the membrane as visualized by trypan blue fluorescence quenching. About 83% and 100% of fluorescence for 1.25 and 2.5 μM **FILnzP14** correspond to internal fluorescence after 1 hour, while about 60% of the fluorescence for

FIP14LRR is internal at both concentrations after 1 hour. Both these results conclude **FILnzP14** is better at accumulating intracellularly than CAPH alone.

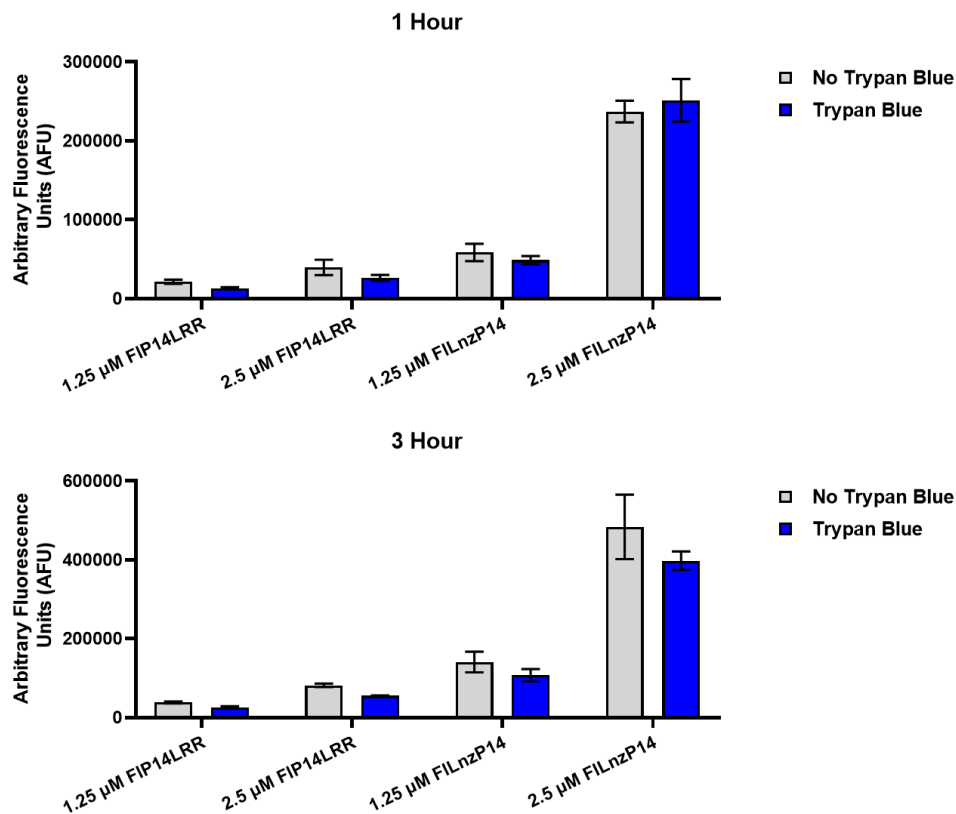


Figure 3.5. Cellular fluorescence of J774A.1 macrophage cells after 1- and 3-hour incubations. Trypan blue was used to quench surface bound fluorescence and selectively quantify internal fluorescence.

After 3 hours of incubation with CAPH and conjugate, we observe somewhat more surface-bound conjugate (**Figure 3.5**). About 80% of fluorescence for both 1.25 and 2.5 μM **FILnzP14** corresponds to internal fluorescence after 3 hours, while about 70% of the fluorescence for **FIP14LRR** is internal at both concentrations after 3 hours. In all cases, the **FILnzP14** conjugate showed enhanced cellular uptake and enhanced internalization of conjugate compared to **FIP14LRR**. These results indicate that surface binding is not an obstacle for **FILnzP14**. Dietsche and Chmielewski coworkers have shown that adding hydrophobic moieties to the N-terminus of the **P14LRR** peptide results in enhanced uptake.⁴⁴ Our results reinforce that conclusion, translating to **FILnzP14** showing enhanced penetration over the CAPH, **FIP14LRR**.

Table 3.2. Comparison of cellular uptake of **FILnzP14** to **FIP14LRR** by flow cytometry fluorescence results.

Fluorescence Ratio of FILnzP14 to FIP14LRR				
Concentration (μ M)	1 hr no TB	1 hr TB	3 hr no TB	3 hr TB
1.25	3	4	4	4
2.5	6	10	6	7

3.2.8 Confocal microscopy to visualize cellular internalization of **FILnzP14**

Not only do therapies need to penetrate and accumulate within cells, but they also need to accumulate and localize in intracellular compartments where their target bacteria may reside. For instance, engulfed *S. aureus* can reside inside phagosomes and phagolysosomes where it can replicate and survive innate immune responses and drug therapies.¹⁰⁹ Based on previous results in which P14LRR was found to penetrate macrophage cells, we wished to study the cellular uptake and internal localization of the linezolid-CAPH conjugate. Therefore, **FILnzP14** was incubated with J774A.1 cells for 1- and 3-hours at a non-cytotoxic concentration, 2.5 μ M (**Figure 3.6, Figure 3.7**). Following incubation, we further labeled cells with an endosomal marker (Lysotracker red, 300 nM) or a mitochondrial marker (Mitotracker red, 100 nM). In this way, if there is colocalization of **LnzP14** with either red stain, a yellow-orange color is visualized in the merge of the red and green confocal laser channels.

After the 1-hour incubation, we observe the green **FILnzP14** fluorescence overlapping with red lysotracker stain. Therefore, the conjugate appears to primarily colocalize within endosomal compartments of the cell, and no visible localization with the mitochondria as was found with **FIP14LRR** (**Figure 3.6**).^{41-42, 45} After the 3-hour incubation, we observe similar green **FILnzP14** fluorescence overlapping with red lysotracker stain; however, we observe some faint overlapping with red mitotracker stain. The conjugate appears to primarily colocalize within endosomal compartments of the cell at 3 hours with some mitochondrial association (**Figure 3.7**). These results indicate that within the 1-3 hour time, **FILnzP14** remains within endosomal compartments of the cell, with minimal observable association with mitochondria and release into

the cytoplasm. This suggests that **LnzP14** may be most suitable to target bacteria that remain trapped in endosomal and lysosomal compartments of mammalian cells.

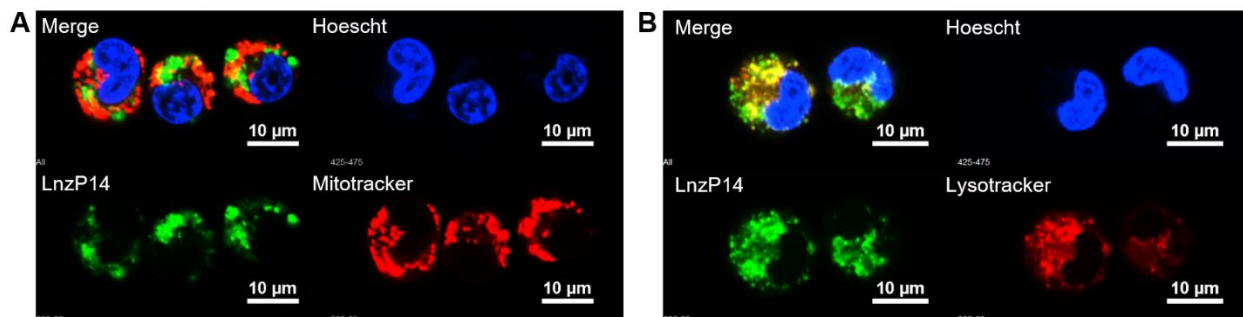


Figure 3.6. Confocal microscopy images of J774A.1 macrophage cells incubated with 2.5 μ M **FILnzP14** for 1 hr. (A) Merge (top left) and separate laser channels with Hoescht 33342 stain (blue), conjugate (green), and mitotracker mitochondrial stain (red). (B) Merge (top left) and separate laser channels with Hoescht 33342 nuclear stain (blue), conjugate (green), and lysotracker lysosomal (red). A yellow-orange color indicates colocalization of conjugate and selected red stain.

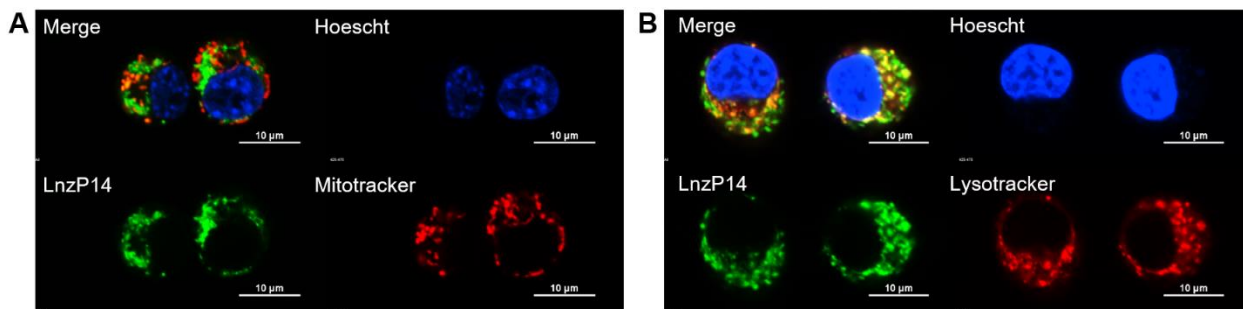


Figure 3.7. Confocal microscopy images of J774A.1 macrophage cells incubated with 2.5 μ M **FILnzP14** for 3 hr. (A) Merge (top left) and separate laser channels with Hoescht 33342 stain (blue), conjugate (green), and mitotracker mitochondrial stain (red). (B) Merge (top left) and separate laser channels with Hoescht 33342 nuclear stain (blue), conjugate (green), and lysotracker lysosomal (red). A yellow-orange color indicates colocalization of conjugate and selected red stain.

3.3 Conclusions

From these preliminary results, we have demonstrated that linezolid-CAPH conjugates have moderate activity against Gram-negative *E. coli* bacteria. This is a promising piece of data that indicate this conjugate may possess broad-spectrum activity, a characteristic linezolid does not show. The mode of action for this activity was determined to be non-lytic to bacterial membranes;

in addition, the **LnzP14** conjugate did not lyse human red blood cells. The conjugates demonstrate observable cytotoxicity to mammalian cells at concentrations 4 μ M and above, indicating that toxicity may be problematic. Remarkably, the linezolid-CAPH conjugate showed superior cellular uptake compared to **P14LRR** with increases in uptake over 3 times higher than CAPH. Overall, the increase in cellular accumulation and subcellular localization within endosomes makes **LnzP14** a promising potential therapy to reach and target endosomally-trapped bacteria such as *S. aureus* and *Mycobacteria*.

3.4 Future Directions

In order to sufficiently determine the usefulness of **LnzP14** as a potential antibiotic therapy, more experiments are needed. Additional experiments to determine antibacterial activity against a range of pathogens, including drug-susceptible, drug-resistant, Gram-positive, and Gram-negative are currently planned. Once a selection of bacteria has been narrowed to those that the conjugate has good activity against, intracellular clearance experiments should be conducted to determine if **LnzP14** can clear these pathogens *in cyto*. More insight into the mode of action that this conjugate utilized will also be performed.

3.5 Materials and Methods

3.5.1 Materials

Starting material for unnatural amino acids, Fmoc-protected natural amino acids, and coupling reagents were purchased from Chem Impex (Wood Dale, IL) or Ana Spec, Inc. (Fremont, CA). H-Rink Amide ChemMatrix resin for peptide synthesis was purchased from PCAS Biomatrix Inc. (Quebec, Canada). Vancomycin hydrochloride was purchased from Millipore Sigma (Burlington, MA). Sterile DMEM supplemented with L-glutamine and Penicillin-Streptomycin were purchased from VWR (Batavia, IL). Buffers (1X PBS) and 10% fetal bovine serum (FBS) used in cell culture were purchased from Corning Inc. (Corning, NY) and Atlanta Biologicals, Inc. (Research and Diagnostic Systems, Inc., Minneapolis, MN), respectively. All bacteria and cell lines for culture were purchased from ATCC (Manassas, VA). All other chemicals and reagents were purchased commercially and were used without further purification unless mentioned from Sigma Aldrich (St. Louis, MO), Alfa Aesar (Haverhill, MA) or Thermo Fischer (Waltham, MA).

Peptides and conjugates were purified using a Waters Delta Prep 4000 HPLC equipped with a Phenomenex C18 semi-preparative column. CAPHs characterization was performed using matrix associated laser desorption ionization-time of flight (MALDI-TOF) mass spectrometry using an Applied Biosystem Voyager-DE TM BioSpectrometry workstation and analytical RP-HPLC using Waters Delta Prep 4000 HPLC equipped with a C18 reverse phase analytical column (5 μ m, 4.6 mm x 250 mm; Phenomenex Luna). Cell uptake was measured using BD sciences Acurri and analyzed using BD software. Absorbance readings were obtained using microplate reader TECAN Infinite F PLEX. Confocal images were obtained using Nikon A1R-MP inverted confocal fluorescence microscope equipped with 60X oil objective. NIS Elements software was used to process images.

3.5.2 Synthesis of linezolid thioamide (T-Lnz)²¹²

The synthesis followed protocol previously described,²⁰⁵ with minor modifications. In a sealed reaction vial with stirrer under N₂ atmosphere, linezolid (680 mg, 2.0 mmol) and Lawesson's reagent (490 mg, 1.2 mmol) were refluxed in 10 mL dry THF for 18 h. The solvent was removed *in vacuo* and the reaction was purified by flash chromatography on silica with hexanes/ethyl acetate 1:3. The product **T-Lnz** was obtained as an amorphous solid weighing 640.3 mg (91% yield).

¹H NMR (400 MHz, CDCl₃): δ 8.55 (br d, J = 48 Hz, 1H), 7.38 (d, J = 12 Hz, 1H), 7.03 (d, J = 8 Hz, 1H), 6.90 (t, J = 8 Hz, 1H), 5.01-4.95 (m, 1H), 4.22-4.18 (m, 1H), 4.09 (dt, J = 8 Hz, J=24 Hz, 2H), 3.85-3.80 (m, 5H), 3.04 (br s, 4H), 2.58 (s, 3H)

¹³C NMR (100 MHz, CDCl₃): δ 203.75, 156.53, 154.55, 154.08, 136.67, 132.36, 118.76, 114.06, 107.75, 107.49, 71.15, 66.83, 50.82, 47.84, 47.67, 33.81

3.5.3 Synthesis of imide-linked Linezolid-tether (Lnz-DTBA)²¹²

The synthesis was adapted from previously reported procedure.²⁰⁵ In a sealed reaction tube with stirrer under N₂ atmosphere, T-Lnz (600 mg, 1.7 mmol) and 4,4'-bis-dithiodibutyric acid (DTBA, 2.43 g, 10.2 mmol) were dissolved in 30 mL dry THF. Silver (I) carbonate (936 mg, 3.4 mmol) was carefully added, then the reaction was kept at room temperature and vigorous stirring for 18 h. The solvent was removed under reduced pressure and the reaction was purified by FCC on silica

gel. Eluent DCM/EtOAc/AcOH was applied at gradient steps of 98:0:2 (400 mL), 73:25:2 (400 mL) and 48:50:2 (800 mL). The combined pure fractions were evaporated three times with toluene to remove excess acetic acid. The product **Lnz-DTBA** was obtained as an amorphous solid weighing 650 mg (69% yield).

ESI (MS) Expected mass: 558 Da. Observed mass: 558.2 Da

¹H NMR (400 MHz, CDCl₃): δ 9.04 (br s, 1H), 7.39 (dd, J = 4 Hz, J = 12 Hz, 1H), 7.07 (dd, J = 4 Hz, J = 8 Hz, 1H), 6.91 (t, J = 8 Hz, 1H), 4.80 (br d, J = 8 Hz, 1H), 4.16 (dd, J = 4 Hz, J = 16 Hz, 1H), 4.08 (t, J = 8, 1H), 3.92-3.84 (m, 5H), 3.70 (t, J = 8 Hz, 1H), 3.03 (t, J = 4 Hz, 4H), 2.89 (dt, J = 8 Hz, J = 4Hz, 2H), 2.71 (q, J = 8 Hz, 4H), 2.46 (s, 5H), 2.02 (dq, J = 24 Hz, J = 8 Hz, 4H)

¹³C NMR (100 MHz, CDCl₃): δ 175.54, 173.53, 156.57, 154.12, 153.81, 136.57, 132.60, 118.80, 113.89, 107.58, 107.32, 71.62, 66.80, 50.83, 48.32, 47.86, 37.57, 36.00, 32.30, 29.58, 26.49, 23.86

3.5.4 Synthesis of resin bound conjugates²¹²

Peptide-drug conjugates **LnzP14** and **FILnzP14** were synthesized using Fmoc-based solid phase peptide synthesis (SPPS). Briefly, Rink amide ChemMatrix resin (0.045 mmol/g) was employed as solid support. Successive coupling and Fmoc-deprotection reactions were carried alternately until the desired sequence was achieved. A washing routine with 5 mL of DMF (2X), DCM (2X), MeOH (2X), DCM (2X), DMF (2X) was carried in between each reaction step. For the coupling protocol, the amino acids (2.0 equiv.) were dissolved in 5 mL DMF along with HATU (2.0 equiv.) and DIEA (4.0 equiv.) for carboxyl activation. Piperidine/DMF 1:3 was used for Fmoc deprotection. The drug-tether moiety **Lnz-DTBA** (2.0 equiv.) was coupled after the last amino acid of the sequence through its free carboxylic acid to the peptide's N-terminus under the above coupling conditions. For the conjugate **FILnzP14**, Fmoc-Lys(Mtt)-OH and Fmoc-Gly-OH were coupled at the end of the peptide sequence. 30% hexafluoroisopropanol (HFIP) in DCM (4 × 7 mL for 30 min. each) was used for removing the Mtt side chain protecting group. The free amino side chain of lysine was then reacted with 5(6)-carboxyfluorescein succinimidyl ester (FI-NHS, 1.2 equiv.) in presence of DIEA (2.4 equiv.) in 5 mL DMF for 18 h. Then, the drug-tether moiety **Lnz-DTBA** (2.0 equiv.) was coupled to the N-terminus through its free carboxylic acid under the above coupling conditions.

3.5.5 Cleavage from resin and purification of LnzP14 and FILnzP14²¹²

For **LnzP14** and **FILnzP14**, a cleavage cocktail of HFIP : triisopropylsilane (TIPS) : HCl_(conc) 96 : 3 : 1 (0.12 N HCl) (8 mL) was added to the resin containing the conjugates and agitated for 3 h, as previously reported,²¹³ to prevent TFA induced hydrolysis of imide group. In each case, the solution was filtered through fritted glass into a 50 mL conical tube. The resin was rinsed with DCM (3 × 5 mL) and the filtrate collected into the same tube. The excess solvent was removed *in vacuo* and the products were precipitated in cold diethyl ether. The supernatant from the precipitation was decanted, and the precipitate dissolved in acetonitrile (ACN): water 1:3 prior to semipreparative RP-HPLC purification. For this step, a Phenomenex C₁₈ column was eluted with water with 0.1% TFA (solvent A) and ACN with 0.1% TFA (solvent B) at 12 mL/min under linear gradient of 25–65% B through 60 min and monitored by dual UV detection at 214 nm and 254 nm. The fractions containing the desired products were collected and lyophilized to yield **LnzP14** and **FILnzP14**.

The purity of peptide-drug conjugates was assessed by analytical RP-HPLC using a Phenomenex C₁₈ column was used with a binary solvent system composed by water with 0.1% TFA (solvent A) and ACN with 0.1% TFA (solvent B) at 1.2 mL/min under linear gradient of 20–65% B through 0–30 min, 65–95% B through 30–40 min, 95% B through 40–45 min and monitored by dual UV detection at 214 nm and 254 nm. Further characterization included the matrix assisted laser desorption ionization – time of flight mass spectrometry (MALDI-TOF-MS) using a Voyager DE Mass Spectrometer (Applied Biosystems) and α -cyano-4-hydroxycinnamic acid as support matrix. Analytical HPLC trace shown in **Figure A 13** for **LnzP14**. Analytical HPLC trace shown in **Figure A 14** for **FILnzP14**.

MALDI-TOF-MS for **LnzP14**: expected m/z 3155.8, observed m/z 3154.6

MALDI-TOF-MS for **FILnzP14**: expected mass 3700.9, observed mass 3698.2

3.5.6 Antibacterial activity against *Escherichia coli*²¹²

Escherichia coli (ATCC 25922) was streaked on an agar plate and incubated at 37°C for 18 h. Colonies of bacteria were recovered from agar and cultivated to mid-exponential phase in Tryptic Soy Broth (TSB) at 37 °C with shaking (5 colonies in 50 mL media). A 5 mL aliquot of the bacterial suspension was centrifuged for 5 min at 3000 rpm, and washed twice with Mueller

Hinton Broth (MHB). The pellet was re-suspended in MHB to a final optical density of 0.001 as measured by absorbance at 600 nm (OD_{600}). To a sterile transparent 96-well plate, 90 μ L of the bacteria suspension was added along with 10 μ L two-fold serially diluted drugs in water. Water was used as negative control and melittin as positive control. The plates were incubated for 20 h at 37 °C with shaking. The bacterial growth was measured at OD_{590} using a microplate reader (TECAN SpectraFluor Plus). The minimum inhibitory concentration (MIC) was determined as the lowest concentration of drug at which no growth was observed. Data was obtained in duplicates from at least two independent experiments.

3.5.7 An *E. coli* Beta-galactosidase release assay

Peptide solutions were prepared in deionized water to make concentrations at 2X and 4X the determined MIC values for peptides and conjugate against *E. coli* ATCC 25922. Five stationary colonies of *E. coli* ATCC 25922 were inoculated in 50 mL Mueller Hinton Broth (MHB) and grown at 37 °C for about 2 hours until the OD_{595} reached 0.1. Following, 5 mL of isopropyl- β -D thiogalactopyranoside (IPTG) (24 mg/10 mL PBS) was added to bacteria culture for a final concentration of 1 mM IPTG. The bacteria were induced until an OD_{595} between 0.3 and 0.5 was reached. A 10 mL aliquot of the bacteria in the IPTG/MHB media was removed and centrifuged at 3500 rpm for 5 min and spent media was decanted. The pelleted bacteria were then washed twice with 10 mL of fresh MHB and resuspended in 10 mL of MHB. The bacteria were plated in a 96-well plate at 90 μ L per well, followed by the addition of 10 μ L of the peptide solution. The bacteria were incubated with the peptide treatment for 1 hour. The 96-well plate was centrifuged for 5 min at 3500 rpm, and 80 μ L of the supernatant was transferred to a new 96-well plate. Then, 20 μ L of a freshly prepared 2-nitrophenyl- β -D-galactopyranoside (ONPG) (40 mg/10 mL of PBS) solution was added to each well for a final concentration of 0.8 mg/mL. The absorbances at 405 nm were read every 5 min over the course of 1 hour using a microplate reader to monitor the levels of o-nitrophenol formed. The leakage of β -galactosidase was calculated based on the absorbance value obtained for treated samples, with a time zero absorbance value subtracted from each time point. For all experiments melittin was used as a positive control, PBS pH 7.4 was used as a negative control, and **FIP14LRR** was included as a reference. Data was obtained in duplicate from at least two independent experiments.

3.5.8 Hemotoxicity of conjugates on human red blood cells

Fresh human red blood cells (hRBCs) (Innovative Research, cat # IWB3CPDA1) were collected by centrifugation at 2000 rpm for 5 min followed by washing three times with 5 mL PBS 1X, pH 7.4. The supernatant of the final wash was aspirated and 200 μ L of the concentrated cell suspension was added to 4.8 mL of PBS to make a 4% suspension (v/v). To a 96-well plate, was added 50 μ L of the hRBC solution, followed by 50 μ L of the peptide treatment prepared in PBS pH 7.4. The plate was incubated at 37 °C under 5% CO₂ for 1 h. The plate was subsequently centrifuged at 1200 rpm for 5 min at 4 °C. Next 75 μ L aliquots of the supernatants in each well were carefully transferred to a new 96-well plate. The absorbances at 405 nm were read on a TECAN Infinite F PLEX microplate reader. As controls, hRBCs were treated with PBS 1X as a negative control, 0.1% Triton X-100 as a positive control, and melittin (Sigma M2272) as a positive control. The percent of hemolysis was calculated based on the 100% release with 0.1% Triton X-100. Data were obtained in duplicate from two independent experiments.

3.5.9 Cell culture

J774A.1 macrophage cell line was cultured in DMEM supplemented with 10% Fetal Bovine Serum (FBS), 1% L-glutamine, 1% penicillin-streptomycin. The cells were grown at 37 °C under a controlled humidified atmosphere with 5% carbon dioxide. Cells were sub-cultured biweekly.

3.5.10 *In vitro* cytotoxic activity against J744A.1 macrophages²¹²

Analysis of the cytotoxicity of the conjugates against mouse macrophages (J774A.1) were carried out using methylthiazolyldiphenyl-tetrazolium bromide (MTT, Sigma M2128).¹⁴⁷ Cells were seeded in 96-well plates in 100 μ L complete Dulbecco's Modified Eagle Medium (DMEM) supplemented with 10% fetal bovine serum (FBS) at density of 20,000 cells/well and incubated at 37 °C in a 5% CO₂ atmosphere for 24 h. The adherent cells were washed with PBS and 100 μ L of fresh medium containing the conjugates two-fold serially diluted (0.5 – 32 μ M) was added. The cells were incubated for 24 h and the spent media was aspirated and replaced with 100 μ L fresh media, followed by the addition of 10 μ L of a 5 mg/mL solution of MTT in PBS pH 7.4. After a 90-minute incubation period, the media was removed and the MTT formazan crystals were dissolved in 100 μ L DMSO. The absorbance was then recorded at 590 nm using a microplate

reader. The percent cell viability was determined in comparison to the non-treated control. Data was obtained in duplicate from at least two independent experiments.

3.5.11 Analysis of reduction with dithiothreitol²¹²

To a previously degassed phosphate buffered saline (PBS) solution pH = 7.4, was added **LnzP14** (40 μ M) and internal standard quinine (10 μ M). The reaction was preheated at 37 °C and reducing agent dithiotreitol (DTT, 10 mM) was added to a final volume of 3 mL. The reaction was incubated at 37°C for 90 minutes and 200 μ L aliquots were collected at 0, 10, 20, 30, 40, 60 and 90 minutes and immediately frozen at -78 °C in a acetone/dry ice bath to halt the reduction reaction. The reagent concentrations for this experiment are expressed as the final concentration for the reaction. The samples were injected in a Waters UPLC-MS system, with a positive mode electrospray detector. The analysis was conducted using a C18 column and a binary solvent system composed by water with 0.1% formic acid (solvent A) and ACN with 0.1% formic acid (solvent B) at 0.5 mL/min under linear gradient of 5–50% B through 0–10 min. This experiment was performed in duplicate and for each analysis a calibration curve for linezolid ranging from 1.23 to 120 μ M in presence of 10 μ M quinine as internal standard was constructed at the same conditions. The area under the curve corresponding to linezolid was corrected with the internal standard and the concentration calculated based on the linear regression equation obtained from the calibration curve. The concentration and standard deviation values for each timepoint were analyzed in GraphPad Prism software and the data was analyzed using a non-linear one phase decay regression curve regression curve to generate a half-life.

3.5.12 Analysis of reduction with glutathione

LnzP14 (25 μ M) was incubated at 37 °C with 10 mM glutathione (GSH) in degassed phosphate buffered saline (PBS 1X, pH 7.4) containing 50 μ M 6-quinolinecarboxylic acid as the internal standard in 1.5 mL total volume. When monitoring linezolid release, an 100 μ L aliquot of the reaction mixture was taken at 0, 0.25, 0.5, 1, 2, 4, 6, and 18 hour time points and directly analyzed by a Waters UPLC-MS system, with a positive mode electrospray detector. The time points were analyzed using a C18 column consisting of 2-50% acetonitrile in water with 0.1% formic acid (0.5 ml/min, column temperature of 40 °C). The peaks corresponding to the m/z for

quinolinecarboxylic acid (+1: 174), linezolid (+1: 338), and P14LRR-SH (+5: 545) were detected and extracted using MassLynx software. This experiment was performed in duplicate and for each analysis a calibration curve for linezolid ranging from 1 to 32 μM in presence of 50 μM 6-quinolinecarboxylic acid as internal standard was constructed at the same conditions. The area under the curve corresponding to linezolid was corrected with the internal standard and the concentration calculated based on the linear regression equation obtained from the calibration curve. The percentage of release vs. time was fitted using GraphPad Prism software to generate a half-life using a non-linear one phase decay regression curve.

3.5.13 Flow cytometry

Macrophage J774A.1 cells were plated in 500 μL of complete DMEM media at 150,000 cells/well in round bottom tubes (BD Biosciences) and incubated overnight at 37°C under 5% CO_2 atmosphere. The cells were centrifuged at 1200 rpm for 7 minutes at 4 °C, and the spent media was aspirated using a Pasteur pipet and vacuum filter. The cells were then treated with 300 μL of conjugates and peptide at concentrations ranging from 2.5- 10 μM prepared in complete DMEM media and were incubated for the desired incubation time at 37°C. Upon completion of the incubation period, the cells were centrifuged at 1200 rpm for 7 minutes at 4 °C, and the spent media was aspirated. The cells were then resuspended in 300 μL of cold PBS pH 7.4 and the fluorescence of the cells was measured using an Accuri flow cytometer (BD Biosciences). All samples were run in duplicate, and each experiment was repeated at least twice. The mean arbitrary fluorescence values of cells were measured and recorded upon excitation of the fluorophore, fluorescein, using a 488 nm laser. For each experiment, a negative control of cells that were incubated with DMEM was also analyzed.

Flow cytometry with Trypan Blue

After incubation with compounds, cells were centrifuged, and media aspirated as described in protocol above. The cells were then resuspended in 300 μL of 1 mg/mL trypan blue in PBS 1X without additional washing. Fluorescence was measured and analyzed as described.

3.5.14 Confocal microscopy in live mammalian cells

High resolution imaging of mammalian cells and subcellular localization of the peptides was performed in J774A.1 macrophage cell line using a Nikon A1R Multiphoton inverted confocal microscope with a 60X oil objective and 405 nm (blue), 488 nm (green), and 572 nm (red) 488 nm lasers employed. J774A.1 cells were seeded into 4-well μ -slide culture chamber (Ibidi, Cat. No. 80426) at a density of 125,000 cells/well with 500 μ L of complete DMEM media. The cells were grown 20 hours (60% confluency) in a humidified 5% CO₂ atmosphere at 37°C. Cells were washed with 500 μ L PBS, then, 400 μ L of conjugate concentration prepared in DMEM was added to the wells at desired concentrations. The cells were incubated for 1 hour or 3 hours in the presence of the conjugate. Following incubation, the adherent cells were washed with 500 μ L of PBS and aspirated. Then, 400 μ L of either 100 nM Mitotracker mitochondrial stain, 1 μ M Hoescht nuclear stain solution prepared in DMEM or 300 nM LysoTracker lysosome stain, 1 μ M Hoescht nuclear stain prepared in DMEM were added to the wells and the cells were incubated for 30 minutes. After incubation, the adherent cells were washed with 500 μ L of PBS and aspirated. Next, 500 μ L of complete DMEM media was added to each well prior to imaging. Live cells were then imaged using Nikon A1R-MP inverted confocal fluorescence microscope equipped with 60X oil objective. NIS Elements software was used to process images.

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APPENDIX A. ADDITIONAL DATA

Analytical HPLC Spectra

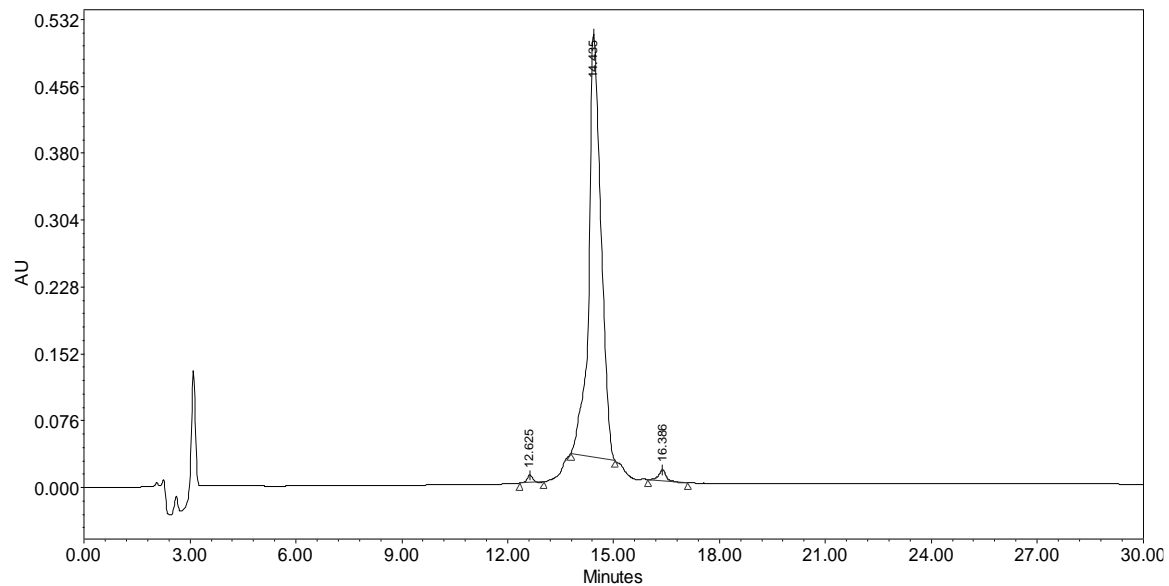


Figure A 1. Analytical purity HPLC spectrum of **P14LRR-SH**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 25-65% acetonitrile in water, 0.1% TFA. Product is 97% pure.

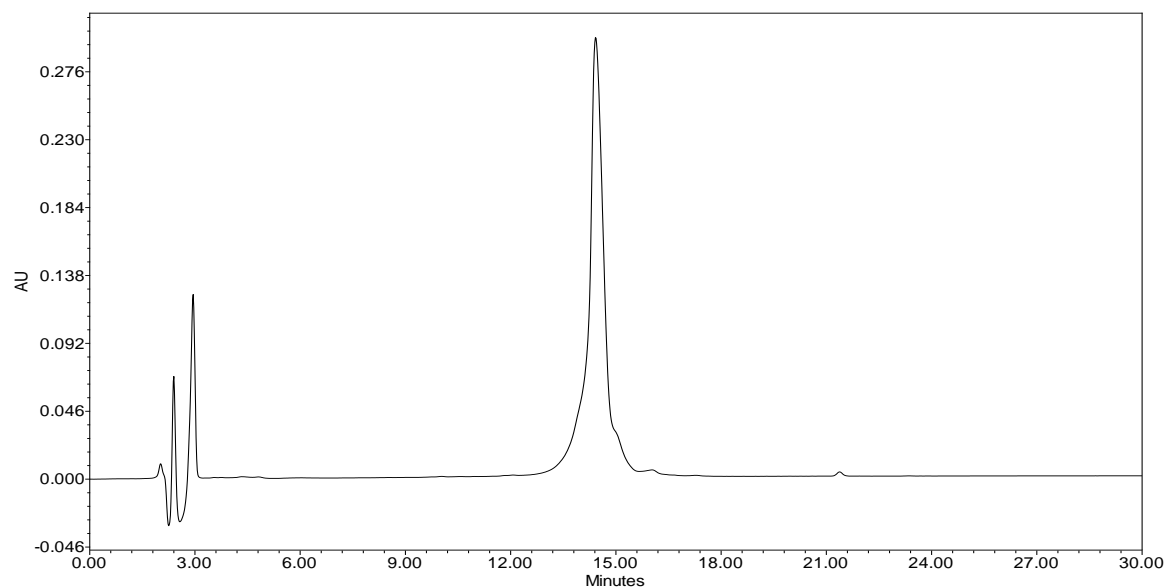


Figure A 2. Analytical purity HPLC spectrum of **FIP14LRR-SH**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 30-60% acetonitrile in water, 0.1% TFA. Product is 99% pure.

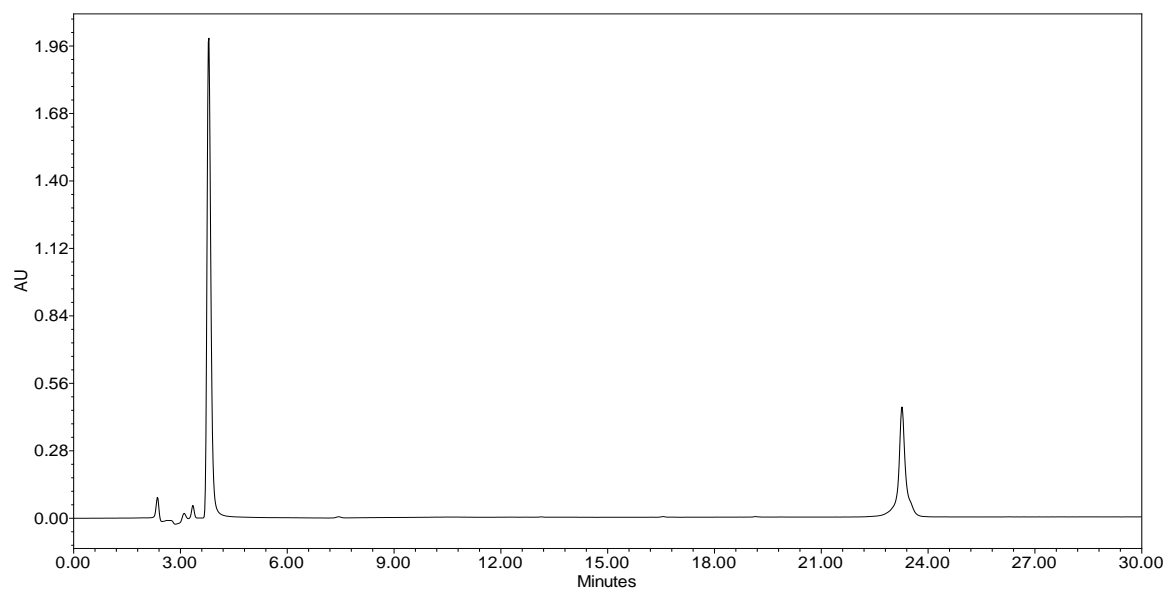


Figure A 3. Analytical purity HPLC spectrum of **VanP14S**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 5-60% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

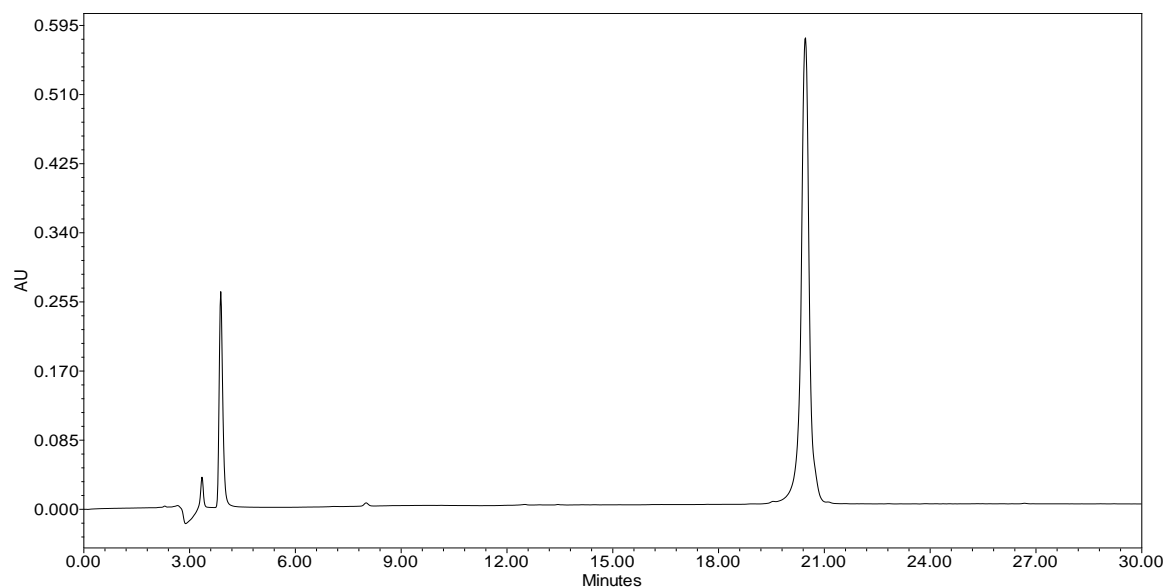


Figure A 4. Analytical purity HPLC spectrum of **VanMalP14**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 5-70% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

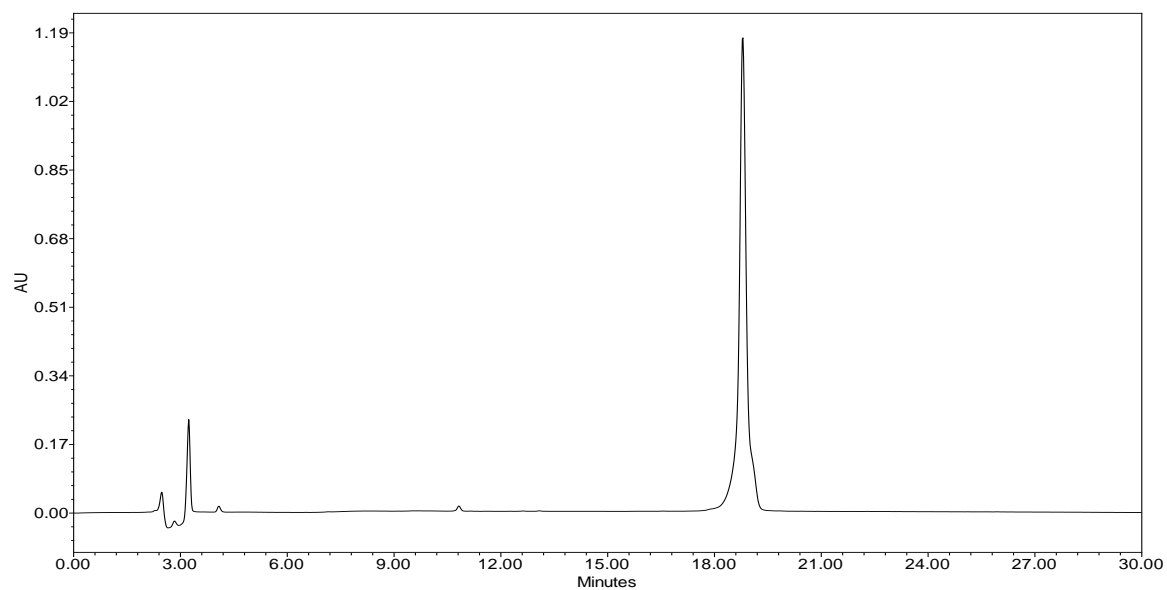


Figure A 5. Analytical purity HPLC spectrum of **FIVanP14S**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 10-80% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

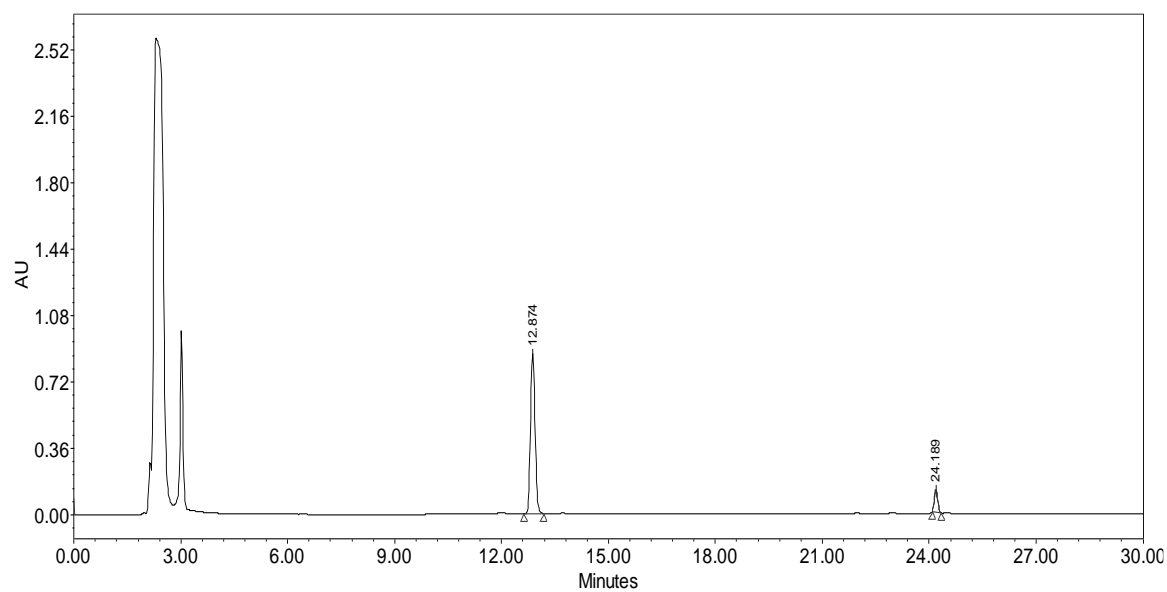


Figure A 6. Analytical purity HPLC spectrum of **FITC-Van**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 20-85% acetonitrile in water, 0.1% TFA. Product is 90% pure.

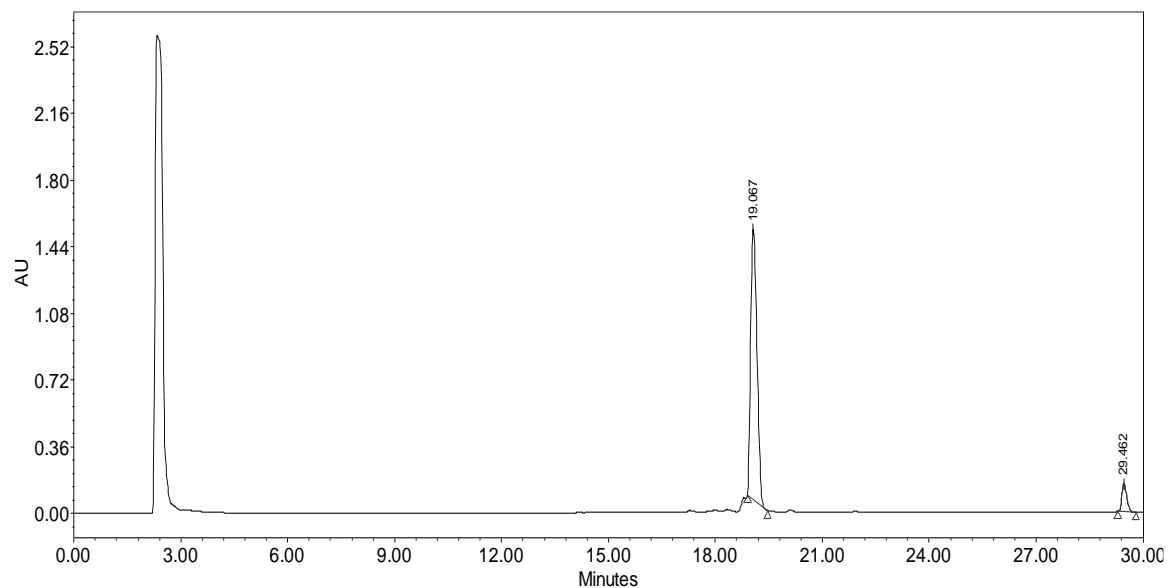


Figure A 7. Analytical purity HPLC spectrum of **RITC-Van**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 15-85% acetonitrile in water, 0.1% TFA. Product is 93% pure.

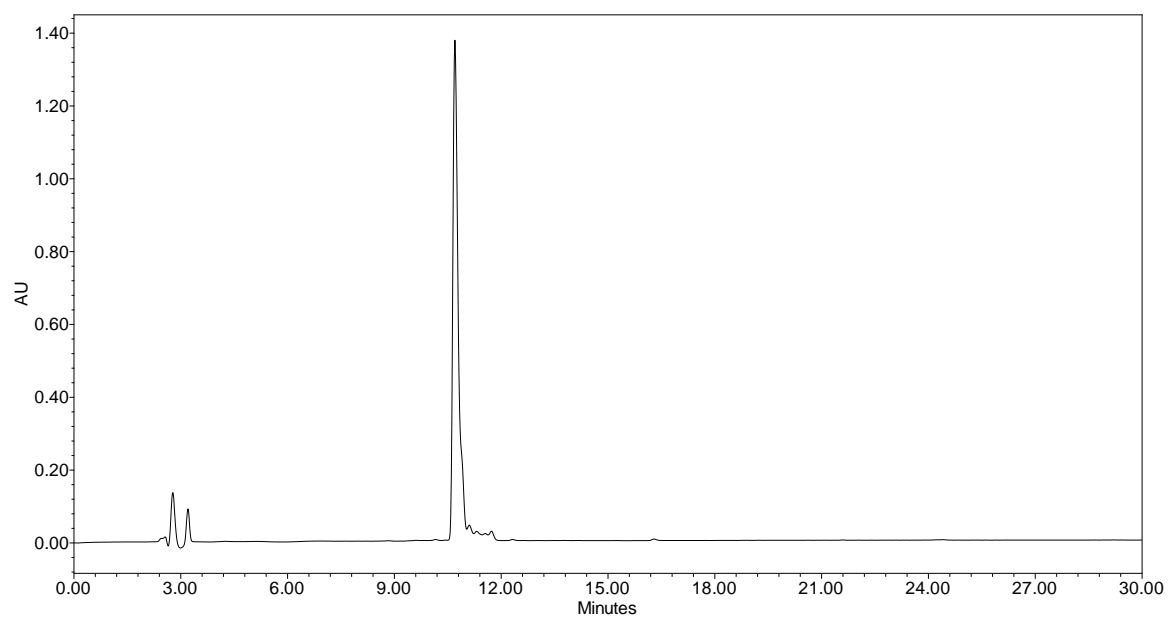


Figure A 8. Analytical purity HPLC spectrum of **Van-SH**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 10-70% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

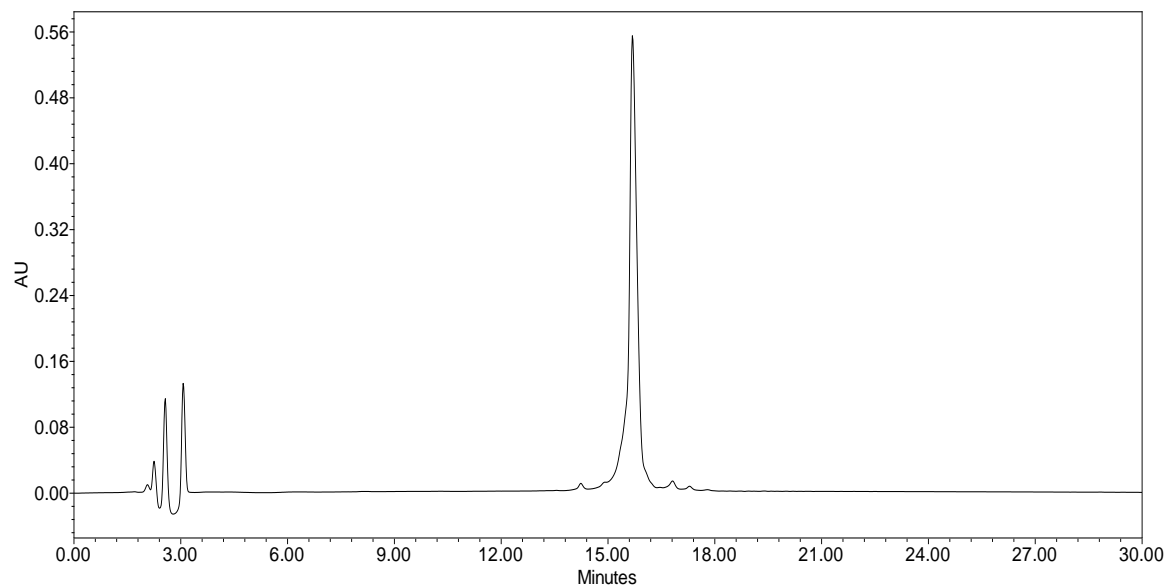


Figure A 9. Analytical purity HPLC spectrum of **P14GAP-SH**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 20-75% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

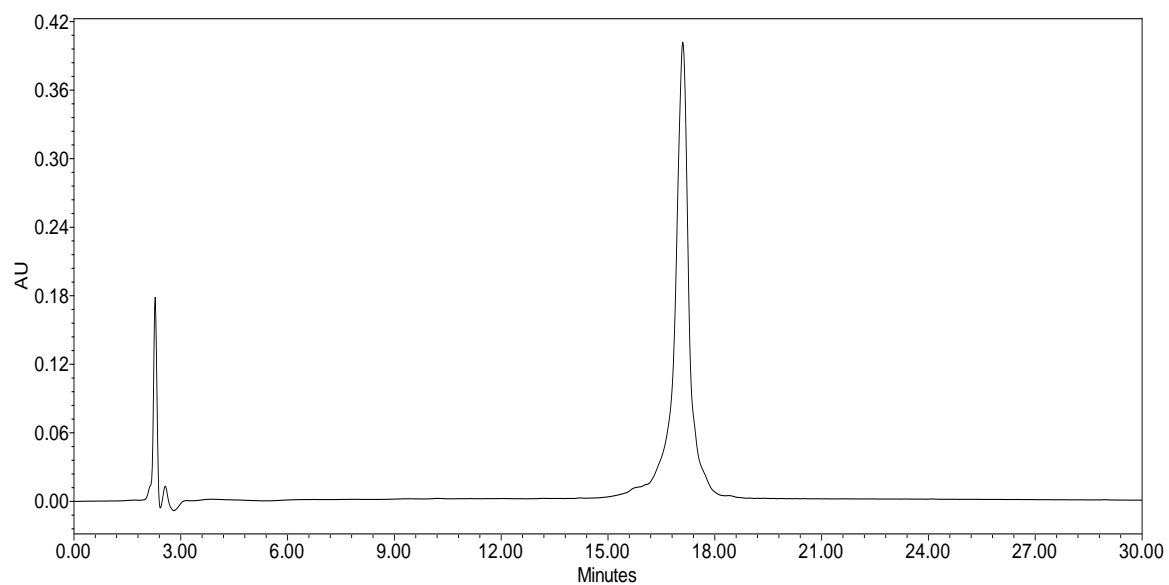


Figure A 10. Analytical purity HPLC spectrum of **FIP14GAP-SH**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 20-75% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

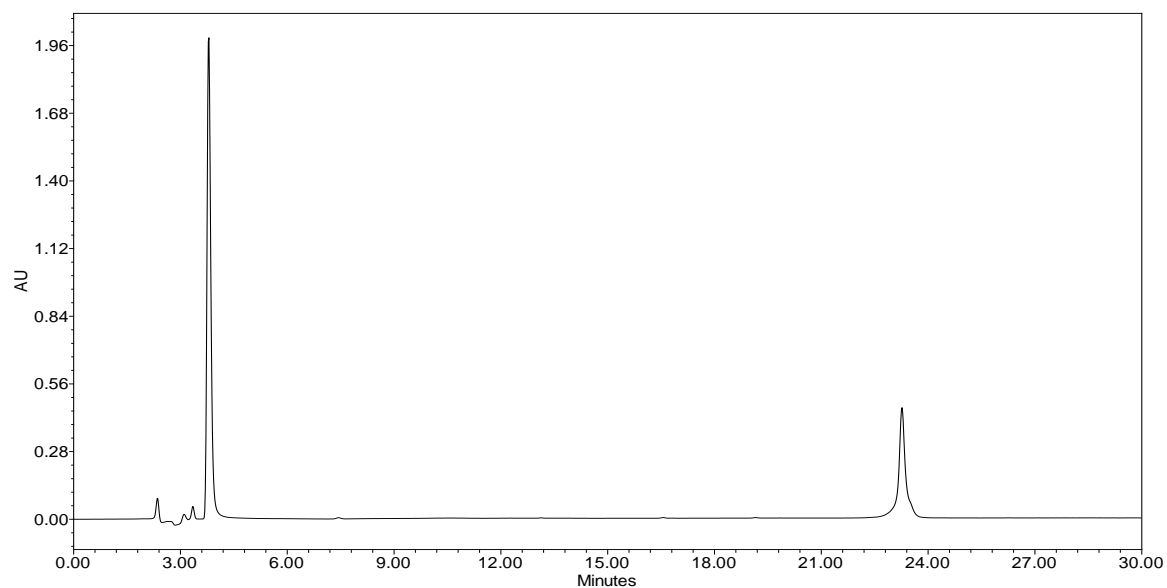


Figure A 11. Analytical purity HPLC spectrum of **VanP14GAPS**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 5-60% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

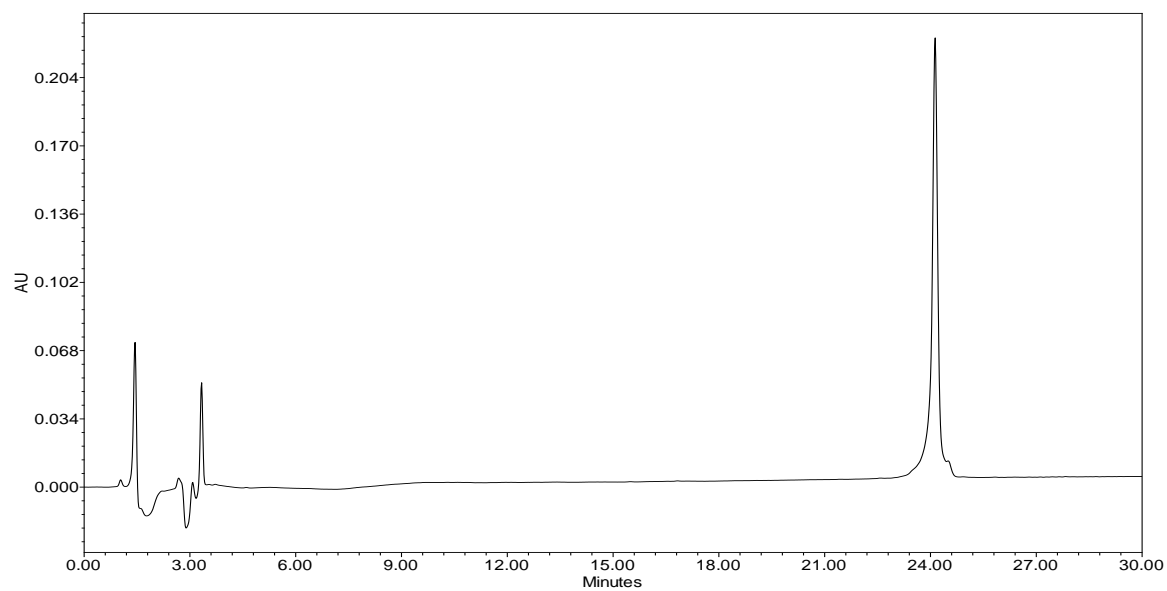


Figure A 12. Analytical purity HPLC spectrum of **FIVanP14GAPS**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 5-60% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

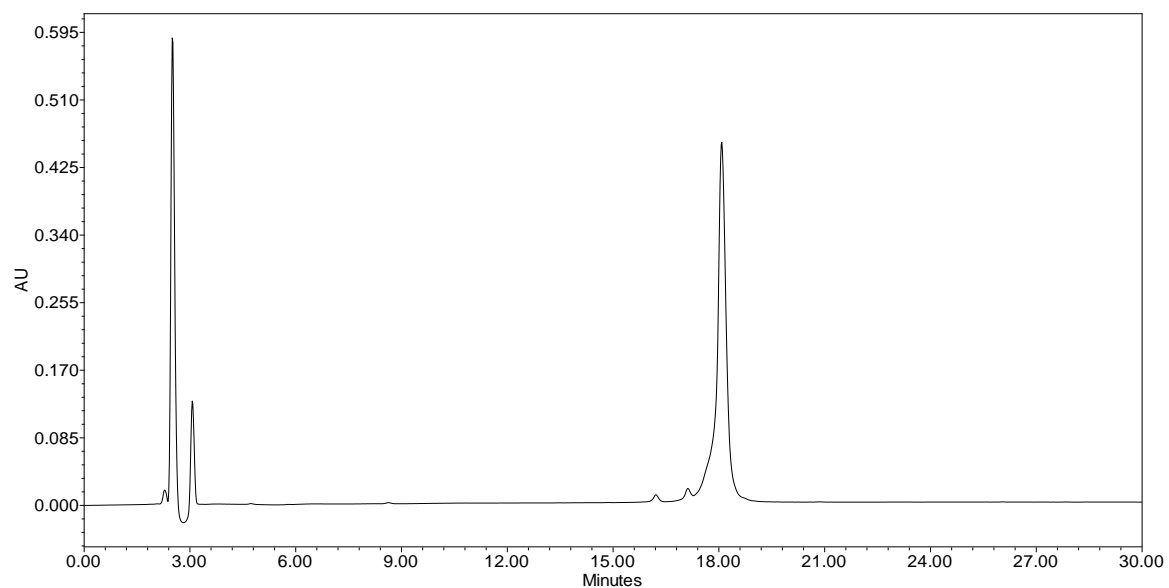


Figure A 13. Analytical purity HPLC spectrum of **LnzP14**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 20-80% acetonitrile in water, 0.1% TFA. Product is > 95% pure.

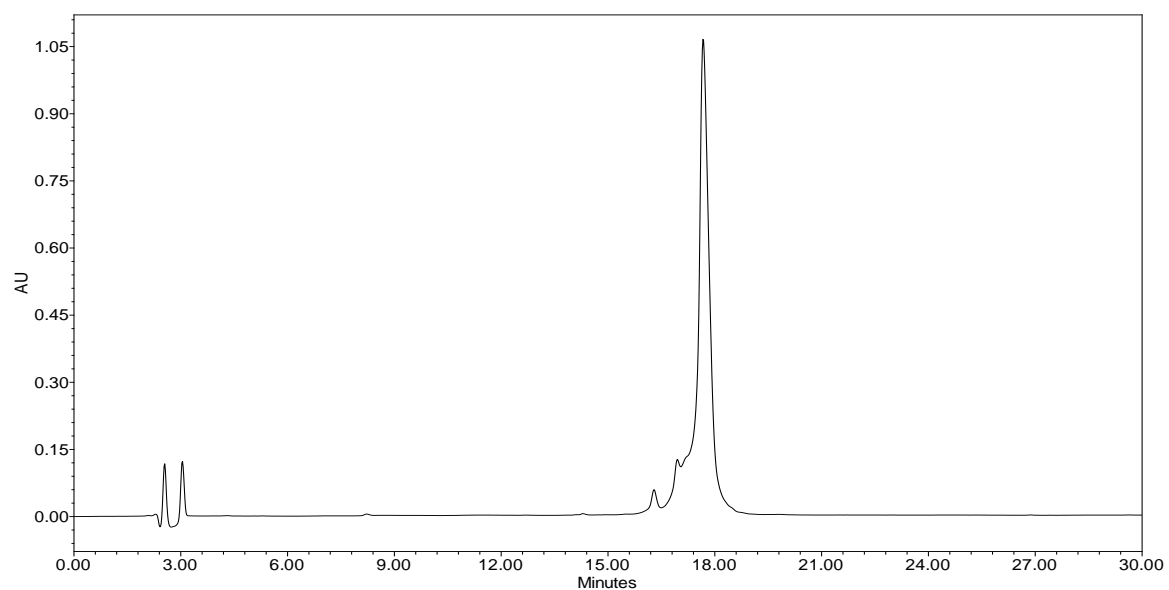


Figure A 14. Analytical purity HPLC spectrum of **FILnzP14**. C18 column, 214 nm, 1.2 mL/min. Gradient eluent, curve 6, 20-80% acetonitrile in water, 0.1% TFA. Product is 95% pure.

NMR Spectra

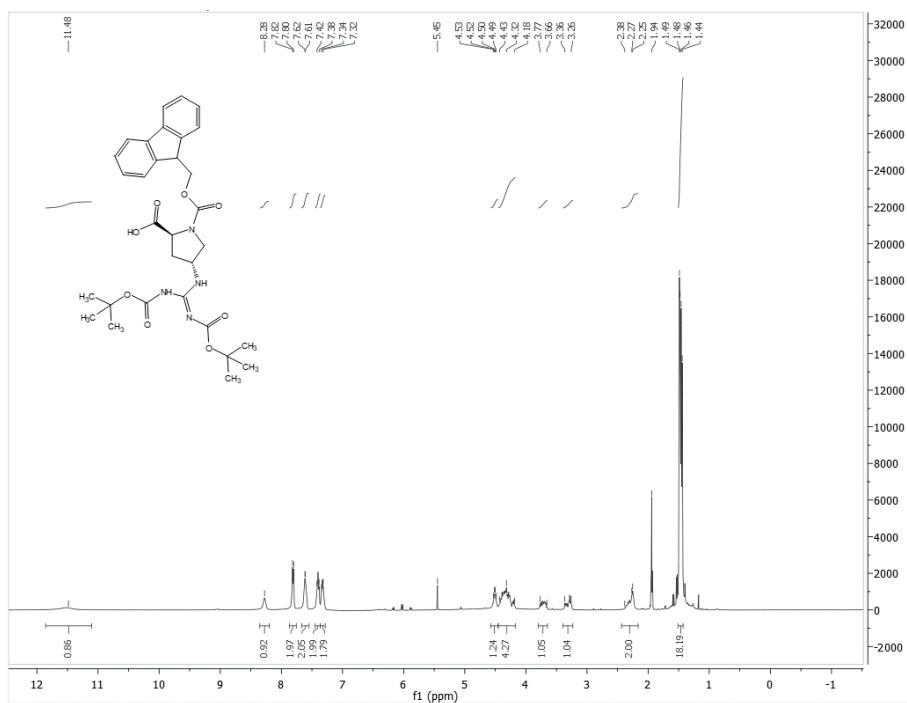


Figure A 15. ^1H NMR spectrum for **Fmoc-GAP**. Compound obtained from R. Blade.

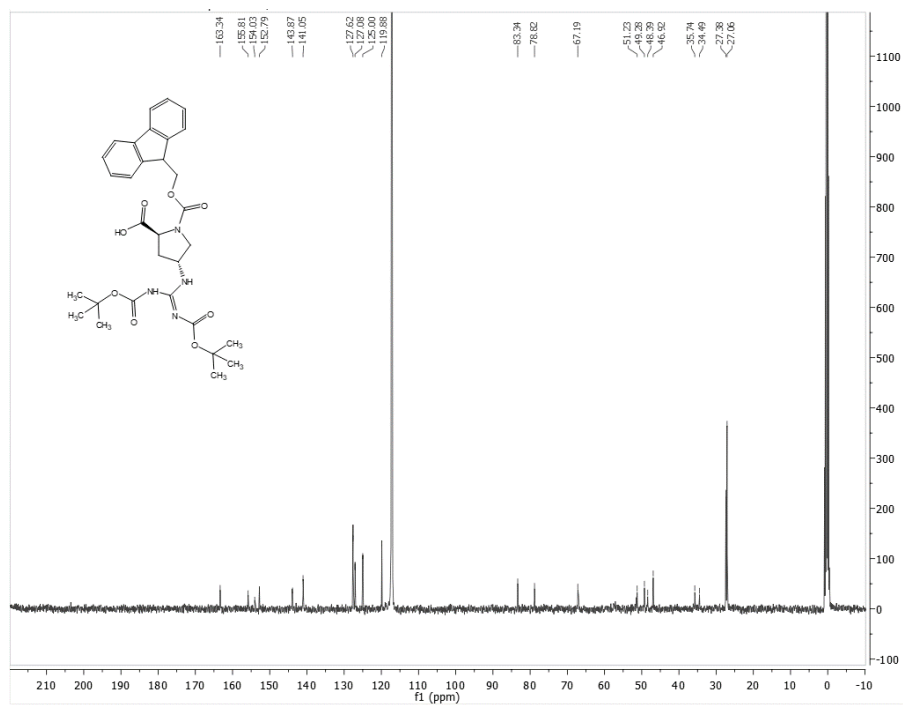


Figure A 16. ^{13}C NMR spectrum for **Fmoc-GAP**. Compound obtained from R. Blade.

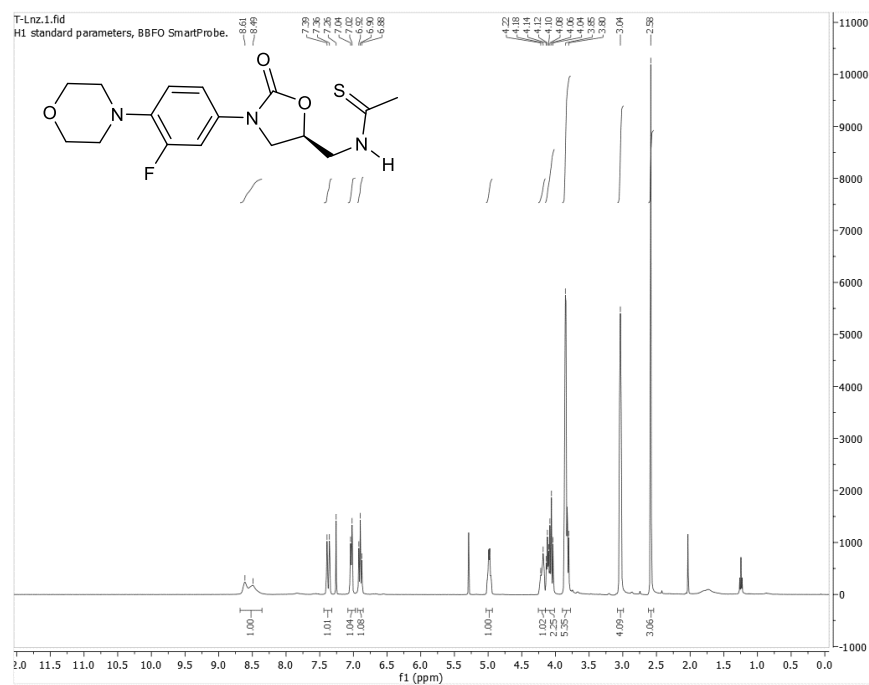


Figure A 17. ^1H NMR spectrum for **T-Lnz** as obtained by P. Pitasse Santos.

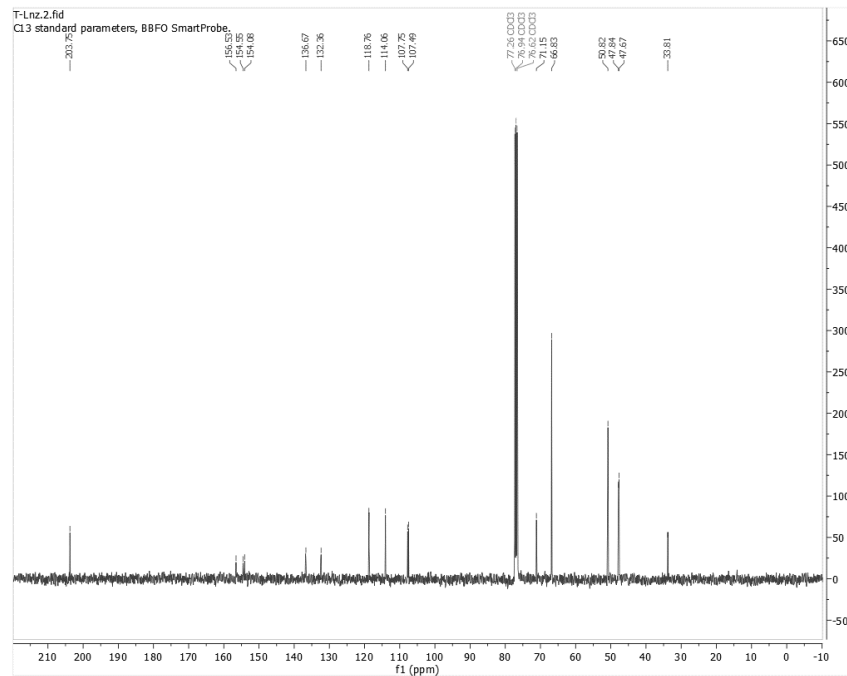


Figure A 18. ^{13}C NMR spectrum for **T-Lnz** as obtained by P. Pitasse Santos. Note: in comparison to linezolid, the most significant signal shifts were observed at the amide and its neighboring positions, indicating the thionylation was selective to this group.

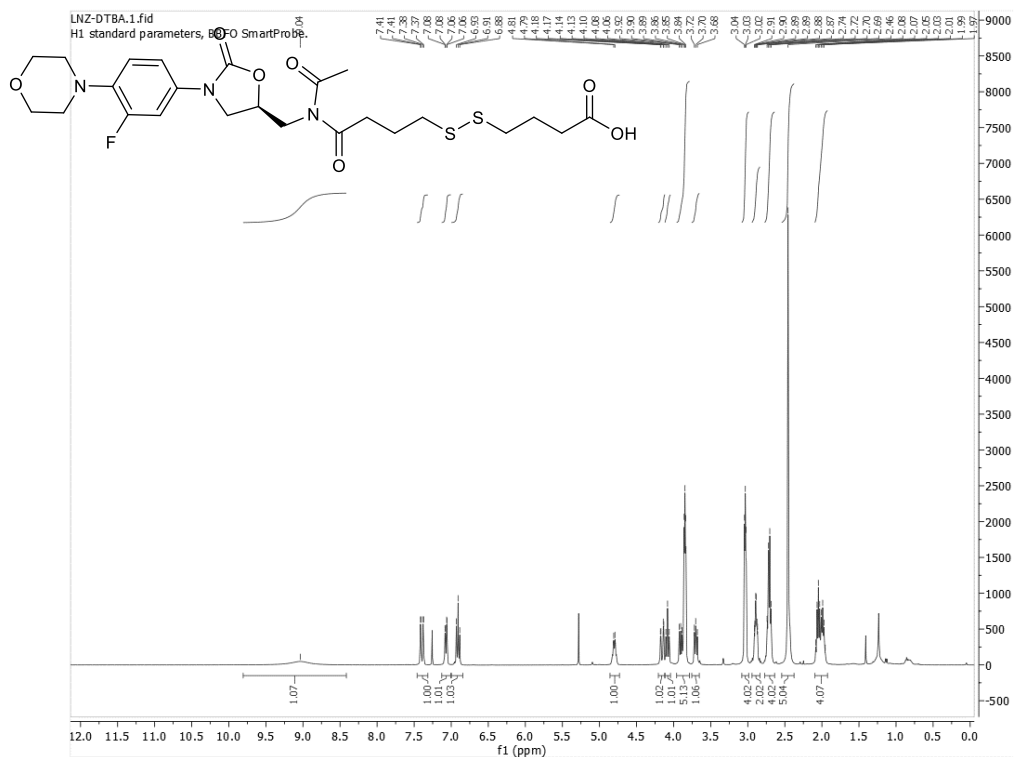


Figure A 19. ^1H NMR spectrum for **Lnz-DTBA** as obtained by P. Pitasse Santos.

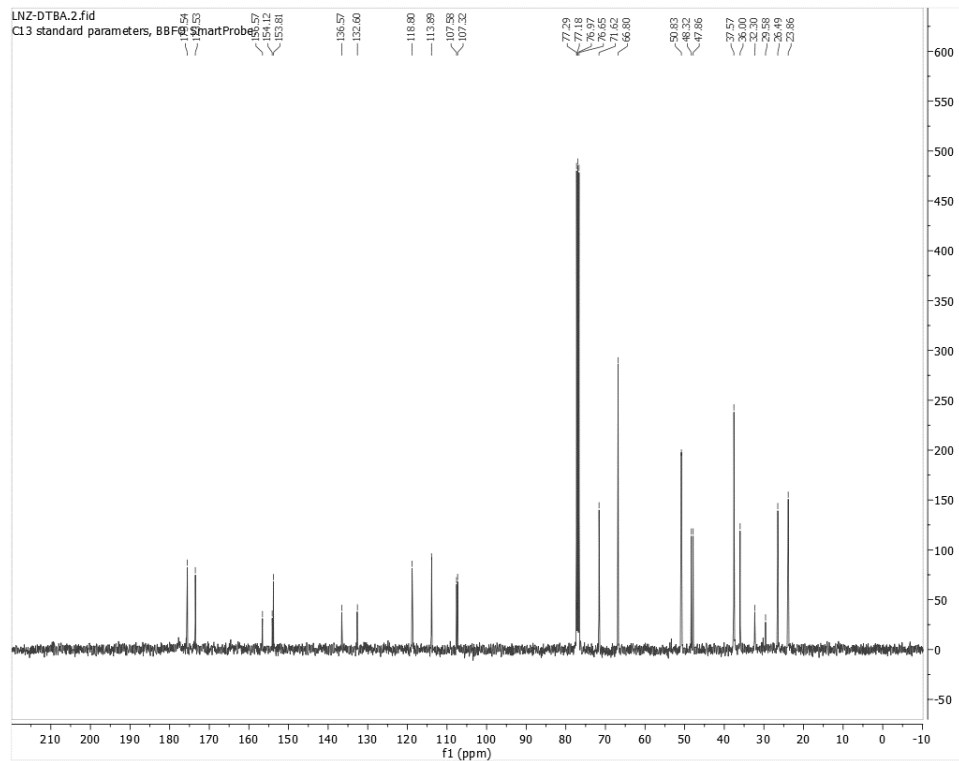


Figure A 20. ^{13}C NMR spectrum for **Lnz-DTBA** as obtained by P. Pitasse Santos.

MALDI-ToF Mass Spectra

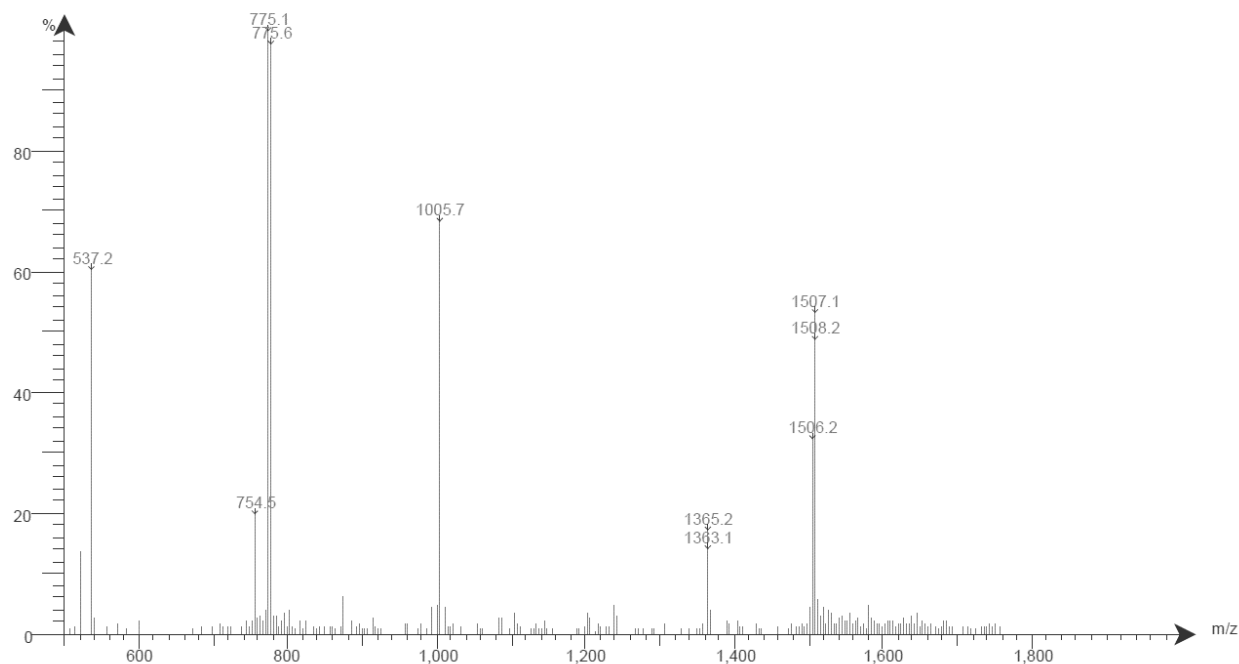


Figure A 21. Van-SH, Expected Mass: 1507, Observed Mass: 1507, 1508

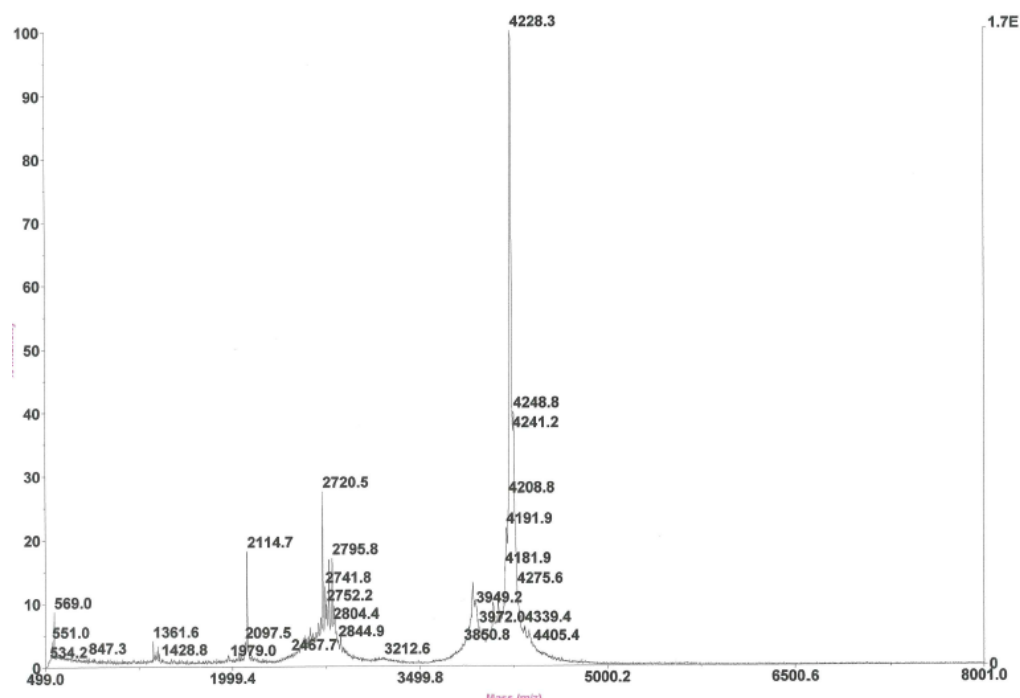


Figure A 22. VanP14S, Expected Mass: 4223, Observed Mass: 4228

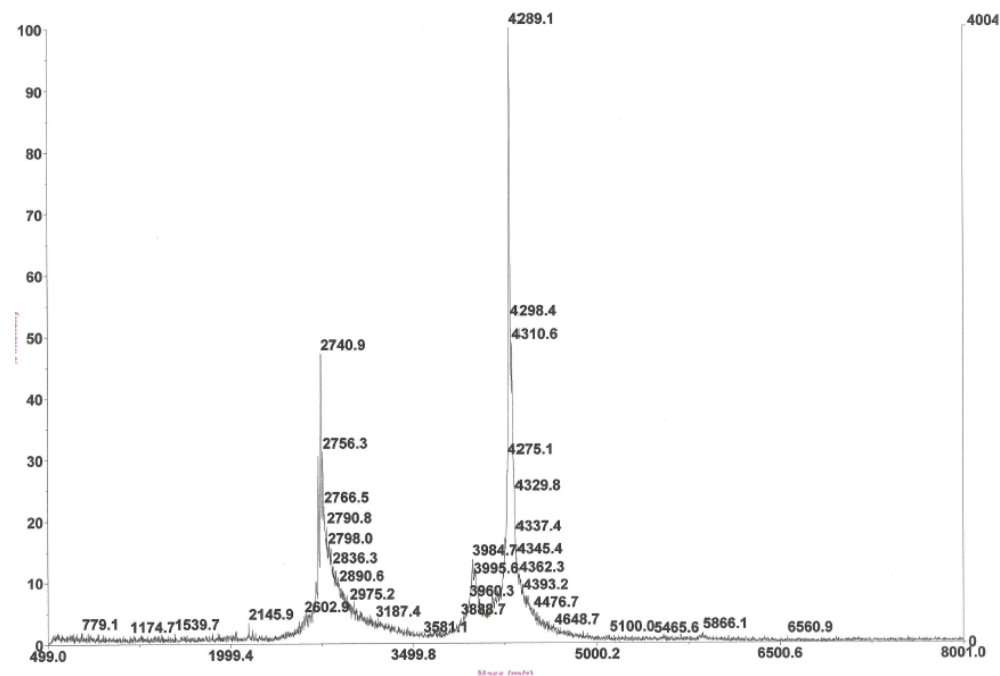


Figure A 23. VanMalP14, Expected Mass: 4286, Observed Mass: 4289

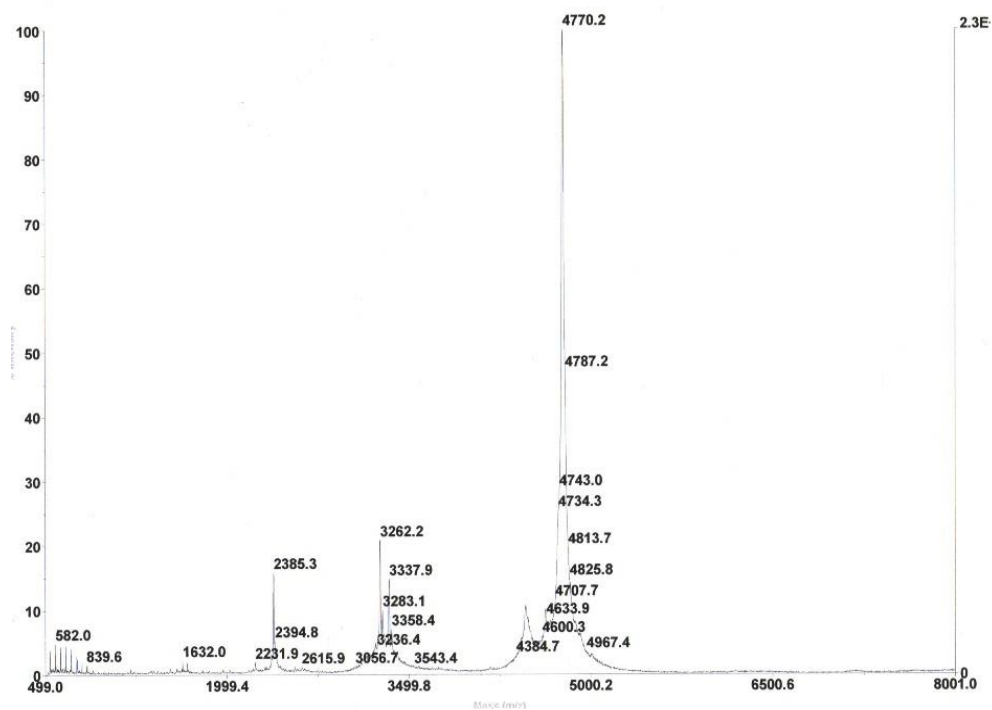


Figure A 24. FIVanP14S, Expected Mass: 4766, Observed Mass: 4770

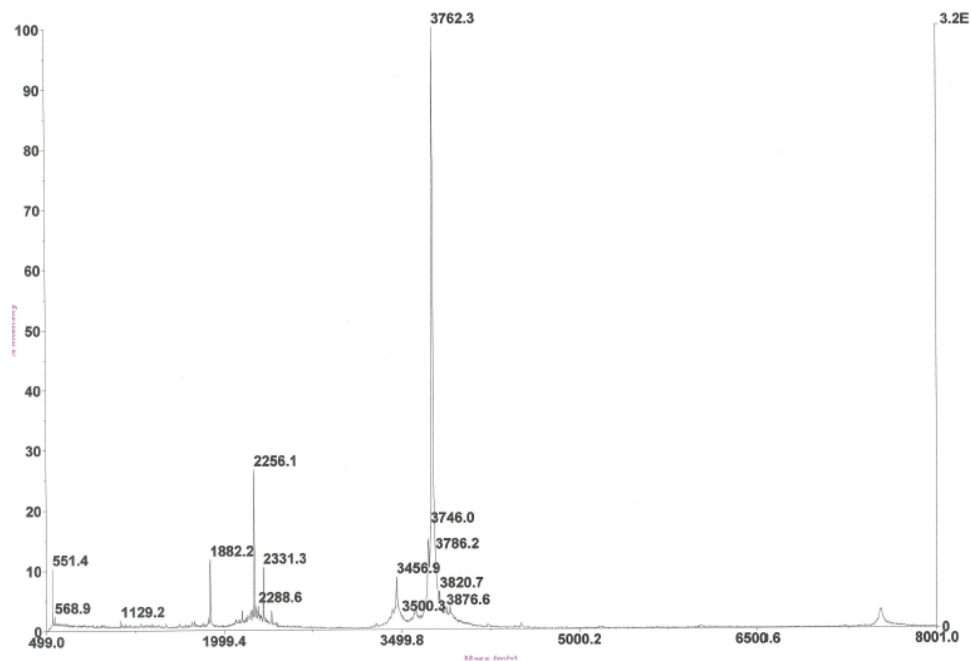


Figure A 25. VanP14GAPS, Expected Mass: 3758, Observed Mass: 3762

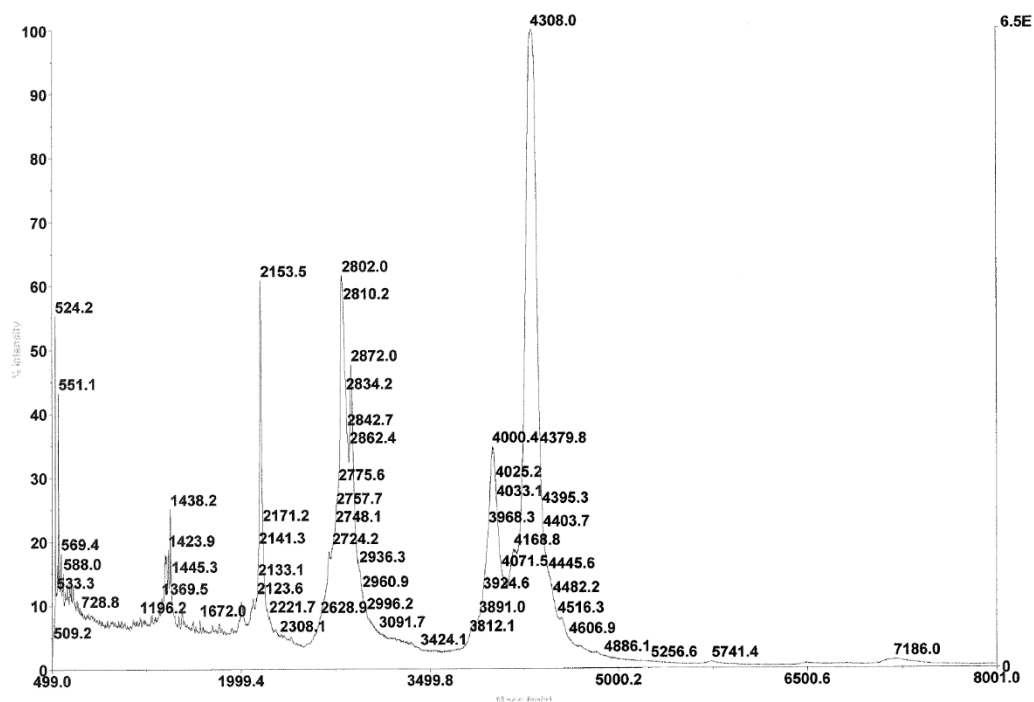


Figure A 26. FIVanP14GAPS, Expected mass: 4306, Observed mass: 4308

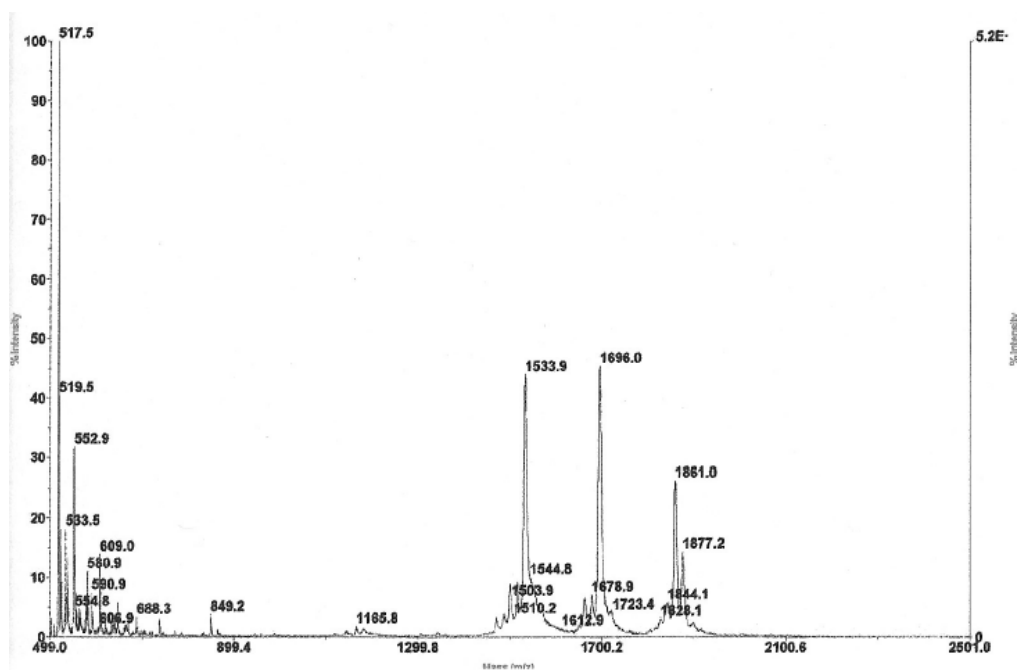


Figure A 27. FITC-Van; Expected masses: $[M+Na]^+=1861$, $[M+H]^+=1839$, $[M-Sugar+H]^+=1696$, $[M-2Sugar+H]^+=1533$; Observed masses: 1861, 1696, 1533

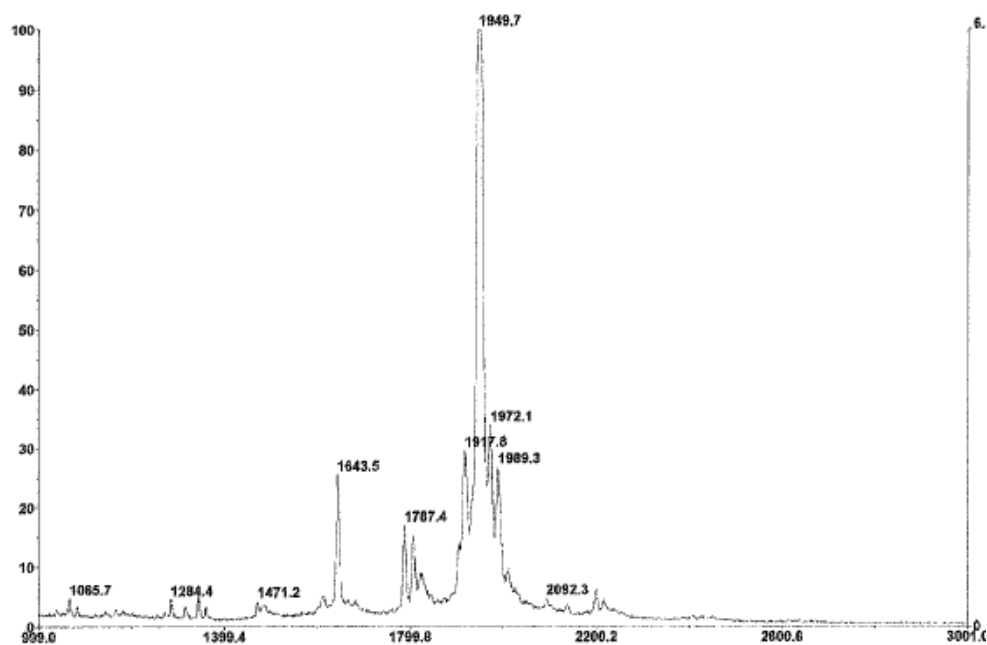


Figure A 28. RITC-Van; Expected masses: $[M+Na]^+=1972$, $[M+H]^+=1950$, $[M-Sugar+H]^+=1807$, $[M-Sugar-H_2O+H]^+=1789$, $[M-2Sugar+H]^+=1645$; Observed masses: 1949, 1787, 1643

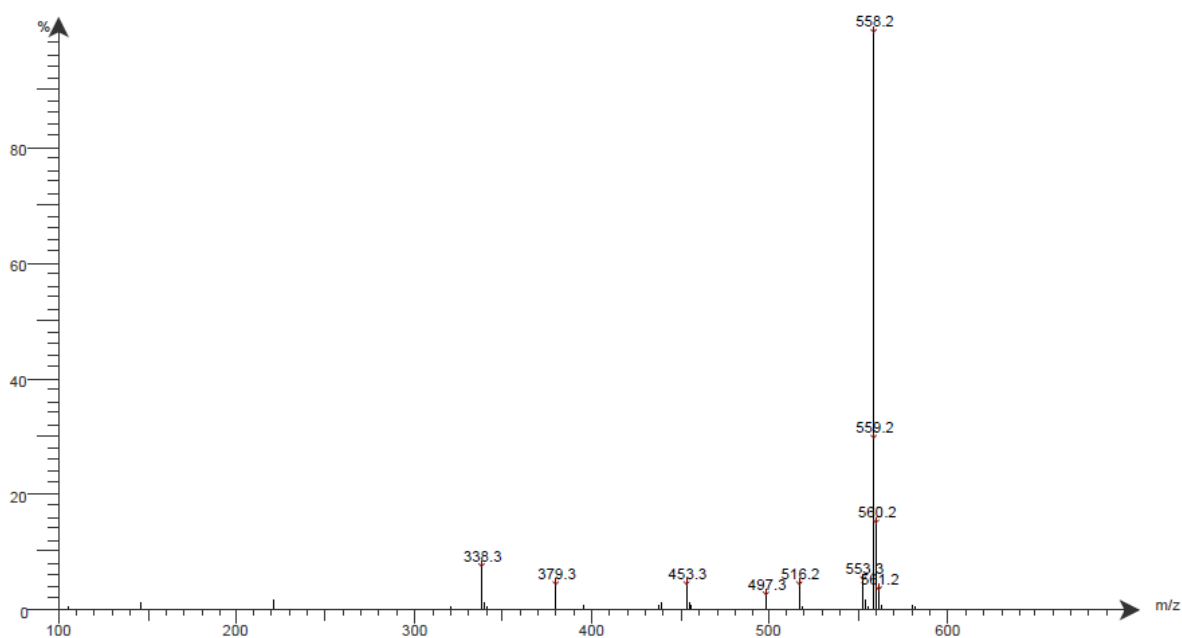


Figure A 29. Expected mass: 558, Observed mass: 558.2. Synthesized by P. Pitasse Santos.

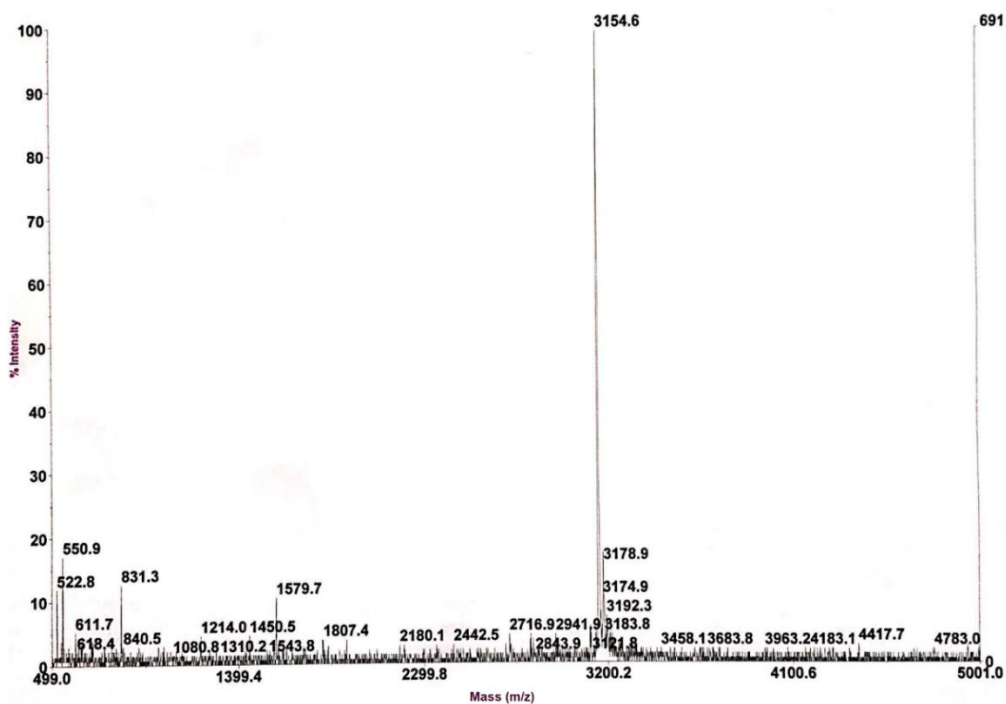


Figure A 30. LnzP14, Expected mass: 3156, Observed mass: 3155. Obtained by P. Pitasse Santos.

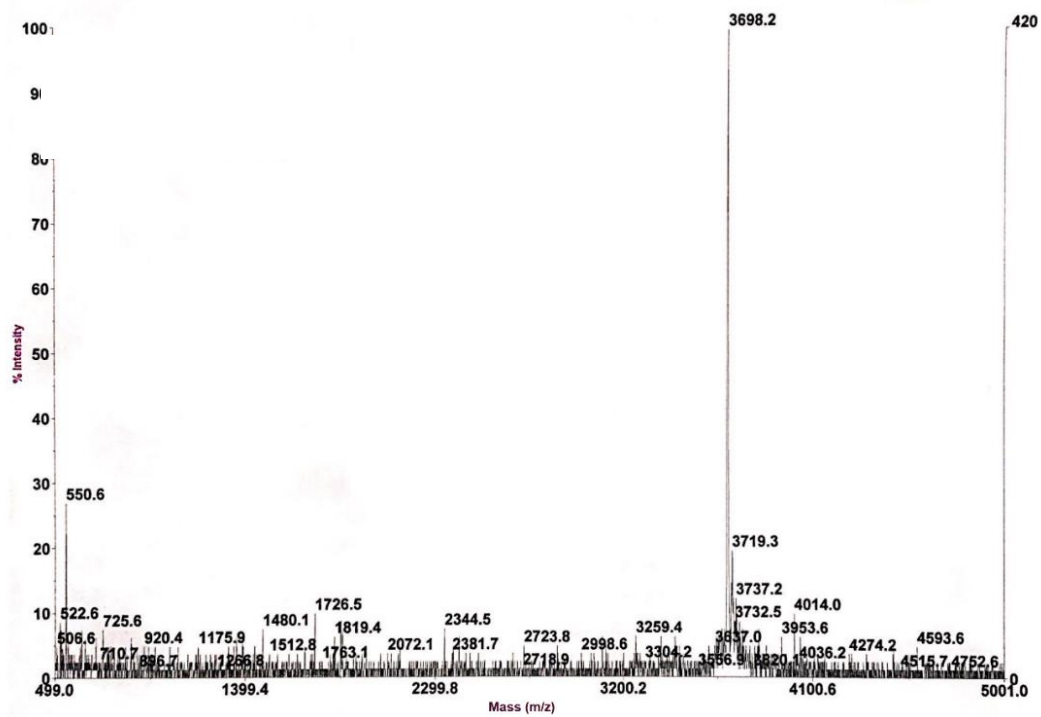


Figure A 31. FILnzP14, Expected mass: 3700, Observed mass: 3698. Obtained by P. Pitasse Santos.

VITA

Samantha Zeiders was born and raised in Harrisburg, PA. Upon graduation from Susquehanna Township High School in 2011, she attended Lafayette College, a small liberal arts school in Easton, Pennsylvania. In 2015, she obtained her B.S. degree in biochemistry with a minor in mathematics after working on undergraduate research projects spanning the subjects of analytical chemistry, environmental chemistry, biochemistry, and organic chemistry. Her research first began in the lab of Dr. Steven Mylon, whom she worked with for two years including two summer research opportunities on various analytical and environmental chemistry projects. Her passion for organic chemistry was sparked the following summer with Dr. William Miles, with whom she coauthored her first paper. Her last year of undergraduate studies culminated in the laboratory of Dr. Charles Nutaitis with her senior research project to graduate with an honors degree in chemistry.

Following the completion of her degree, Samantha pursued a postbaccalaureate fellowship for one year at the National Institutes of Health in Bethesda, Maryland. She worked in the tuberculosis laboratory of the National Institute of Allergy and Infectious Disease under the supervision of Dr. Clifton Barry.

Ultimately, Samantha joined the chemistry department at Purdue University in August 2016. She will complete her studies in the area of chemical biology and organic chemistry under the direction of Jean Chmielewski and will receive her PhD in chemistry in May 2021. Her post-graduation plans are presently kept open in search of the best opportunity and fit.

Targeting Intracellular Pathogenic Bacteria Through N-Terminal Modification of Cationic Amphiphilic Polyproline Helices

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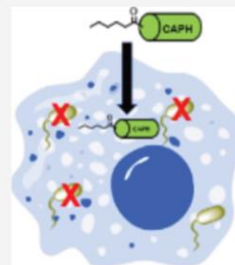
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ABSTRACT: Intracellular pathogens can thrive within mammalian cells and are inaccessible to many antimicrobial agents. Herein, we present a facile method of enhancing the cell penetrating and antibacterial properties of cationic amphiphilic polyproline helices (CAPHs) with modifications to the hydrophobic moiety at the N-terminus. These altered CAPHs display superior cell penetration within macrophage cells, and in some cases, minimal cytotoxicity. Furthermore, one CAPH, **Pentyl-P14** exhibited excellent antibacterial activity against multiple strains of pathogenic bacteria and promoted the clearance of intracellular *Shigella* within macrophages.



INTRODUCTION

The rise of antimicrobial resistance has created a global health crisis whereby many antibiotics have become ineffective. A number of antibiotic-resistant strains, including the multidrug-resistant *Acinetobacter baumannii* (*A. baumannii*), remain difficult to treat.^{1,2} This problem is further compounded by the issue that many bacteria such as *Listeria monocytogenes* (*L. monocytogenes*), *Mycobacterium tuberculosis* (*M. tuberculosis*), and *Shigella flexneri* (*S. flexneri*) can invade mammalian cells.^{3,4} Once inside the cell, these bacteria evade host immune responses and many therapeutics that do not accumulate sufficiently within cells.^{5,6} To combat this challenge, new antibiotics are needed that can effectively penetrate mammalian cells and eliminate these pathogenic bacteria.

Delivery systems targeting intracellular pathogens have been developed, such as nanoparticles containing antibiotics.^{7,8} An alternative approach has been to use cell-penetrating peptides in conjugation with antibiotics and peptide nucleic acids.^{9–11} Cell-penetrating peptides with inherent antibacterial properties have also been developed.^{12,13} These synthetic peptides were composed of a cationic amphiphilic polyproline helix (CAPH) scaffold and contained both hydrophobic and cationic groups to imbue the molecule with amphiphilicity (Figure 1A). One such CAPH, **P14LRR**, has displayed promising cell penetration, broad-spectrum antibacterial activity, and modest reduction of bacteria within macrophage cells.¹⁴

In efforts to facilitate greater cell penetration, peptides have previously been modified with various aliphatic fatty acid moieties at the N-terminus to generate lipopeptides. Although these lipopeptides demonstrated an improvement in cell

uptake, they suffered from extensive cytotoxicity.¹⁵ While long aliphatic fatty acids may be impractical for potential therapeutics, it is possible that other hydrophobic moieties at the N-terminus could improve the cell penetration of CAPHs, while remaining noncytotoxic. Herein, we present our efforts to develop potent CAPHs with N-terminal hydrophobic modifications that could easily be adapted to other peptides to improve cell penetration and antimicrobial activity, while remaining nontoxic to cells. Furthermore, we demonstrate that these modifications on CAPHs can be utilized to clear intracellular pathogens *in cyto*.

RESULTS AND DISCUSSION

Altering the CAPH hydrophobic face has resulted in enhanced targeting of intracellular bacteria.¹⁶ Therefore, we wished to evaluate the role of N-terminal hydrophobic moieties on cell penetration and antimicrobial activity (Figure 1B). To this end, five different functionalities at the N-terminus of CAPHs were designed: acetyl and pentyl aliphatic groups, and phenyl, naphthyl, and quinolyl aromatic groups (Figure 1B). These functionalities could easily be introduced on the N-terminus of resin-bound peptides. A 4-methyltrityl (Mtt)-protected lysine

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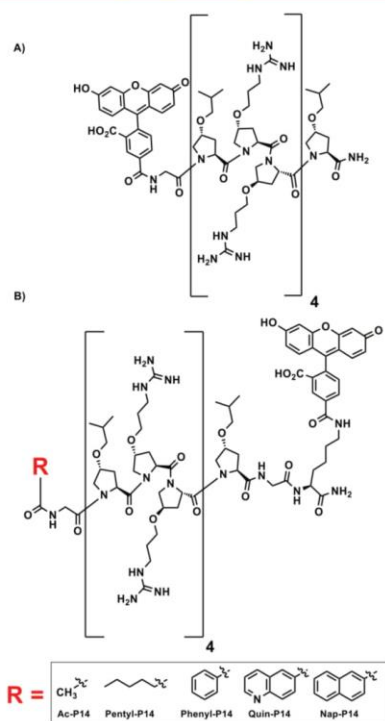


Figure 1. CAPHs. (A) Structure of P14LRR and (B) structure of CAPHs modified at the amino terminus with hydrophobic groups.

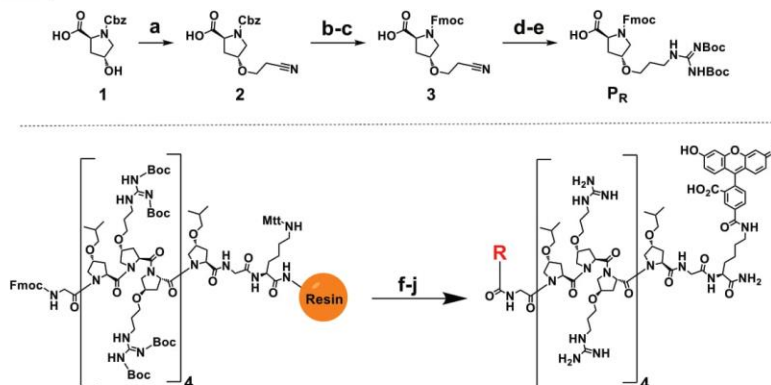
introduced at the carboxyl terminus of the peptide could be selectively deprotected and labeled with fluorescein for intracellular tracking. A glycine spacer between the (Mtt)-lysine residue was introduced to prevent unfavorable steric interactions with fluorescein.

The proline-modified unnatural amino acid containing the hydrophobic (P_L) functionality was synthesized as previously described.¹⁷ The guanidine-containing cationic P_R amino acid was synthesized following an improved three-step procedure (Scheme 1, top) as compared to the previously reported synthesis.¹⁷ Briefly, amino acid analogue 2 was prepared by treating Cbz-hydroxyproline 1 with sodium hydride, followed by acrylonitrile. The Cbz protecting group of 2 was removed through hydrogenation and replaced with an Fmoc protecting group to provide analogue 3. Finally, reduction of the nitrile group of 3, followed by addition of *N,N'*-bis-Boc-1-guanylpiprazole yielded P_R . This new synthetic strategy doubles the overall yield of the final product from 25 to 50% compared to the previously reported synthesis.¹⁷

The CAPH peptides were synthesized on a Rink amide resin using the standard Fmoc-based solid-phase peptide synthesis (SPPS) (Scheme 1, bottom). Following successful coupling at the N-terminus with the hydrophobic modifications, the (Mtt)-lysine was deprotected with hexafluoro-2-propanol (HFIP), and the resulting free amine was reacted with *N*-hydroxysuccinimide fluorescein. The peptides were cleaved from resin with concomitant side chain deprotection using a trifluoroacetic acid (TFA) cocktail. All peptides were purified to homogeneity by reverse-phase high-performance liquid chromatography (RP-HPLC) and characterized by matrix-assisted laser desorption/ionization (MALDI) mass spectrometry.

Circular dichroism (CD) was used to evaluate the conformation of the newly synthesized CAPHs as it has previously been reported that modifications to polyproline type II (PPII) helices can affect conformational stability.¹⁸ All CAPHs displayed a positive ellipticity at 225 nm that is

Scheme 1. Synthesis of P_R (Top) and On-Resin Coupling of Hydrophobic Functionalities on the N-Terminus of CAPH Peptides (Bottom)^a



^a(a) Acrylonitrile, NaH, THF, 16 h, 80%. (b) H_2 , Pd/C, MeOH. (c) Fmoc-OSu, $NaHCO_3$, acetone/ H_2O , quant. (d) H_2 , PtO_2 , AcOH, MeOH, (e) *N,N'*-bis-Boc-1-guanylpiprazole, TEA, DCM, 62%, (f) piperidine (25% in DMF), 25 min. (g) $R-CO_2H$, DIEA, HATU, 3 h. (h) HFIP (30% in DCM), 2 \times 30 min. (i) NHS-fluorescein, DIEA, DMF, 16 h. (j) 95% TFA, 2.5% TIPS, 2.5% H_2O , 2 h.

characteristic of the PPII¹⁹ (Figure S2), as was observed with the previous CAPHs.¹⁷ These data support that the N-terminal modifications did not affect the secondary structure of the polyproline helix within CAPHs.

As many therapeutics suffer from the inability to effectively rescue mammalian cells from intracellular pathogens, it is imperative that antibacterial agents transverse the mammalian cell membrane. Thus, the N-terminal CAPHs were assessed for their ability to accumulate within J774A.1 macrophage cells using flow cytometry. Cells were incubated with each CAPH derivative, and the cellular fluorescence was measured in the presence of trypan blue (TB) to quench extracellular fluorescence that may be caused by a membrane-bound peptide.²⁰ Overall, the installation of various hydrophobic groups at the N-terminus drastically improved the ability of CAPHs to penetrate macrophage cells in comparison to P14LRR and the control peptide Ac-P14 (Figure 2). Minor

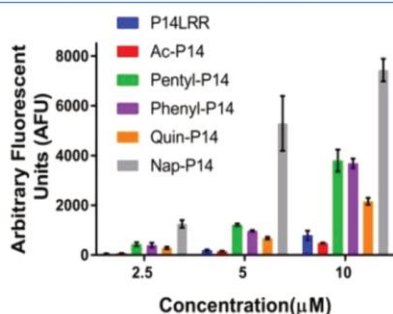


Figure 2. Cellular uptake of N-terminal CAPHs in J774A.1 macrophage cells. Cellular fluorescence was measured by flow cytometry after 1 h incubation with peptides. TB was used to quench fluorescence of membrane-bound peptides.

levels of membrane binding were observed for **Pentyl-P14**, **Phenyl-P14**, **Ac-P14**, and **Nap-P14**, as the decrease in cellular fluorescence with added TB was only 5–15% overall. In comparison, **Quin-P14** exhibited less-efficient cell penetration, as there was a decrease of up to 35% in cellular fluorescence when TB was added. At a concentration of 10 μM , **Phenyl-P14** and **Pentyl-P14** displayed about a fivefold increase in cell uptake compared to **P14LRR**, while **Quin-P14** displayed a

more modest threefold increase. **Nap-P14** yielded the highest improvement compared to **P14LRR**, showing about an 11-fold higher cell uptake. Interestingly, **Nap-P14** displays almost four times the cell accumulation as **Quin-P14**, despite only differing by one nitrogen atom. Moreover, the CAPH derivative with the medium aliphatic chain, **Pentyl-P14**, was approximately eight times more efficient at cell penetration than **Ac-P14**, demonstrating that the length of the hydrophobic tail is crucial for cell penetration. These trends were consistent even at concentrations as low as 2.5 μM . Together, these results show that the N-terminal hydrophobic moiety can greatly impact the ability of CAPHs to penetrate mammalian cells.

Intracellular pathogens can take refuge in different subcellular locations such as vacuoles (*Mycobacterium*) and the cytosol (*L. monocytogenes* and *S. flexneri*), making treatment difficult. We had previously shown that altering the hydrophobic moieties on the backbone of CAPHs resulted in enhanced targeting of intracellular pathogens.¹⁶ Similarly, we wished to investigate if the N-terminal modifications on CAPHs could affect their ability to localize to specific subcellular compartments. To this end, the subcellular localization of the N-terminal CAPHs was visualized via confocal microscopy. J774A.1 cells were incubated with CAPHs over a range of concentrations (2.5, 5, and 10 μM), and the cells were further treated with a Hoechst stain, Lysotracker, or Mitotracker to visualize the nucleus, endosomes, or mitochondria, respectively. At 5 μM , all CAPHs were found to localize primarily to the mitochondria with some cytosolic localization (Figure 3), whereas **Quin-P14** and **Nap-P14** also displayed nuclear localization in some cells (Figure S8). Interestingly, we observed concentration differences in the subcellular locations of these peptides. In all cases, with the exception of **Nap-P14**, there was a threshold concentration at which the peptides switched from endosomal to mitochondrial localization; at 2.5 μM **Pentyl-P14**, **Phenyl-P14**, and **Quin-P14** were observed in both compartments, and for **Ac-P14**, this concentration was 5 μM (Figure S8). Below this concentration, the peptides were endosomal, and above this concentration, the peptides were localized to the mitochondria. This concentration dependence on subcellular location has been observed previously with **P14LRR**.²¹ At all of the concentrations evaluated, **Nap-P14** was found colocalized with Lysotracker and Mitotracker (Figure S8). These data demonstrate that the hydrophobic moieties at the N-terminus can direct CAPHs to the mitochondria and endosomes, with some cytosolic localization, and by fine-tuning the hydro-

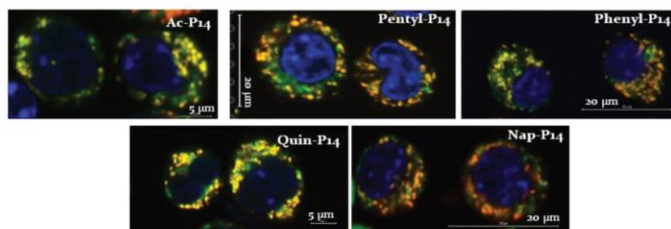


Figure 3. Subcellular localization of N-terminal CAPHs (green) in J774A.1 macrophage cells. Cells were treated with peptides (5 μM) (green) for 1 h. Following the incubation, cells were stained with Mitotracker (red) and Hoechst 3342 (blue) to visualize the mitochondria and nuclei, respectively. Cells were visualized by confocal microscopy, and the images show the overlay of the green, red, and blue channels for the treated cells.

phobic group (bis-aromatic), CAPHs are found to localize in the nucleus as well.

To rescue macrophage cells from intracellular pathogens, mammalian cells must maintain their viability in the presence of antibiotic therapies. Thus, to ensure that the N-terminal modifications did not reduce cell viability below a reasonable level, CAPHs were screened against J774A.1 macrophage cells using the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay at 10 μ M for 9 h (Figure S3).²² As mentioned earlier, long fatty acid N-terminal modifications have been shown to greatly improve cell penetration, but were toxic.¹⁵ In contrast, four out of five of the N-terminally modified CAPHs displayed acceptable cell viability. **Ac-P14** and **P14LRR** displayed minimal toxicity in macrophage cells, whereas **Phenyl-P14**, **Pentyl-P14**, and **Quin-P14** showed acceptable cell viability (75–90%). **Nap-P14** displayed the lowest cell viability (60%), despite its improvement on cell penetration. From the cell uptake and cell viability data, it was determined that **Pentyl-P14** and **Phenyl-P14** satisfied two main requirements for application against intracellular pathogens: (1) efficient cell penetration and (2) limited toxicity in J774A.1 macrophage cells.

After determining that the N-terminal CAPHs boosted cell penetration, we investigated the antibacterial activity against a selection of Gram-positive and Gram-negative pathogens. An *in vitro* broth dilution assay was used to screen the potency of the CAPH derivatives against the intracellular bacteria *M. tuberculosis*, *L. monocytogenes*, and *S. flexneri*, *Escherichia coli*, and the multidrug resistant *A. baumannii*.^{11,23–25} **Quin-P14** was excluded because of the less-effective cell penetration, and **Nap-P14** was deemed to be too cytotoxic to continue forward. **Pentyl-P14** showed superior activity as compared to **Ac-P14** (Table 1), with two-fold increased potency against *M.*

Table 1. In Vitro Antibacterial Activity of CAPHs against Pathogenic Bacteria-MIC (μ M)

	<i>L. monocytogenes</i>	<i>S. flexneri</i>	<i>M. tuberculosis</i>	<i>A. baumannii</i>	<i>E. coli</i>
P14LRR	8	8	16	16	4
Ac-P14	>16	16	16	16	8
Pentyl-P14	8	4	8	8	2
Phenyl-P14	16	8	16	8	4

tuberculosis, *A. baumannii*, and *L. monocytogenes*, and four-fold enhanced potency against *E. coli* and *S. flexneri*. **Phenyl-P14** showed slightly lower activity against the before-mentioned pathogens as compared to **Pentyl-P14**. Although fluorophores were included within these peptides, we have previously shown that they play a minimal role in the antibiotic potency of CAPHs.¹⁵

A β -galactosidase release assay in *E. coli* was used to determine if the peptides acted through a cell lysis mechanism (Figure 4A). At 2 \times the minimum inhibitory concentration (MIC) values, **P14LRR** and **Pentyl-P14** showed minimal bacterial cell lysis, whereas the positive control mellitin displayed significant levels of lysis. **Ac-P14** and **Phenyl-P14** showed somewhat higher levels of lysis. At 4 \times the MIC values, the N-terminally modified CAPHs demonstrated lysis (Figure S4). This suggests that at higher peptide concentrations, membrane lysis could play a role in antibacterial activity.²⁶ However, there was no observed hemolysis with human red blood cells (hRBCs) up to 40 μ M (Figure 4B), signifying that

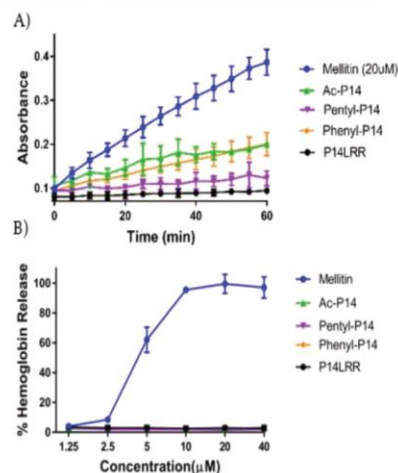


Figure 4. Investigating the mode of antimicrobial action of N-terminally modified CAPHs. (A) Leakage of β -galactosidase from *E. coli* after 1 h with CAPHs at 2 \times the MIC values. (B) Hemolysis of hRBCs when treated with CAPHs for 1 h.

CAPHs may selectively lyse the bacterial membrane at higher concentrations, while leaving hRBCs intact.

With the confirmation that the N-terminally modified CAPHs possess both antibacterial activity and mammalian cell penetration, we probed clearance of the intracellular pathogen *S. flexneri* within J774A.1 macrophage cells.²⁷ **Pentyl-P14** was selected for this assay, as it exhibited the best combination of cell uptake, cell viability, and antimicrobial activity. **Ac-P14** was also evaluated, although it exhibited much lower cellular uptake and antibacterial activity. **Pentyl-P14** was found to clear \sim 75% of *S. flexneri* within macrophage cells after 12 h, whereas **Ac-P14** exhibited a modest reduction (\sim 40%) of intracellular *S. flexneri* (Figure 5). These clearance results are likely due to a combination of several factors, including efficiency of cell penetration, antimicrobial activity, and

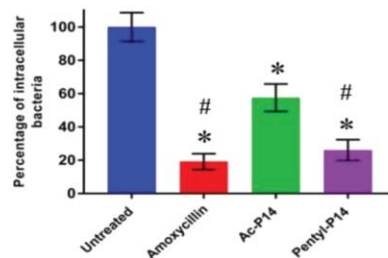


Figure 5. Intracellular bacterial clearance activity of **Pentyl-P14** and **Ac-P14** (10 μ M) against *S. flexneri* in J774A.1 murine macrophage cells after 12 h. (*) denotes the statistical difference between the tested groups and the untreated control. (#) indicates a statistical significance with respect to **Ac-P14**. *#P values of <0.05 are considered significant.

subcellular location. **Pentyl-P14** and **Ac-P14** were shown to localize mainly to the mitochondria with some cytosolic localization at 10 μ M. Previous studies of CAPHs have shown a direct transport mechanism into the cytosol, followed by migration to the mitochondria.²⁸ While in the cytosol, **Pentyl-P14** is likely to interact with and eradicate *S. flexneri*, a cytoplasm-residing pathogen. Furthermore, **Pentyl-P14** possesses superior cell penetration and antibacterial activity against *S. flexneri* when compared to **Ac-P14**. Therefore, it stands to reason that **Pentyl-P14** would be more effective in clearing intracellular *S. flexneri*.

To further investigate if the pentyl moiety was an optimal N-terminal modification, four additional aliphatic modifications of CAPHs were synthesized containing C4–C7 tails: **Butyl-P14**, **Branched-P14**, **Hexyl-P14**, and **Heptyl-P14** (Figure 6).

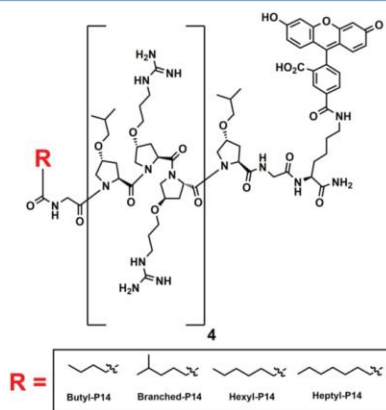


Figure 6. Structure of CAPHs modified at the amino terminus with additional aliphatic groups.

These CAPHs were assessed for their ability to penetrate mammalian cells via flow cytometry (Figure S6). At a concentration of 10 μ M, **Butyl-P14** was about 10% less efficient at cell penetration as compared to **Pentyl-P14**. Conversely, **Branched-P14**, **Hexyl-P14**, and **Heptyl-P14** displayed increased cell accumulation at 10 μ M as compared to **Pentyl-P14** (about four, two, and ninefold, respectively).

Unfortunately, the CAPHs that displayed superior cellular uptake as compared to **Pentyl-P14** also suffered from drastically reduced cell viability. **Branched-P14**, **Hexyl-P14**, and **Heptyl-P14** showed significant levels of toxicity, with cell viability ranging from 4 to 43% (Figure S7). The observation that longer N-terminal aliphatic chains result in severe cell toxicity is consistent with previous lipopeptide reports.¹⁵ These data support the premise that the N-terminal pentyl modification represents the best compromise between cell accumulation and viability, as shorter aliphatic chains (**Ac-P14** and **Butyl-P14**) were not as efficacious in cell penetration. Conversely, longer aliphatic modifications (**Branched-P14**, **Hexyl-P14**, and **Heptyl-P14**) possessed greater cell penetration but were much more cytotoxic. Therefore, **Pentyl-P14** was deemed to be the most effective peptide, as it had excellent mammalian cell penetration while remaining noncytotoxic. We have previously observed an increase in cell toxicity with longer

C6-aliphatic groups within the side chain functionalities of CAPHs.¹⁶ Interestingly, this is also observed here with a single modification at the amino terminus.

CONCLUSIONS

In conclusion, combating intracellular bacteria remains a difficult challenge in the context of developing new therapeutics.^{1–5} Functionalizing CAPHs with hydrophobic moieties at the N-terminus improved cell uptake and, in some cases, antibacterial activity. These N-terminal modifications are installed in a facile manner on a resin, making them easily adaptable to other peptides to improve their cell penetration or activity against intracellular pathogenic bacteria. By altering CAPHs to bear the five-carbon aliphatic chain, we found that **Pentyl-P14** was effective at clearing the intracellular pathogen *S. flexneri*. Therefore, **Pentyl-P14** is an excellent candidate to further the overarching goal of designing new therapeutics to target difficult-to-treat bacteria that reside within mammalian cells.

EXPERIMENTAL SECTION

Materials and Methods. Unless otherwise stated, common chemicals and solvents including Fmoc-protected amino acids, resins, carboxylic acids, and coupling reagents for SPPS were purchased from commercial sources and used without further purification. Compound 2 was purchased from Chem-Impex International. H-Rink Chem Matrix Rink amide resin (0.46 mmol/g) was purchased from Pcas Biomatrix Inc. Sterile media [Dulbecco's modified Eagle medium (DMEM) ι -glutamine, fetal bovine serum (FBS), and penicillin–streptomycin], fluorescent dyes (NHS-fluorescein, Mitotracker, LysoTracker, and Hoechst), and buffers [phosphate buffered saline (PBS)] used in cell culture were purchased from commercial sources. Melittin and amoxicillin were purchased from Sigma-Aldrich.

Synthesis of Compound 2. The synthesis of 2 proceeded as previously reported,¹⁷ with the following minor modifications. To a solution of NaH (1.58 g, 65.9 mmol) in tetrahydrofuran (THF, 150 mL) with 4 Å mol sieves at 0 °C and under N₂ atmosphere was added an ice-cooled solution of 1 (5.0 g, 18.8 mmol) in THF (50 mL). The mixture was stirred at 0 °C for 1 h. To this mixture was added acrylonitrile (5.0 mL, 93.4 mmol), and the reaction was allowed to warm to room temperature and stirred for 24 h. The reaction mixture was cooled to 0 °C, and water (100 mL) was added to quench the excess NaH. The THF was removed *in vacuo*, and HCl (10%) was added to bring the solution to a pH of 1. The resulting solution was extracted with EtOAc (3 \times), the organic layers were dried over anhydrous MgSO₄, and the solvent was removed *in vacuo*. The desired product was purified by silica gel column chromatography [95% dichloromethane (DCM), 4% MeOH, 1% AcOH] to provide 2 as a colorless oil in 80% yield. Characterization of 2 was previously reported.¹⁷

Synthesis of Compound 3. To a solution of 2 (600 mg, 1.9 mmol) in 20 mL of ethanol was added 10% Pd/C (60 mg). The solution was stirred under 1 atm of hydrogen for 3 h. The solution was gravity filtered through filter paper, the solvent was removed *in vacuo*, and the resulting material was used in the next step without further purification. The material from the previous step was dissolved in deionized water (10 mL), the solution was cooled to 0 °C, and sodium bicarbonate (475 mg, 5.7 mmol) was added. To this cooled mixture, a solution of Fmoc-OSu (699 mg, 2.1 mmol) in 10 mL of acetone was added dropwise, and the resulting slurry was allowed to warm to room temperature and stirred overnight. The reaction mixture was treated with 10% HCl to a pH of 1 and extracted with EtOAc (3 \times). The organic layers were dried over anhydrous Na₂SO₄, and the solvent was removed *in vacuo*. The desired product was purified by silica gel column chromatography (96% DCM, 3% MeOH, 1% AcOH) to provide 3 as a white solid in quantitative yield. ¹H NMR (400 MHz, CDCl₃): δ 10.14 (br s, 1H), 7.73 (dd, J = 24 Hz, 8

H₂, 2H), 7.55 (m, 2H), 7.34 (m, 4H), 4.40–4.51 (m, 3H), 4.24–4.36 (m, 1H), 4.13 (m, 1H), 3.51–3.75 (m, 4H), 2.55 (q, *J* = 8 Hz, 2H), 2.33–2.48 (m, 1H), 2.11–2.26 (m, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 177.2*, 175.6, 155.7, 154.5*, 143.6, 141.2, 129.0, 127.7, 127.0, 124.9, 119.9, 117.4, 67.9, 63.7, 57.9, 57.2*, 51.4, 47.0, 36.6*, 34.7, 19.0 (*indicates minor rotamer). HRMS (APCI): calcd for C₂₃H₂₂N₂O₃ [M + H]⁺, 407.1601 *m/z*; found, 407.1598.

Synthesis of P_R.¹⁷ To a solution of 3 (700 mg, 1.7 mmol) in 20 mL of MeOH was added 1 mL of AcOH, followed by PtO₂ (70 mg). The solution was stirred under 1 atm of hydrogen overnight. The solution was filtered through celite, the solvent was removed *in vacuo*, and toluene was used to remove excess AcOH. The resulting material was used in the next step without further purification. The material from the previous step was dissolved in 15 mL DCM and cooled to 0 °C. To this cooled mixture, a solution of triethylamine (0.720 mL, 5.2 mmol) in 5 mL DCM was added dropwise, followed by *N,N'*-bis-Boc-1-guanlypyrazole (695 mg, 2.2 mmol). The resulting slurry was allowed to warm to room temperature and stirred overnight. The reaction was extracted with DCM (3×), the organic layers were dried over anhydrous Na₂SO₄, and the solvent was removed *in vacuo*. The desired product was purified by silica gel column chromatography (96% DCM, 3% MeOH, 1% AcOH) to provide P_R as a white solid in 62% yield. Characterization of P_R was previously reported.¹⁷

Peptide Synthesis. H-Rink Chem Matrix Rink amide resin (200 mg, 0.46 mmol/g, 100–200 mesh) was added to a 10 mL peptide synthesis flask and washed with DMF, CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each). The resin was swelled by adding DMF (7 mL) to the flask, and the flask was agitating for 30 min. After 30 min, the DMF was drained, and the resin was washed with DMF, CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each). Following this wash, Fmoc-protected amino acids [(Fmoc-P_R (2 equiv), Fmoc-P_L (2 equiv), Fmoc-Lys-Mtt (4 equiv), or Fmoc-gly (4 equiv))] were dissolved in 7 mL of DMF, and HATU (2 or 4 equiv) and DIEA (4 or 8 equiv) were added. This amino acid solution was added to the peptide flask and agitated for 3 h at room temperature. After 3 h, the solution was drained, and the resin was washed with DMF, CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each). To deprotect the amino acid, a piperidine (7 mL, 25% in DMF) solution was added to the flask and agitated for 25 min. The piperidine solution was drained and washed with DMF, CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each). This procedure was repeated for each sequential amino acid in the peptide sequence. The Fmoc-protecting group of the final amino acid was deprotected with piperidine (7 mL, 25% in DMF). The N-terminus of CAPHs was functionalized by adding a DMF (7 mL) solution of carboxylic acid [featuring one of the hydrophobic groups (4 equiv)], HATU (4 equiv), and DIEA (8 equiv) to the peptide flask and agitating for 3 h. Successful amino acid couplings/deprotections were monitored via Kaiser (1° amino acid) or Chloranil (2° amino acid) tests.^{29,30}

Deprotection of MTT Side Chain, Coupling of Fluorescein. Following coupling of the hydrophobic group to the N-terminus, the above resin was washed with CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each). For each N-terminus hydrophobic modification, an aliquot of resin (~50 mg) was transferred to a separate 10 mL synthesis flask. The resin was treated with HFIP (7 mL, 30% in DCM) and agitated for 30 min. This procedure was repeated twice, and the resin was washed with CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each) after each HFIP wash. Successful Mtt deprotection was monitored by the Kaiser test. The resin was washed with DMF, CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each), and N-hydroxysuccinimide fluorescein (1.2 equiv) and DIEA (2.4 equiv) in DMF (7 mL) was allowed to react with the resin for 16 h in the dark. The resin was then washed with CH₂Cl₂, MeOH, and CH₂Cl₂ (7 mL, 2× each).

Cleavage and Purification of Peptides. A fresh TFA cocktail solution of 10 mL (95% TFA, 2.5% TIPS, 2.5% H₂O) was prepared and was added to the resin and agitated for 2 h. After 2 h, the solution was drained into a 50 mL centrifuge tube. The resin was subsequently washed with the TFA cocktail (10 mL, 2×) and CH₂Cl₂ (10 mL, 2×), and all the filtrate was collected. TFA was removed *in vacuo*, and the peptide was precipitated with ice-cold diethyl ether (15 mL) overnight at –79 °C. The filtrate was centrifuged (3500 RPM, 15

min) and dried. The peptide was purified to homogeneity (>95% pure, Figure S1) using RP-HPLC using a gradient of 25–80% of CH₃CN (0.1% TFA) in water (0.1% TFA). The gradient was run for 60 min using a Luna C18 semi-prep column, using a flow rate of 10.0 mL/min and monitored with a UV detector at 214 and 254 nm. The mass of the peptides was confirmed via MALDI-time of flight (MALDI-TOF) mass spectrometry, and the concentration of aqueous stock solutions of peptides were confirmed using UV–visible spectroscopy at a wavelength of 518 nm.

Circular Dichroism. Data were recorded on a Jasco CD spectropolarimeter (Model J-1500) at 25 °C using a 1 mm path length quartz cell. The spectra were averaged over three scans taken from 260 to 190 nm, with a data pitch and bandwidth of 1 nm, at a scan rate of 20 nm/min. All peptides were analyzed at 50 μM concentration in 20 mM phosphate buffer (pH 7.2). The spectra were background-subtracted and processed from degrees of rotation to mean residue ellipticity by dividing by the appropriate path length, peptide concentration, and number of residues in the peptide.

Cell Uptake. J774A.1 cells were cultured in DMEM supplemented with 10% FBS, and 1% penicillin/streptomycin at 37 °C under 5% CO₂ atmosphere. J774A.1 cells (125,000) were harvested and transferred to round-bottom tubes (BD Biosciences) and allowed to adhere overnight. The next day, these cells were treated with CAPHs (2.5–10 μM) in 10% FBS-supplemented DMEM (300 μL) and were incubated for 1 h at 37 °C. Cells with no CAPHs treatment (DMEM supplemented with 10% FBS only) served as a control for the experiment. Upon completion of the 1 h incubation period, the cells were centrifuged (1100 rpm, 7 min @ 4 °C), and the spent media was aspirated. The cells were resuspended in TB (400 μL, 1 mg/mL in PBS), and the fluorescence of the cells was measured using a FACSCalibur Flow Cytometer (BD Biosciences) equipped with a 488 nm argon laser. Emissions for fluorescein-labeled peptides were collected in the FL 1 channel. Data were obtained in duplicate from three independent experiments and processed using the BD software.

Cell Toxicity. The viability of J774A.1 cells with CAPHs peptides was determined via a colorimetric MTT assay. In Brief, J774A.1 cells (20,000) were cultured at 37 °C under 5% CO₂ atmosphere and were seeded into a sterile 96-well plate and allowed to adhere for 24 h. The spent media was aspirated, and the J774A.1 cells were treated with CAPHs (10 μM in DMEM supplemented with 10% FBS) for 9 h at 37 °C. After this incubation period, the treatment was aspirated, and the cells were washed with PBS. Next, 100 μL of fresh DMEM media was added to each well, followed by 10 μL of 12 mM MTT reagent. The plate was incubated for an additional 2 h at 37 °C under 5% CO₂ atmosphere. Finally, the MTT solution was aspirated, and 100 μL of DMSO was added to each well. The 96-well plate was allowed to shake for 5 min at room temperature, and the absorbance of each well was measured at 590 nm using a microplate reader. Results were expressed as the percentage of viable cells as compared to a control that contained no CAPHs treatment. Data were obtained in duplicate from three independent experiments.

Subcellular Localization. J774A.1 cells (200,000) were seeded in a 4-well Lab-Tek chambered slide and were allowed to adhere for 18 h at 37 °C under 5% CO₂ atmosphere. The media was then aspirated, and the cells were washed with PBS (400 μL). Next, CAPHs were added to each well in 400 μL of DMEM supplemented with 10% FBS. The cells were incubated for 1 h at 37 °C under 5% CO₂ atmosphere. After the incubation period, the treatment was aspirated and washed with PBS (400 μL). The cells were further treated with Hoechst 33342 (1000 nM) and either Mitotracker (100 nM) or LysoTracker (300 nM) for 30 min at 37 °C under 5% CO₂ atmosphere. The excess dye was aspirated, the cells were washed with 400 μL of PBS, and fresh serum-supplemented DMEM was added to each well. Imaging was performed using a Nikon AIR multiphoton inverted confocal microscope under a 60× oil objective. Fluorescein, Hoechst 33342, and Mitotracker/LysoTracker were excited using 488, 350, and 561 nm lasers, respectively.

β-Galactosidase Assay. *E. coli* (ATCC 25922) was grown to the mid-exponential phase (OD₅₉₀ = ~0.6) in Mueller–Hinton Broth (MHB) at 37 °C with shaking. β-Galactosidase expression was

induced for 1 h using a freshly prepared solution of isopropyl- β -D-thiogalactopyranoside in PBS (1 mM final concentration). Following the induction, an aliquot (10 mL) of the bacterial suspension was centrifuged, washed twice with fresh MHB, and plated into a sterile 96-well plate. Next, 10 μ L aliquots of CAPHs were added to give final concentrations corresponding to 1 \times , 2 \times , or 4 \times the MIC value (against *E. coli*). Bacteria were also treated with sterile water and melittin (20 μ M final concentration), which served as controls. The 96-well plate was incubated for 1 h at 37 $^{\circ}$ C. At the end of the incubation period, the plate was centrifuged at 4000 rpm for 10 min. A volume of 80 μ L of the supernatant from each well as added to a new sterile 96-well plate. Next, 20 μ L of freshly prepared 2-nitrophenyl- β -D-galactopyranoside in PBS was added to each well (0.8 mg/mL final concentration). The β -galactosidase activity was monitored at OD_{405nm} every 5 min over the course of 1 h using a microplate reader. Data were obtained in duplicate from at least two independent experiments.

Hemolysis. hRBCs were centrifuged at 1200 rpm at 4 $^{\circ}$ C for 5 min. The hRBCs were washed with PBS (pH 7.4) two times, and a 4% suspension (v/v) of hRBCs was prepared in PBS. A volume of 50 μ L of the hRBCs solution was transferred to a sterile 96-well plate, and 50 μ L of CAPHs in PBS was added to each well to give a final suspension of 2% (v/v) of hRBCs. The plate was then incubated for 1 h at 37 $^{\circ}$ C. At the end of the incubation, the plate was centrifuged at 1200 rpm for 5 min at 4 $^{\circ}$ C. From each well, 80 μ L of the supernatant was then transferred to a new sterile 96-well plate. Finally, the release of hemoglobin caused by hemolysis was quantified by measuring the absorbance of the wells at 405 nm with a microplate reader. For controls, wells were treated with PBS, melittin, or 0.1% (final concentration) Triton X-100 in PBS. The percentage of hemolysis was calculated on the basis of the 100% hemolysis release with 0.1% Triton X-100. Data were obtained in duplicate from at least two independent experiments.

Antimicrobial Susceptibility Testing. The antimicrobial activity of the CAPHs and control antibiotics against the tested isolates, except for *M. tuberculosis*, was performed following the Clinical and Laboratory Standards Institute guidelines.³¹ Briefly, bacterial cells were cultured overnight on trypticase soya agar plates (Becton Dickinson). Colonies were picked and suspended in NaCl 0.9% to a density of 0.5 McFarland. The bacterial suspensions were further diluted 1:60 in CA-MHB (cation adjusted Mueller–Hinton Broth Becton Dickinson), and 100 μ L portions of the bacterial suspension were added to 96-well plates containing the CAPHs and control antibiotics at a concentration range of 0.125–16 μ M. The 96-well plates were incubated for 16–20 h at 37 $^{\circ}$ C before recording the MIC values. The MICs reported represent the lowest concentration of each peptide or standard antibiotic necessary to inhibit the bacterial growth. For testing the antimicrobial activity of the CAPHs and control antibiotic against *M. tuberculosis*, a resazurin microtiter assay was performed following a previously reported protocol.³² Briefly, mycobacterial colonies grown on the Lowenstein–Jensen (Becton Dickinson) medium were suspended in the 7H9-S broth, adjusted spectrophotometrically to a no. 1 McFarland tube standard, and further diluted 1:10 in 7H9-S broth. Then, 100 μ L of the 7H9-S broth was dispensed in each well of a sterile flat-bottom 96-well plate, and serial twofold dilutions of each of the peptides and control antibiotics were prepared directly in the plate. One-hundred microliters of the prepared inoculum was added to each well. Plates were covered, sealed in a plastic bag, and incubated at 37 $^{\circ}$ C under a normal atmosphere. After 7 days of incubation, 30 μ L of resazurin solution, 0.02% (w/v) in distilled water, was added to each well, and the plate was re-incubated overnight. A change in color from blue to pink indicated the growth of bacteria, and the MIC was defined as the lowest concentration of the compound that prevented this change in color.

Activity against Intracellular Bacteria. Following previously described protocols, the ability of the peptides to reduce the burden of intracellular *S. flexneri* was evaluated.^{16,33} Murine macrophage cells (J774A.1) were cultured in DMEM supplemented with 10% FBS at 37 $^{\circ}$ C under a CO₂ (5%) atmosphere. J774.1 cells were infected with

S. flexneri 1a ATCC 9199 cells at a multiplicity of infection of approximately 1:100. After 1 h of infection, J774A.1 cells were washed with DMEM and further incubated with gentamicin (100 μ g/mL) for 1 h to kill extracellular bacteria. The peptides at the respective concentrations were added to each well (four replicates per test agent). After 12 h incubation at 37 $^{\circ}$ C with 5% CO₂, the compounds were removed, J774A.1 cells were washed and lysed using 0.1% Triton-X. The cell lysates were serially diluted in PBS and transferred to trypticase soy agar plates to determine viable bacterial cfu, inside the J774A.1 cells. The plates were incubated at 37 $^{\circ}$ C for 20 h before counting viable cfu/mL. Data are presented as the percentage of intracellular cfu/mL in treated murine macrophage cells relative to the untreated control.

■ ASSOCIATED CONTENT

■ Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.joc.0c00871>.

Mass spectrometry data, RP-HPLC retention times and traces, and CD spectra of N-terminal CAPHs; cell viability test data; β -galactosidase assay data; Cellular uptake of aliphatic N-terminal CAPHs in J774A.1 macrophage cells; and cell localization data (PDF)

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The manuscript was written through contributions of all authors.

Notes

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